

Production of Syngas from Ethanol CO₂ Reforming on La-doped Cu/Al₂O₃: Impact of Promoter Loading

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Abstract. Incipient wetness impregnation (IWI) method was applied to prepared 10%Cu/Al₂O₃ whereas M%La-doped 10%Cu/Al₂O₃ (Mwt%= 1%, 2%, 3%, 4% and 5%) were synthesized by employing sequential IWI technique. The prepared catalysts were evaluated from ethanol CO₂ reforming (ECR) at 1023 K and stoichiometric feed ratio. Average crystallite size of CuO particle is reduced with La-promoter addition probably caused by lanthana dilution effect that prevent agglomeration from occur within CuO particles. H₂ reduction process produce complete CuO reduction and constant signal is appear beyond 525 K suggests that the catalysts were completely reduced beyond that temperature. 3%La catalyst identified as optimal promoter loading based on reactant conversions. C₂H₅OH and CO₂ conversions were achieved on 3%La loading is 87.6% and 55.1%, respectively. Carbon was identified on catalyst surface based on X-ray diffraction (XRD) and scanning electron microscopy (SEM).

Keywords: Ethanol dry reforming; La₂O₃; Syngas; Cu/Al₂O₃ catalyst; Hydrogen

INTRODUCTION

The diminution of fossil fuels for global energy necessities largely contribute to the environmental concerns such as air pollution as well as greenhouse emissions. Carbon dioxide (CO₂) discharged into the environment as one type of greenhouse gases lead to major environmental issues. In addition, fossil fuels known to be non-renewable resources that eventually can depleted in the future. Thus, extensive studies into renewables and environmentally friendly approach are on demand.

In addition, ethanol (C₂H₅OH) is gaining a lot of interests as a replacement of fossil fuels because of its low toxicity, low cost, high availability and renewability [1-3]. Additionally, C₂H₅OH is mostly produced from fermentation of biomass such as sugarcane, corn and wheat [4]. Furthermore, syngas (mixture of H₂ + CO) is rising as potential alternatives for sources of energy because of its renewable nature and nature-friendly [5]. Moreover, syngas can be employed for the downstream production of higher chain hydrocarbons through Fisher-Tropsch synthesis [6-7] and other petrochemical industries such as methanol, dimethyl ether (DME) and methyl tert-butyl ether (MTBE) [8-9].