

Process optimization of methylene blue adsorption onto eggshell-treated palm oil fuel ash

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ABSTRACT

Palm oil fuel ash (POFA) and eggshell (EG) were explored towards efficient methylene blue (MB) removal. Calcined EG, with a high content of CaO was used as a low-cost activation replacement for the alkali treatment of POFA. The adsorption performance of EG-POFA was studied by comparing its performance with untreated POFA, calcined EG and commercially treated Ca-POFA, and the competency was following the order of EG < POFA < EG-POFA \approx Ca-POFA, indicating that EG effectively enhanced the efficiency of POFA. The response surface methodology (RSM) was executed to optimize the operating conditions of MB adsorption onto EG-POFA by investigating the effect of EG-POFA dosage (X1), initial pH (X2) and initial MB concentration (X3) towards MB removal (Y). X1 = 1.18 g/L, X2 = 6, and X3 = 182 mg/L, with Y = 80.36% were found to be optimal conditions. The experimental data disclosed that the MB adsorption onto EG-POFA was in accordance with the Langmuir isotherm model ($q_m = 714.29$ mg/g) and followed pseudo-second-order kinetic model. The thermodynamic study discovered the endothermic and spontaneous natures of MB adsorption onto EG-POFA. This study discovered the feasibility of EG as an eco-friendly activation alternative for the alkali treatment towards remarkable MB adsorption.

KEYWORDS: EG-POFA, Low-cost adsorbent, Methylene blue, Adsorption, Optimization