PERFORMANCE INVESTIGATION OF WATER-SOLUBLE ADDITIVES FOR DRAG REDUCTION AGENT IN PIPELINES

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DOCTOR OF PHILOSOPHY (CHEMICAL ENGINEERING) UNIVERSITI MALAYSIA PAHANG

PERFORMANCE INVESTIGATION OF WATER-SOLUBLE ADDITIVES FOR DRAG REDUCTION AGENT IN PIPELINES

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Thesis submitted in fulfilment of the requirements for the award of the degree of Doctor of Philosophy (Chemical Engineering)

Faculty of Chemical Engineering and Natural Resources UNIVERSITI MALAYSIA PAHANG

MARCH 2016

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LIST OF SYMBOLS

m	Meter
Ppm	Part per million
Р	Fluid density
μ	Dynamic viscosity of fluid
D	Internal diameter
L	Pipe length
V	Average velocity
% DR	Percentage drag reduction
С	Concentration
L	Liter
ΔP	Pressure loss
TEM	Transmission Electronic Microscopy

LIST OF ABBREVIATIONS

Re	Reynolds Number
DR	Drag Reduction
DRA	Drag Reducing Agent
MW	Molecular weight
I.D	Internal diameter
ΔP	Pressure Drop (Pressure Different)
RDA	Rotating disk apparatus
°C	Degree Celsius
%	Percent
L	Liter
Wt	Weight
TEM	Transmission Electronic Microscopy
TAPS	Trans-Alaska Pipeline System
PS	pump station
bbl/d	

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ABSTRACT

In the past few decades, several passive and active techniques to enhance the flow in pipelines have been suggested by scientists and implemented by the oil and gas industry. The most commercially feasible flow enhancement (drag reduction) technique is the injection of minute quantities of viscoelastic polymeric additives in the main flow stream. At the same time, this technique comes with a major disadvantage: the polymeric additives are resistant to the high shear forces exerted by the pumps and/or the turbulence inside the pipe. The present work addresses the said problem by proposing an alternative technique that involves the formation of polymer-surfactant complexes to create a highly shear-resistant additive through physical interaction with oppositely and similarly charged surfactants. Polyacrylic acid (PAA), Polyacrylamideco-diallyl-dimethylammonium chloride P(AAm-co-DADMAC), hydroxypropyl cellulose (HPC), and polydiallyldimethylammoniumchloride (PDADMAC) polymers are adopted as drag-reducing agents (DRA). Sodium oleate and Tween 20 surfactants are also used as DRAs and complex creation agents. One of the major objectives of the present work is to prove that complexes can be formed even with similarly charged ingredients (i.e., polymers and surfactants). The experimental work is divided into three major phases. The first phase tests the flow behavior and shear resistance of the polymeric, surfactant, and complex DRAs using a rotating disk apparatus (RDA). The second phase detects the morphology of the formulated complexes using transmission electron microscopy (TEM) and cryo-TEM. The third phase conducts a pipeline drag reduction test using a closed-loop liquid circulation system, in which the pressure drop and flow rate measurements are taken to evaluate the drag reduction performance of a selected complex and its initial polymeric and surfactant substances. The RDA results show that, when tested at 700 ppm concentration and Re = 816650, all the polymeric additives have drag reduction potential with a maximum %DR of 16%, 32%, 40%, and 12% for PAA, P(AAm-co-DADMAC), HPC, and PDADMAC polymers, respectively. Moreover, when tested at 700 ppm concentration and Re = 816650, most of the surfactant additives show an acceptable drag reduction performance with a maximum %DR of 16% and 12% for sodium oleate and Tween 20 surfactants, respectively. The complexes created from the initial polymeric and surfactant additives significantly improve drag reduction performance and resistance to shear forces. The resistance of PAA is enhanced by 66% when tested at 500-ppm sodium oleate and Re = 489990. The resistance of PAA is massively enhanced by 203% when tested at 500-ppm sodium oleate and Re = 914648. The morphology of the formulated complexes is tested using TEM and cryo-TEM, and the results indicate that similarly charged polymer and surfactant molecules have the ability to form certain aggregates with the aid of the free counter ions in water. The TEM shows network-like aggregates that capture surfactant clusters in a network of polymers and small surfactant micelles. A similarly charged PAA-sodium oleate complex is tested in a pipeline system. The experimental results clearly indicate that the drag reduction performance of the polymers is massively improved by 51% when forming a complex with 1000-ppm concentration. In addition, the pressure drop reading results show that resistance to high-shear forces is highly modified when the complex is formed, and no detectable degradation is reported. It is believed that the same technique should be implemented using crude oil additives in the future due to the increasing need for such complexes in the oil and gas industry field.

ABSTRAK

Beberapa dekad yang lalu, beberapa teknik pasif dan aktif untuk meningkatkan aliran dalam saluran paip telah dicadangkan oleh ahli sains dan dilaksanakan oleh industri minyak dan gas. Teknik peningkatan aliran yang paling ekonomik dilaksanakan secara komersil (pengurangan seretan) adalah suntikan sekuantiti bahan tambahan polimer viscoelastic didalam arus aliran utama. Pada masa yang sama, teknik ini datang dengan satu kelemahan utama: bahan tambahan polimer dapat metahan daya ricih yang tinggi yang dikenakan oleh pam dan / atau kegeloraan di dalam paip. Kajian yang dikatakan boleh menangani masalah tersebut dengan mencadangkan teknik alternatif yang melibatkan pembentukan kompleks surfaktan polimer untuk mewujudkan aditif yang sangat tahan ricih melalui interaksi fizikal dengan surfaktan yang bercas berlawanan atau sama. Asid Polyacrylic (PAA), Polyacrylamide-co-diallyl-dimethylammonium klorida P (AAM-co-DADMAC), selulosa hydroxypropyl (HPC). dan polvdiallvldimethylammoniumchloride (PDADMAC) polimer yang digunakan sebagai agen pengurangan seretan (DRA). natrium oleat dan surfaktan Tween 20 juga digunakan sebagai DRAs dan ejen penciptaan kompleks. Salah satu objektif utama kajian ini ialah untuk membuktikan bahawa kompleks boleh dibentuk walaupun dengan bahan-bahan yang sama cas (iaitu, polimer dan surfaktan). Kerja-kerja eksperimen dibahagikan kepada tiga fasa utama. Fasa pertama menguji kelakuan aliran dan rintangan ricih polimer, surfaktan, dan kompleks DRAs menggunakan alat cakera berputar (RDA). Fasa kedua mengesan morfologi kompleks yang dirumuskan menggunakan mikroskop elektron penghantaran (TEM) dan Cryo-TEM. Fasa ketiga menjalankan ujian pengurangan seretan paip menggunakan sistem peredaran cecair gelung tertutup, di mana kejatuhan tekanan dan ukuran kadar aliran diambil untuk menilai prestasi pengurangan seretan kompleks terpilih dan bahan-bahan awal polimer dan surfaktan. Keputusan RDA menunjukkan bahawa, apabila diuji pada kepekatan 700 ppm dan Re = 816650, semua bahan tambahan polimer mempunyai keupayaan penurunan seretan dengan % DR maksimum sebanyak masing-masing pada 16%, 32%, 40%, dan 12% untuk PAA. P (AAM-co -DADMAC), HPC dan PDADMAC polimer. Selain itu, apabila diuji pada kepekatan 700 ppm dan Re = 816650, kebanyakan bahan tambahan surfaktan menunjukkan prestasi pengurangan seretan dengan maksumim % DR masing-masing pada 16% dan 12% untuk natrium oleat dan Tween 20 surfaktan. Kompleks dicipta daripada bahan tambahan polimer dan surfaktan awal menunjukkan peningkatkan prestasi yang ketara pada pengurangan seretan dan rintangan kepada daya ricih. Rintangan PAA dipertingkatkan ke 66% apabila diuji pada 500 ppm kalium oleat dan Re = 489990. Rintangan PAA dipertingkatkan secara besar sebanyak 203% apabila diuji pada 500 ppm natrium oleat dan Re = 914648. Morfologi kompleks dirumuskan diuji menggunakan TEM dan Cryo-TEM, dan keputusan menunjukkan bahawa molekul-molekul polimer dan surfaktan bercas sama mempunyai keupayaan untuk membentuk agregat tertentu dengan bantuan ion pembilang bebas didalam air. TEM menunjukkan agregat rangkaian yang menangkap kelompok surfaktan dalam rangkaian polimer dan misel kecil surfaktan. PAA natrium kompleks oleat bercas sama diuji dalam sistem saluran paip. Keputusan eksperimen jelas menunjukkan bahawa prestasi pengurangan seretan daripada polimer meningkat secara besar sebanyak 51% apabila membentuk kompleks dengan kepekatan 1000 ppm. Di samping itu, keputusan bacaan kejatuhan tekanan menunjukkan bahawa daya tahan terhadap kuasa-kuasa tinggi ricih diubahsuai apabila terbentuknya kompleks, dan tiada kemenurunan dilaporkan.

REFERENCES

- Abdulbari, H.A., Kamarulizam, N.S. and Nour, A. 2012. Grafted natural polymer as new drag reducing agent: An experimental approach. *Chemical Industry and Chemical Engineering Quarterly.* 18(3): 361-371.
- Al-Sarkhi, A. 2010. Drag reduction with polymers in gas-liquid/liquid-liquid flows in pipes:A literature review. *Journal of Natural Gas Science and Engineering*. 241-48.
- Al-Sarkhi, A. 2012. Effect of mixing on frictional loss reduction by drag reducing polymer in annular horizontal two-phase flows. *International Journal of Multiphase Flow*. **39**(0): 186-192.
- Al-Sarkhi, A. and Hanratty, T.J. 2001. Effect of drag-reducing polymers on annular gas–liquid flow in a horizontal pipe. *International Journal of Multiphase Flow*. 27(7): 1151-1162.
- Al-Yaari, M., Soleimani, A., Abu-Sharkh, B., Al-Mubaiyedh, U. and Al-sarkhi, A. 2009. Effect of drag reducing polymers on oil-water flow in a horizontal pipe. *International Journal of Multiphase Flow.* 35(6): 516-524.
- Alramadhni, S., Saleh, N. and Rassol, G. 2013. Experimental Study of Drag Reduction Phenomena within Pipe-Flow in a Closed Circuit System Using Surfactant Additives. *Eng. &Tech.Journal.*, **31**1-13.
- Arifuzzaman Khan, G., Saheruzzaman, M., Abdur Razzaque, S., Sakinul Islam, M., Shamsul Alam, M. and Mainul Islam, M. 2009. Grafting of acrylonitrile monomer onto bleached okra bast fibre and its textile properties. *Indian journal* of fibre & textile research. 34(4): 321.
- Arnold, C., Ulrich, S., Stoll, S., Marie, P. and Holl, Y. 2011. Monte Carlo simulations of surfactant aggregation and adsorption on soft hydrophobic particles. *Journal* of Colloid and Interface Science. 353(1): 188-195.
- Bandyopadhyay, P. 1986. Review—mean flow in turbulent boundary layers disturbed to alter skin friction. *Journal of fluids engineering*. **108**(2): 127-140.
- Bari, H.A.A., Kamarulizam, S.N. and Man, R.C. 2011. Investigating drag reduction characteristic using Okra Mucilage as new drag reduction agent. *Journal of Applied Sciences*. **11**2554-2561.
- Baron, A. and Quadrio, M. 1995. Turbulent drag reduction by spanwise wall oscillations. *Applied Scientific Research*. **55**(4): 311-326.
- Baron, A., Quadrio, M. and Vigevano, L. 1993. On the boundary layer/riblets interaction mechanisms and the prediction of turbulent drag reduction. *International journal of heat and fluid flow*. **14**(4): 324-332.

Bechert, D. and Hage, W. 2006. Flow Phenomena in Nature. WIT Press. 2457-458.

- Benzi, R. 2010. A short review on drag reduction by polymers in wall bounded turbulence. *Physica D.* **239**(14): 1338-1345.
- Bewersdorff, H.-W. (1990). Drag reduction in surfactant solutions *Structure of turbulence and drag reduction* (pp. 293-312): Springer.
- Brostow, W., Majumdar, S. and Singh, R.P. 1999. Drag reduction and solvation in polymer solutions. *Macromolecular rapid communications*. **20**(3): 144-147.
- Carpenter, P. and Garrad, A. 1985. The hydrodynamic stability of flow over Kramertype compliant surfaces. Part 1. Tollmien-Schlichting instabilities. *Journal of Fluid Mechanics*. **155**465-510.
- Chapman, B.G. 2005. Study of drag reduction by zwitterionic and non-ionic surfactants in low temperature ethylene glycol/water recirculation systems.
- Cheng, L., Mewes, D. and Luke, A. 2007. Boiling phenomena with surfactants and polymeric additives: A state-of-the-art review. *International Journal of Heat and Mass Transfer.* **50**(13–14): 2744-2771.
- Choi, H.J., Kim, C.A., Sohn, J.-I. and Jhon, M.S. 2000. An exponential decay function for polymer degradation in turbulent drag reduction. *Polymer Degradation and Stability*. **69**(3): 341-346.
- Choi, K.-S. 1989. Near-wall structure of a turbulent boundary layer with riblets. *Journal* of fluid mechanics. **208**417-458.
- Coustols, E. and Cousteix, J. (1990). Experimental investigation of turbulent boundary layers manipulated with internal devices: riblets *Structure of Turbulence and Drag Reduction* (pp. 577-584): Springer.
- Davidson, P. (2004). Turbulence. An introduction for scientists and engineers Oxford University Press: Oxford.
- De Angelis, E., Casciola, C.M. and Piva, R. 2002. DNS of wall turbulence: dilute polymers and self-sustaining mechanisms. *Computers & Fluids*. **31**(4–7): 495-507.
- De Gennes, P.-G. 1990. Introduction to polymer dynamics. CUP Archive.
- Den Toonder, J., Draad, A., Kuiken, G. and Nieuwstadt, F. 1995. Degradation effects of dilute polymer solutions on turbulent drag reduction in pipe flows. *Applied scientific research.* 55(1): 63-82.
- Den Toonder, J., Hulsen, M., Kuiken, G. and Nieuwstadt, F. 1997. Drag reduction by polymer additives in a turbulent pipe flow: numerical and laboratory experiments. *Journal of Fluid Mechanics*. **337**193-231.

- Deshmukh, S. and Singh, R. 1986. Drag reduction characteristics of graft copolymers of xanthangum and polyacrylamide. *Journal of applied polymer science*. **32**(8): 6163-6176.
- Dimitropoulos, C.D., Sureshkumar, R. and Beris, A.N. 1998. Direct numerical simulation of viscoelastic turbulent channel flow exhibiting drag reduction: effect of the variation of rheological parameters1. *Journal of Non-Newtonian Fluid Mechanics*. **79**(2–3): 433-468.
- Djenidi, L., Anselmet, F., Liandrat, J. and Fulachier, L. 1994. Laminar boundary layer over riblets. *Physics of Fluids (1994-present)*. **6**(9): 2993-2999.
- Drzazga, M., Gierczycki, A., Dzido, G. and Lemanowicz, M. 2013. Influence of Nonionic Surfactant Addition on Drag Reduction of Water Based Nanofluid in a Small Diameter Pipe. *Chinese Journal of Chemical Engineering*. 21(1): 104-108.
- Du, Y., Symeonidis, V. and Karniadakis, G. 2002. Drag reduction in wall-bounded turbulence via a transverse travelling wave. *Journal of Fluid Mechanics*. 4571-34.
- Dubief, Y., Terrapon, V.E., White, C.M., Shaqfeh, E.S., Moin, P. and Lele, S.K. 2005. New answers on the interaction between polymers and vortices in turbulent flows. *Flow, turbulence and combustion.* **74**(4): 311-329.
- El-Samni, O., Chun, H. and Yoon, H. 2007. Drag reduction of turbulent flow over thin rectangular riblets. *International Journal of Engineering Science*. **45**(2): 436-454.
- Gaard, S. and Isaksen, O.T. (2003). *Experiments With Various Drag Reducing Additives In Turbulent Flow In Dense Phase Gas Pipelines*. Paper presented at the PSIG Annual Meeting.
- Gad-el-Hak, M. 2001. Flow control: The future. Journal of Aircraft. 38(3): 402-418.
- Gadd, G. (1971). Encyclopedia of Polymer Science and Technology, Vol. 15 (pp. 224-253): Wiley-Interscience, New York.
- Gasljevic, K., Aguilar, G. and Matthys, E. 2007. Measurement of temperature profiles in turbulent pipe flow of polymer and surfactant drag-reducing solutions. *Physics of Fluids (1994-present)*. **19**(8): 083105.
- Ge, W. (2008). Studies on the nanostructure, rheology and drag reduction characteristics of drag reducing cationic surfactant solutions. (PhD), The Ohio State University.
- Goddard, E.D. and Ananthapadmanabhan, K. 1998. *Applications of polymer-surfactant systems*. Marcel Dekker: New York.

- Harada, A. and Kataoka, K. 2006. Supramolecular assemblies of block copolymers in aqueous media as nanocontainers relevant to biological applications. *Progress in Polymer Science*. **31**(11): 949-982.
- Hayder, A.A.B. 2006. Studying the effect of polycrylamide on increasing the percentage drag reduction established by using two types of suspended solids in water. Paper presented at the 11th APCChE, Kuala Lumpur.
- Holmberg, K., Shah, D.O. and Schwuger, M.J. 2002. *Handbook of applied surface and colloid chemistry*. Wiley New York.
- Hou, Y., Somandepalli, V.S. and Mungal, M. 2006. A technique to determine total shear stress and polymer stress profiles in drag reduced boundary layer flows. *Experiments in fluids.* 40(4): 589-600.
- Hoyt, J. 1972. A Freeman Scholar Lecture: The Effect of Additives on Fluid Friction. Journal of Fluids Engineering. 94(2): 258-285.
- Inaba, H., Haruki, N. and Horibe, A. 2000. Flow drag and heat transfer reduction of flowing water containing fibrous material in a straight pipe. *International Journal of Thermal Sciences*. **39**(1): 18-29.
- Indartono, Y., Usui, H., Suzuki, H. and Komoda, Y. 2005. Temperature and diameter effect on hydrodynamic characteristic of surfactant drag-reducing flows. *Korea-Australia Rheology Journal*. **17**(4): 157-164.
- Jaafar, A.a.P., R.J. 2011. Drag Reduction of Biopolymer Flows. *Journal of Applied Sciences*. **11**(9): 1544-1551.
- Janosi, I., Jan, D. and Szabo, K.T., T. 2004 Turbulent drag reduction in dam-break flows. *Exp. Fluids.* **37**(2): 219-229.
- Joseph, D. 1990. Fluid dynamics of viscoelastic fluids. Springer.
- Jung, W., Mangiavacchi, N. and Akhavan, R. 1992. Suppression of turbulence in wallbounded flows by high-frequency spanwise oscillations. *Physics of Fluids A: Fluid Dynamics (1989-1993).* 4(8): 1605-1607.
- Karami, H.R. and Mowla, D. 2012. Investigation of the effects of various parameters on pressure drop reduction in crude oil pipelines by drag reducing agents. *Journal* of Non-Newtonian Fluid Mechanics. **177–178**(0): 37-45.
- Kim, J.T., Kim, C.A., Zhang, K., Jang, C.H. and Choi, H.J. 2011. Effect of polymer– surfactant interaction on its turbulent drag reduction. *Colloids and Surfaces A: Physicochemical and Engineering Aspects.* **391**(1–3): 125-129.
- Kim, N.-J., Kim, S., Lim, S.H., Chen, K. and Chun, W. 2009. Measurement of drag reduction in polymer added turbulent flow. *International Communications in Heat and Mass Transfer.* 36(10): 1014-1019.

King, R.P. 2002. Introduction to practical fluid flow. Butterworth-Heinemann.

- Kit, E., Tsinober, A., Balint, J., Wallace, J. and Levich, E. 1987. An experimental study of helicity related properties of a turbulent flow past a grid. *Physics of Fluids* (1958-1988). **30**(11): 3323-3325.
- Kramer, M.O. 1960. Boundary-layer stabilization by distributed damping. J Aero. *Space Sci.* **27**(1): 69.
- Lee, K.-H., Zhang, K. and Choi, H.J. 2010. Time dependence of turbulent drag reduction efficiency of polyisobutylene in kerosene. *Journal of Industrial and Engineering Chemistry*. **16**(4): 499-502.
- Li, F.-C., Yu, B., Wei, J.-J. and Kawaguchi, Y. 2012. *Turbulent drag reduction by surfactant additives*. John Wiley & Sons.
- Li, Y.-H. and Plano, T. (1991). United State Patent No. 5,020,561.
- Lin, J.Z., Gao, Z.Y., Zhou, K. and Chan, T.L. 2006. Mathematical modeling of turbulent fibre suspension and successive iteration solution in the channel flow. *Applied Mathematical Modelling*. **30**(9): 1010-1020.
- Luettgen, C.O., Lindsay, J.D. and Stratton, R.A. 1991. Turbulent dispersion in pulp flow: preliminary results and implications for the mechanisms of fibre-turbulence interactions.
- Lumley, J.L. 1969. Drag reduction by additives. *Annual review of fluid mechanics*. **1**(1): 367-384.
- Malcher, T. and Gzyl-Malcher, B. 2012. Influence of polymer–surfactant aggregates on fluid flow. *Bioelectrochemistry*. **87**42-49.
- Manhart, M. 2003. Rheology of suspensions of rigid-rod like particles in turbulent channel flow. *Journal of Non-Newtonian Fluid Mechanics*. **112**(2): 269-293.
- Marhefka, J.N., Marascalco, P.J., Chapman, T.M., Russell, A.J. and Kameneva, M.V. 2006. Poly (N-vinylformamide) A Drag-Reducing Polymer for Biomedical Applications. *Biomacromolecules*. 7(5): 1597-1603.
- Marheineke, N. and Wegener, R. 2011. Modeling and application of a stochastic drag for fibres in turbulent flows. *International Journal of Multiphase Flow.* **37**(2): 136-148.
- Massah, H. and Hanratty, T. 1997. Added stresses because of the presence of FENE-P bead-spring chains in a random velocity field. *J. Fluid Mech.* **337**67-101.
- Mata, J., Patel, J., Jain, N., Ghosh, G. and Bahadur, P. 2006. Interaction of cationic surfactants with carboxymethylcellulose in aqueous media. *Journal of Colloid* and Interface Science. 297(2): 797-804.

- Matras, Z., Malcher, T. and Gzyl-Malcher, B. 2008. The influence of polymer– surfactant aggregates on drag reduction. *Thin Solid Films*. **516**(24): 8848-8851.
- McCormick, C.L., Hester, R.D., Morgan, S.E. and Safieddine, A.M. 1990. Watersoluble copolymers. 30. Effects of molecular structure on drag reduction efficiency. *Macromolecules*. 23(8): 2124-2131.
- McCORMICK, M.E. and Bhattacharyya, R. 1973. Drag reduction of a submersible hull by electrolysis. *Naval Engineers Journal*. **85**(2): 11-16.
- Min, T. and Choi, H. 2005. Combined effects of polymers and active blowing/suction on drag reduction. *Journal of Non-Newtonian Fluid Mechanics*. **131**(1–3): 53-58.
- Min, T., Yoo, J.Y., Choi, H. and Joseph, D.D. (2001). A role of elastic energy in *turbulent drag reduction by polymer additives*. Paper presented at the Turbulence and Shear Flow Phenomena, Second International Symposium.
- Mishra, A. and Pal, S. 2007. Polyacrylonitrile-grafted Okra mucilage: A renewable reservoir to polymeric materials. *Carbohydrate polymers*. **68**(1): 95-100.
- Moaven, K., Rad, M. and Taeibi-Rahni, M. 2013. Experimental investigation of viscous drag reduction of superhydrophobic nano-coating in laminar and turbulent flows. *Experimental Thermal and Fluid Science*. **51**(0): 239-243.
- Moussa, T., Tiu, C. and Sridhar, T. 1993. Effect of solvent on polymer degradation in turbulent flow. *Journal of non-newtonian fluid mechanics*. **48**(3): 261-284.
- Mowla, D. and Naderi, A. 2006. Experimental study of drag reduction by a polymeric additive in slug two-phase flow of crude oil and air in horizontal pipes. *Chemical Engineering Science*. **61**(5): 1549-1554.
- Mukerjee, P. and Mysels, K.J. (1971). Critical micelle concentrations of aqueous surfactant systems: DTIC Document.
- Mya, K.Y., Jamieson, A.M. and Sirivat, A. 2000. Effect of temperature and molecular weight on binding between poly (ethylene oxide) and cationic surfactant in aqueous solutions. *Langmuir*. **16**(15): 6131-6135.
- Myska, J. and Mik, V. 2003. Application of a drag reducing surfactant in the heating circuit. *Energy and Buildings*. **35**(8): 813-819.
- Nelson, J. and Hennion, L. 2003. Optimising the Production of Maturing Oil Fields Using Drag Reducing Agents in Water Injection Wells.
- Nikitin, N. 2000. On the mechanism of turbulence suppression by spanwise surface oscillations. *Fluid dynamics*. **35**(2): 185-190.

- Nisewanger, C. (1964). Flow noise and drag measurements of vehicle with compliant coating: DTIC Document.
- Noh, Y. and Fernando, H. 1991. Dispersion of suspended particles in turbulent flow. *Physics of Fluids A: Fluid Dynamics (1989-1993)*. **3**(7): 1730-1740.
- Nowak, M. 2003. Time-dependent drag reduction and ageing in aqueous solutions of a cationic surfactant. *Experiments in fluids*. **34**(3): 397-402.
- Oliver, D.R. and Young Hoon, A. 1968. *Two-phase non-Newtonian flow*. Trans. Inst.Chem. Eng.
- Orlandi, P. 1995. A tentative approach to the direct simulation of drag reduction by polymers. *Journal of Non-Newtonian Fluid Mechanics*. **60**(2–3): 277-301.
- Park, S.-R. and Wallace, J.M. 1993. The influence of instantaneous velocity gradients on turbulence properties measured with multi-sensor hot-wire probes. *Experiments in fluids.* 16(1): 17-26.
- Park, S.-R. and Wallace, J.M. 1994. Flow alteration and drag reduction by riblets in a turbulent boundary layer. *AIAA journal*. **32**(1): 31-38.
- Pérez-Gramatges, A., Matheus, C.R.V., Lopes, G., da Silva, J.C. and Nascimento, R.S.V. 2013. Surface and interfacial tension study of interactions between water-soluble cationic and hydrophobically modified chitosans and nonylphenol ethoxylate. *Colloids and Surfaces A: Physicochemical and Engineering Aspects.* **418**124-130.
- Péron, N., Mészáros, R., Varga, I. and Gilányi, T. 2007. Competitive adsorption of sodium dodecyl sulfate and polyethylene oxide at the air/water interface. *Journal of Colloid and Interface Science*. **313**(2): 389-397.
- Petkova, R., Tcholakova, S. and Denkov, N.D. 2013. Role of polymer–surfactant interactions in foams: Effects of pH and surfactant head group for cationic polyvinylamine and anionic surfactants. *Colloids and Surfaces A: Physicochemical and Engineering Aspects.* **438**174-185.
- Phukan, S., Kumar, P., Panda, J., Nayak, B., Tiwari, K. and Singh, R. 2001. Application of drag reducing commercial and purified guargum for reduction of energy requirement of sprinkler irrigation and percolation rate of the soil. *Agricultural water management.* 47(2): 101-118.
- Pojják, K. and Mészáros, R. 2011. Association between branched poly (ethyleneimine) and sodium dodecyl sulfate in the presence of neutral polymers. *Journal of colloid and interface science*. **355**(2): 410-416.
- Ptasinski, P., Boersma, B., Nieuwstadt, F., Hulsen, M., Van den Brule, B. and Hunt, J. 2003. Turbulent channel flow near maximum drag reduction: simulations, experiments and mechanisms. *Journal of Fluid Mechanics*. **490**251-291.

- Ptasinski, P., Nieuwstadt, F., Van Den Brule, B. and Hulsen, M. 2001. Experiments in turbulent pipe flow with polymer additives at maximum drag reduction. *Flow, Turbulence and Combustion.* **66**(2): 159-182.
- Puryear, F. 1962. Boundary layer control drag reduction by compliant surfaces. *David Taylor Model Basin, Bethesda, Maryland, Report No. 1668.*
- Qi, Y., Kesselman, E., Hart, D.J., Talmon, Y., Mateo, A. and Zakin, J.L. 2011. Comparison of oleyl and elaidyl isomer surfactant–counterion systems in drag reduction, rheological properties and nanostructure. *Journal of Colloid and Interface Science*. 354(2): 691-699.
- Quadrio, M. and Ricco, P. 2003. Initial response of a turbulent channel flow to spanwise oscillation of the walls. *Journal of Turbulence*. **4**(7): 1-23.
- Quadrio, M. and Ricco, P. 2004. Critical assessment of turbulent drag reduction through spanwise wall oscillations. *Journal of Fluid Mechanics*. **521**251-271.
- Rashidi-Alavijeh, M., Javadian, S., Gharibi, H., Moradi, M., Tehrani-Bagha, A.R. and Shahir, A.A. 2011. Intermolecular interactions between a dye and cationic surfactants: Effects of alkyl chain, head group, and counterion. *Colloids and Surfaces A: Physicochemical and Engineering Aspects.* 380(1–3): 119-127.
- Righetti, M. and Romano, G. 2004. Particle–fluid interactions in a plane near-wall turbulent flow. *Journal of Fluid Mechanics*. **505**,93-121.
- Ritter, H. and Messum, L. 1964. Water tunnel measurements of turbulent skin friction on six different compliant surfaces of 1 ft length. *Admiralty Res. Lab., ARL N.* **4**.
- Rosen, M.J. 2004. Micelle formation by surfactants. *Surfactants and Interfacial Phenomena, Third Edition*105-177.
- Roy, A. and Larson, R.G. 2005. A mean flow model for polymer and fibre turbulent drag reduction. *Applied Rheology*. **15**(6): 370-389.
- Safri, A. and Bouhadef, M. 2008. Experimental study of the reduction of pressure drop of flowing silt in horizontal pipes. *J. Eng. Applied Sci.* **3**(6): 476-481.
- SAIF, M.T.A., JOSEPH, D., Riccius, O. and Christodoulou, C. 1990. Drag reduction in pipes lined with riblets. AIAA journal. 28(10): 1697-1698.
- Savins, J.G. 1967. A stress-controlled drag-reduction phenomenon. *Rheologica Acta*. **6**(4): 323-330.
- Sellin, R., Hoyt, J. and Scrivener, O. 1982. The effect of drag-reducing additives on fluid flows and their industrial applications part 1: basic aspects. *Journal of Hydraulic Research.* 20(1): 29-68.

- Seto, S.P. 2005. *Investigation of Pipeline Drag Reducers in Aviation Turbine Fuels*. Aviation Vehicle Fuel, Lubricant, and Equipment Research Committee of the Coordinating Research Council.
- Shah, S.N., Ahmed, K. and Zhou, Y. 2006. Drag reduction characteristics in straight and coiled tubing — An experimental study. *Journal of Petroleum Science and Engineering*. 53(3–4): 179-188.
- Shanliang, Z., Jianzhong, L.I.N. and Weifeng, Z. 2007. Numerical Research on the Fibre Suspensions in a Turbulent T-shaped Branching Channel Flow*. *Chinese Journal of Chemical Engineering*. 15(1): 30-38.
- Shklovskii, B.I. 1999. Screening of a macroion by multivalent ions: correlation-induced inversion of charge. *Physical Review E*. **60**(5): 5802-5811.
- Singh, R. 1990. Polymer Flow Engineering. *Encyclopedia of fluid mechanics*. **9**425-480.
- Soleimani, A., Al-Sarkhi, A. and Hanratty, T.J. 2002. Effect of drag-reducing polymers on pseudo-slugs—interfacial drag and transition to slug flow. *International Journal of Multiphase Flow.* **28**(12): 1911-1927.
- Suksamranchit, S., Sirivat, A. and Jamieson, A.M. 2006. Polymer–surfactant complex formation and its effect on turbulent wall shear stress. *Journal of Colloid and Interface Science*. **294**(1): 212-221.
- Suzuki, Y. and Kasagi, N. 1994. Turbulent drag reduction mechanism above a riblet surface. *AIAA journal.* **32**(9): 1781-1790.
- Takashi, S., Guzman Manuel R, D., Hiroaki, M., Hiromoto, U. and Tatsuo, N. 2000. A Flow Visualization Study on the Mechanism of Turbulent Drag Reduction by Surfactants. *NIHON REOROJI GAKKAISHI* 28(1): 35-40.
- Tamano, S., Ikarashi, H., Morinishi, Y. and Taga, K. 2015. Drag reduction and degradation of nonionic surfactant solutions with organic acid in turbulent pipe flow. *Journal of Non-Newtonian Fluid Mechanics*. 215(0): 1-7.
- Tang, Y. and Clark, D. 1993. On near-wall turbulence-generating events in a turbulent boundary layer on a riblet surface. *Applied scientific research*. **50**(3-4): 215-232.
- Toms, A. 1946. Folkestone Warren landslips: research carried out in 1939 by the Southern Railway Company. *Institute of Civil Engineers, Railway paper*(19).
- Vancko Jr, R.M. (1997). Effect of a Drag Reducing Agent on Pressure Drop and Flow Regime Transitions in Multiphase Horizontal Low Pressure Pipelines. Ohio University.
- Virk, P., Mickley, H.S. and Smith, K. 1970. The ultimate asymptote and mean flow structure in Toms' phenomenon. *Journal of Applied Mechanics*. **37**(2): 488-493.

Virk, P.S. 1975. Drag reduction fundamentals. AIChE Journal. 21(4): 625-656.

- Virk, P.S., Merrill, E., Mickley, H., Smith, K. and Mollo-Christensen, E. 1967. The Toms phenomenon: turbulent pipe flow of dilute polymer solutions. *Journal of Fluid Mechanics.* **30**(02): 305-328.
- Viswanath, P. 2002. Aircraft viscous drag reduction using riblets. *Progress in Aerospace Sciences.* **38**(6): 571-600.
- Vlassopoulos, D. and Schowalter, W. 1994. Steady viscometric properties and characterization of dilute drag-reducing polymer solutions. *Journal of Rheology* (1978-present). **38**(5): 1427-1446.
- Vukoslavcevic, P., Wallace, J. and Balint, J.-L. 1992. Viscous drag reduction using streamwise-aligned riblets. *AIAA journal*. **30**(4): 1119-1122.
- Walsh, M.J. (1982). *Turbulent boundary layer drag reduction using riblets*. Paper presented at the AIAA, Aerospace Sciences Meeting.
- Walsh, M.J. 1983. Riblets as a viscous drag reduction technique. *AIAA journal.* **21**(4): 485-486.
- Walsh, M.J. 1990. Effect of detailed surface geometry on riblet drag reduction performance. *Journal of Aircraft.* **27**(6): 572-573.
- Warholic, M., Massah, H. and Hanratty, T. 1999. Influence of drag-reducing polymers on turbulence: effects of Reynolds number, concentration and mixing. *Experiments in fluids*. 27(5): 461-472.
- Xia, G., Liu, Q., Qi, J. and Xu, J. 2008. Influence of surfactant on friction pressure drop in a manifold microchannel. *International Journal of Thermal Sciences*. 47(12): 1658-1664.
- Xin, X., Xu, G., Wu, D., Li, Y. and Cao, X. 2007. The effect of CaCl2 on the interaction between hydrolyzed polyacrylamide and sodium stearate: Rheological property study. *Colloids and Surfaces A: Physicochemical and Engineering Aspects.* 305(1–3): 138-144.
- Yang, K.S., Choi, H.J., Kim, C.B., Kim, I.S. and Jhon, M.S. 1994. Characterization of turbulent drag reduction in rotating disk system. *Korean Journal of Chemical Engineering*. 11(1): 8-13.
- You, Z., Lin, J., Shao, X. and Zhang, W. 2004. Stability and drag reduction in transient channel flow of fibre suspension. *Chinese Journal of Chemical Engineering*. 12(3): 319-323.
- Yu, B. and Kawaguchi, Y. 2003. Effect of Weissenberg number on the flow structure: DNS study of drag-reducing flow with surfactant additives. *International Journal of Heat and Fluid Flow*. 24(4): 491-499.

- Yu, B., Li, F. and Kawaguchi, Y. 2004. Numerical and experimental investigation of turbulent characteristics in a drag-reducing flow with surfactant additives. *International Journal of Heat and Fluid Flow.* 25(6): 961-974.
- Yunus, A.C. and Cimbala, J.M. 2006. Fluid mechanics: fundamentals and applications. *International Edition, McGraw Hill Publication*185-201.
- Zadrazil, I., Bismarck, A., Hewitt, G. and Markides, C. 2012. Shear layers in the turbulent pipe flow of drag reducing polymer solutions. *Chemical Engineering Science*. **72**142-154.
- Zakin, J.L. and Lui, H.-L. 1983. Variables affecting drag reduction by nonionic surfactant additives. *Chemical Engineering Communications*. 23(1-3): 77-88.
- Zakin, J.L., Poreh, M., Brosh, A. and Warshavsky, M. 1971. *Exploratory study of friction reduction in slurry flows*. Paper presented at the Chemical Engineering Professional Symposium Series.
- Zhang, Y., Schmidt, J., Talmon, Y. and Zakin, J.L. 2005. Co-solvent effects on drag reduction, rheological properties and micelle microstructures of cationic surfactants. *Journal of Colloid and Interface Science*. 286(2): 696-709.