

Synthesis and Characterization of MnCo₂O₄ Cuboidal Microcrystals as a High Performance Pseudocapacitor Electrode

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ABSTRACT

Manganese cobaltite (MnCo₂O₄) is currently under screening as a high performance supercapacitor electrode owing to its high theoretical capacitance, improved electrical conductivity and long term cyclic stability. Herein, we report synthesis of MnCo₂O₄ cuboidal microcrystals using hydrothermal method and compare its performance with its flakes prepared by solid combustion process. Crystal structure, surface properties, and electrochemical properties of the flakes are studied using X-ray diffraction, gas adsorption, field emission scanning electron microscopy, cyclic voltammetry, galvanostatic charge–discharge cycling, and electrochemical impedance spectroscopy. The electrochemical properties of MnCo₂O₄ flakes synthesized using hydrothermal synthesis are superior to that synthesized using the solid combustion process. Electrochemical properties of the cuboidal microcrystals (~specific capacitance, $CS \sim 600 \text{ F g}^{-1}$ @ 0.5 A g^{-1}) are superior to those synthesized by the combustion process ($CS \sim 128 \text{ F g}^{-1}$) due to improved faradic utilization of active surface area, layered cuboidal morphology, faster OH⁻ ion penetration owing to higher diffusion coefficient, and larger voltage range available for electrochemical reaction.

KEYWORDS: MnCo₂O₄; Pseudocapacitors; Electrochemical energy storage; Metal cobaltites

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