

THERMAL OXIDATION AND
HYDROXYAPATITE COATING OF SLMed 316L
STAINLESS STEEL SUBSTRATES

RAHIL IZZATI BINTI MOHD ASRI

Master of Science

UNIVERSITI MALAYSIA PAHANG



SUPERVISOR'S DECLARATION

We hereby declare that we have checked this thesis and in our opinion, this thesis is adequate in terms of scope and quality for the award of the degree of Master of Science.

(Supervisor's Signature)

Full Name : ASSOC. PROF. DR. WAN SHARUZI BIN WAN HARUN

Position : ASSOCIATE PROFESSOR

Date :

(Co-supervisor's Signature)

Full Name : DR. MAS AYU BINTI HASSAN

Position : SENIOR LECTURER

Date :



STUDENT'S DECLARATION

I hereby declare that the work in this thesis is based on my original work except for quotations and citations which have been duly acknowledged. I also declare that it has not been previously or concurrently submitted for any other degree at Universiti Malaysia Pahang or any other institutions.

(Student's Signature)

Full Name : RAHIL IZZATI BINTI MOHD ASRI

ID Number : MMM 15003

Date :

THERMAL OXIDATION AND HYDROXYAPATITE COATING OF SLMed 316L
STAINLESS STEEL SUBSTRATES

RAHIL IZZATI BINTI MOHD ASRI

Thesis submitted in fulfillment of the requirements
for the award of the degree of
Master of Science

Faculty of Mechanical Engineering
UNIVERSITI MALAYSIA PAHANG

JUNE 2018

ACKNOWLEDGEMENTS

The time I have spent at Universiti Malaysia Pahang (UMP) was certainly fruitful and enjoyable. UMP not only opened the door to scientific research, but also broadened my view in many aspects to the world. I feel very grateful to spend part of my life in this wonderful place.

First of all, I would like to express my greatest attitude and my appreciation to my respected advisor, Assoc. Prof. Dr. Wan Sharuzi Wan Harun of the Faculty of Mechanical Engineering at UMP for giving me invaluable advices, guidance, constant encouragements and suggestions during the course of this research. The door to Assoc. Prof. Dr. Wan Sharuzi Wan Harun office was always open whenever I ran into a trouble spot or had a question about my research or writing. He consistently allowed this paper to be my own work, but steered me in the right the direction whenever he thought I needed it. It is certainly my pleasure of having Assoc. Prof. Dr. Wan Sharuzi Wan Harun as my advisor for this research and also having the golden opportunity to learn from him.

Secondly, I would like to thank the experts who were involved for this research: Dr. Mas Ayu Hassan, Dr. Yuli Panca Asmara, Dr. Tedi Kurniawan, Dr. Juliawati Alias and Dr. Jun Haslinda Hj Sharifuddin who are either directly or indirectly had helped towards the success of this research. Without their passionate participation and input, the research could not have been successfully conducted. Also thank to technical staffs from UMP, Hi-Tech Instruments, DRB Hicom University and Kolej Kemahiran Tinggi Mara Kuantan (KKTMKu) for their willingness to help me all this time.

Truly thanks to the main financial support of the Ministry of Higher Education (MOHE), Malaysia for funding the tuition fee under MyBrain15 scholarship program. Special thanks to all my teams from Green Research Advance Materials (GRAMs) for their caring hands help me a lot in my research. They have given me a lot of advices during my difficulties in completing this research. We were struggling together during this period.

Finally, I must express my very profound gratitude and sincere appreciation towards my beloved family especially my parents (Encik Mohd Asri Mat Zain and Puan Azemah Ali) and my fiancé (Megat Mohd Fitri) for their endless support and continuous understanding through my years of study. This accomplishment would not have been possible without them.

Thank you all.

TABLE OF CONTENT

DECLARATION	
TITLE PAGE	
ACKNOWLEDGEMENTS	ii
ABSTRAK	iii
ABSTRACT	iv
TABLE OF CONTENT	v
LIST OF TABLES	viii
LIST OF FIGURES	viii
LIST OF SYMBOLS	ixi
LIST OF ABBREVIATIONS	xiii
CHAPTER 1 INTRODUCTION	1
1.1 Research Background	1
1.2 Problem Statement	2
1.3 Research Objectives	4
1.4 Research Scopes	4
1.5 Thesis Outline	5
CHAPTER 2 LITERATURE REVIEW	6
2.1 Introduction	6
2.2 Metallic Materials as Biomedical Implants	6
2.3 Material Selection Criteria for Implant	8
2.4 Selective Laser Melting of 316L Stainless Steel	10

2.5	Current Issues with Metallic Biomaterials: Corrosion	12
2.5.1	Types of Corrosion in Metal Implants	15
2.5.2	Corrosion: <i>In Vitro</i> and <i>In Vivo</i> Studies	17
2.5.3	Effect of Implant corrosion to Human Body	19
2.6	Requirements of Implant Surface Modification	20
2.6.1	Hydroxyapatite Coating Materials for Implants Surface Modification	20
2.6.2	Sol-Gel Dip Coating Method	24
2.6.3	Introduction of Oxide Layer on Metallic Substrate by Thermal Oxidation	26
2.7	Summary of the Literature Review	30
CHAPTER 3 METHODOLOGY		32
3.1	Introduction	32
3.2	Substrates Preparation	34
3.2.1	The 316L SS Substrate	35
3.2.2	Fabrication of 316L SS Substrate	35
3.2.3	Post Processing of SLMed 316L SS Substrate	37
3.3	Thermal Oxidation Process	37
3.4	Sol-Gel Dip Coating Process	39
3.4.1	Sol-Gel Slurry	39
3.4.2	Dip Coating Process	39
3.4.3	Sintering Process	40
3.4.4	Preliminary Studies on Hydroxyapatite Coating Parameter	40
3.5	Physical Testing	41
3.6	Surface Characterisation	43
3.7	Corrosion Testing	46

3.8	Summary of the Methodology	48
CHAPTER 4 RESULTS AND DISCUSSION		49
4.1	Introduction	49
4.2	Powder Metal and Substrates Characterisation	49
4.3	Formation of the Oxide Layer by Thermal Oxidation	51
4.3.1	Substrate Physical Condition	52
4.3.2	Weight Increment	52
4.3.3	Surface Characterisation	53
4.3.4	Corrosion Behaviour	58
4.4	Sol-Gel Dip Coating	62
4.4.1	Preliminary Results on Hydroxyapatite Coating	64
4.4.2	Weight Increment	64
4.4.2	Surface Characterisation	65
4.4.3	Corrosion Behaviour	67
4.5	Discussions	70
4.6	Summary	71
CHAPTER 5 CONCLUSIONS AND RECOMMENDATIONS		72
5.1	Introduction	72
5.2	Conclusions	72
5.3	Knowledge Contribution	73
5.4	Recommendations	73
REFERENCES		75
APPENDIX A LIST OF PUBLICATIONS		92

LIST OF TABLES

Table 2.1	Four classes of metallic biomaterials and their primary applications as implants	7
Table 2.2	Elements in the human body	9
Table 2.3	Materials used total joint replacements	14
Table 2.4	Standards for corrosion resistance testing of biomaterials	15
Table 2.5	Composition of SBF's, Ringer's and Hank's Solution	18
Table 2.6	Composition of different artificial saliva	19
Table 2.7	HAp requirements as coating material	22
Table 2.8	Different techniques to deposit HAp coatings	23
Table 2.9	Possible surface oxide layers on various biocompatible metals	28
Table 3.1	Schematic diagram of four different conditions of SLMed 316L SS substrates	34
Table 3.2	Parameters for Selective Laser Melting (SLM) process in this study	36
Table 3.3	Parameters for dip coating process	39
Table 3.4	Chemical composition of Ringer's solution	48
Table 4.1	Chemical composition of 316L SS gas powder	50
Table 4.2	Particle size distribution of 316L SS gas atomised powder	50
Table 4.3	Density of SLMed 316L SS after fabrication	50
Table 4.4	Average thickness of oxide layer formation	54
Table 4.5	Electrochemical data obtained from potentiodynamic polarisation curves for non-oxidised and thermally oxidised substrates	60
Table 4.6	Electrochemical data obtained from potentiodynamic polarisation curves for coated non-oxidised and coated thermally oxidised substrates	69

LIST OF FIGURES

Figure 2.1	Schematics of selective laser melting (SLM)	11
Figure 2.2	Corrosion scale on a Charnley SS stem, and (b) pitting and corrosion of a Muller SS stem after implant removal	14
Figure 2.3	SEM microscopic images (at 100x) of samples after tests; (a) Pitting corrosion (b) Uniform corrosion	17
Figure 2.4	Flow chart of HAp coating preparation by a sol-gel process	25
Figure 2.5	Fundamental stages of dip coating technique	26
Figure 2.6	The Deal-Grove Model assumption on thermal oxidation reaction	27
Figure 2.7	SEM images of (a) bare SS in Ringer's solution (b) SS surface coated in un-passivated-HAp (c) surface of borate passivated in Ringer's solution (d) borate passivated-HAp coated surface in Ringer's solution	29
Figure 2.8	Time taken for regeneration of surface oxide layers for several alloys	30
Figure 3.1	Research flow chart	33
Figure 3.2	3D CAD design for SLMed 316L SS substrate	35
Figure 3.3	SLM 280 HL machine used to fabricate SLMed 316L SS substrates	36
Figure 3.4	SLMed fabricated 316L SS substrates fabricated via SLM process	36
Figure 3.5	(a) EDM Die Sinking Machine and (b) the process smoothing the substrate by copper electrode	37
Figure 3.6	Graph pattern used for TO conducted at 700 °C for 150h	38
Figure 3.7	The sol-gel dip coating process of the SLMed substrates	40
Figure 3.8	Weighing Machine Precisa Series 320 XB - Model XB220A	41
Figure 3.9	The density of substrate is measured using Weighing Machine	42
Figure 3.10	The sample is measured in (a) air and (b) water using Archimedes Principle to obtain density	42
Figure 3.11	Tabletop SEM with EDX equipment	43
Figure 3.12	Laser scattering particle analyzer	44
Figure 3.13	MEIJI light optical microscope	44
Figure 3.14	FESEM with EDX attachment used to examine the substrate morphology of oxide layer and HAp coating	45
Figure 3.15	X-ray diffractometer used to determine phase and composition analysis of the oxidised substrate and HAp coating	45
Figure 3.16	Electrodes and electrolyte used for electrochemical testing	46

Figure 3.17	Exposed area of working electrode before and after coated by acrylic painting for electrochemical corrosion testing	47
Figure 3.18	Experimental set-up for electrochemical corrosion testing	47
Figure 4.1	SEM image of 316L SS gas atomised powder	49
Figure 4.2	Particle size distribution of 316L SS gas atomised powder	50
Figure 4.3	OM image of SLMed fabricated 316L SS substrate after fabrication	51
Figure 4.4	SLMed 316L SS substrate; (a) before TO (b) after TO 150 h (c) after TO 200 h (d) after TO 250 h	52
Figure 4.5	Weight increment of SLMed 316L SS substrate before and after TO	53
Figure 4.6	SEM cross-sections on the non-oxidised; (a) and thermally oxidised at 700 °C for; (b) 150 h (c) 200 h (d) 250 h	54
Figure 4.7	EDX line analysis of the thermally oxidised at 700 °C for; (a) 150 h (b) 200 h (c) 250 h	55
Figure 4.8	EDX area analysis of the thermally oxidised at 700 °C for; (a, d)150 h (b, e) 200 h (c, f) 250 h	56
Figure 4.9	XRD analysis of the non-oxidised and thermally oxidised at 700 °C for 150 h, 200 h and 250 h	57
Figure 4.10	Variation of open-circuit potential with time of the non-oxidised and thermally oxidised SLMed substrates at 700 °C for 150 h, 200 h and 250 h	59
Figure 4.11	Potentiodynamic polarisation curves of the non-oxidised and thermally oxidised SLMed substrates at 700 °C for 150 h, 200 h and 250 h	60
Figure 4.12	Corrosion rates of the non-oxidised and thermally oxidised SLMed substrates at 700 °C for 150 h, 200 h and 250 h	61
Figure 4.13	XRD pattern of the supplied HAp powder	63
Figure 4.14	Morphology of (a) supplied HAp powder and (b) particle size of the HAp powder	63
Figure 4.15	Surface morphology of different number of dipping of HAp coating on SLMed substrates; (a) two times dipping, (b) three times dipping and (c) four times dipping	64
Figure 4.16	Weight increment of SLMed coated substrates for condition; (b) non-oxidised (C1) and (d) thermally oxidised (C2)	65
Figure 4.17	FESEM morphologies of HAp coating on SLMed coated substrates for condition; (a) non-oxidised (C1) and (c) thermally oxidised 150 h (C2); and FESEM cross-section of the HAp coating thickness on SLMed coated substrates for condition; (b) non-oxidised (C1) and (d) thermally oxidised 150 h (C2)	66

Figure 4.18	XRD patterns of the SLMed coated (a) non-oxidised (C1) and (b) thermally oxidised substrates (C2)	67
Figure 4.19	Variation of open-circuit potential with time of the non-oxidised substrate (B1), thermal oxidised substrate (B2_TO150 h), coated non-oxidised substrate substrates (C1) and coated thermally oxidised substrate (C2)	68
Figure 4.20	Potentiodynamic polarisation curves of SLMed non-oxidised substrate (B1), thermal oxidised substrate (B2_TO150 h), coated non-oxidised substrate substrates (C1) and coated thermally oxidised substrate (C2)	69
Figure 4.21	Corrosion rate of SLMed non-oxidised substrate (B1), thermal oxidised substrate (B2_TO150 h), coated non-oxidised substrate substrates (C1) and coated thermally oxidised substrate (C2)	70

LIST OF SYMBOLS

%	Percentage
°	Degree
2 θ	2 theta
Å	Angstrom
CR	Corrosion rate (mm/yr)
D ₁₀	Range of particle size diameter at 10 %
D ₅₀	Range of particle size diameter at 50 %
D ₉₀	Range of particle size diameter at 90 %
<i>E_{corr}</i>	Potential current density (mV vs SCE)
EW	Equivalent weight
<i>I_{corr}</i>	Corrosion current density (mA/cm ²)
K	Constant (3.27 x 10 ⁻³ mmg/μAcmyr)
λ	Wavelength

LIST OF ABBREVIATIONS

Al ₂ O ₃	Aluminium dioxide
Δ-Fe	Delta ferrite
316L SS	316L stainless steel
3D	Three dimensional
AM	Additive manufacturing
ASTM	American Society for Testing and Materials
CAD	Computer aided design
Ca-P	Calcium phosphate
Cr ₂ O ₃	Chromium oxide
EDM	Electrical discharge machining
EDX	Energy dispersive X-ray
FDA	Food and Drug Administration
Fe ₂ O ₃	Iron oxide
FESEM	Field emission scanning electron microscope
HAp	Hydroxyapatite
ICDD	The International for Diffraction Data
ISO	The International Organization for Standardization
OCP	Open-circuit potential
OM	Optical microscope
PMMA	Poly(methyl methacrylate)
SCE	Saturated calomel electrode
SEM	Scanning electron microscopy
SLM	Selective laser melting
TCP	Tricalcium phosphate
THA	Total Hip Arthroplasty
TiO ₂	Titanium oxide
TTCP	Tetracalcium phosphate
UK	United Kingdom
USA	United State of America
XRD	X-ray Diffraction
ZrO ₂	Zirconium oxide
γ-Fe	Austenite

THERMAL OXIDATION AND HYDROXYAPATITE COATING OF SLMed 316L
STAINLESS STEEL SUBSTRATES

RAHIL IZZATI BINTI MOHD ASRI

Thesis submitted in fulfillment of the requirements
for the award of the degree of
Master of Science

Faculty of Mechanical Engineering
UNIVERSITI MALAYSIA PAHANG

JUNE 2018

ABSTRAK

Keluli tahan karat 316L (316L SS) merupakan satu daripada biomaterial logam yang telah digunakan sebagai implan logam di dalam bidang bioperubatan sebagai pengganti atau pemulihan fungsi tisu-tisu atau organ-organ yang merosot. Dalam kajian ini, 316L SS telah dipilih sebagai substrat logam dan difabrikasikan dengan menggunakan teknologi-teknologi pembuatan budaya tambah (AM) melalui kaedah teknik laser lebur (SLM). Bersandarkan masalah kakisan 316L SS, pengoksidaan haba (TO) dan penyalutan *hydroxyapatite* (HAp) telah dilakukan dalam kajian ini. TO telah dijalankan pada 700 °C selama 150 jam, 200 jam dan 250 jam untuk membentuk lapisan oksida. 316L SS *SLMed* yang terma teroksida telah diselidiki berdasarkan pencirian permukaan dan sifat kakisan. Keadaan optimum bagi TO diteruskan dengan penyalutan HAp. Substrat 316L SS *SLMed* yang tidak teroksida dan terma teroksida disaluti *sol-gel* penyalutan celup dan disinter selama 1 jam pada 500 °C. Kesan daripada TO dan penyalutan HAp telah dikaji ke atas substrat 316L SS. Peratus berat substrat dan ketebalan lapisan oksida telah meningkat dengan peningkatan masa dedahan semasa TO. Kemunculan besi oksida (Fe_2O_3) sebagai lapisan oksida telah dikesan pada substrat terma teroksida. Namun begitu, kromium oksida (Cr_2O_3) telah dikesan pada substrat terma teroksida selama 150 jam. Dalam kajian ini, hanya TO 150 jam menunjukkan penambahbaikan dalam sifat kakisan kerana kehadiran lapisan-lapisan Fe_2O_3 dan Cr_2O_3 . Masa pendedahan yang lama menyebabkan tiada perubahan dalam sifat kakisan. Oleh yang demikian, terma teroksida 316L SS *SLMed* telah diteruskan kepada penyalutan HAp. Salutan HAp ke atas terma teroksida 316L SS *SLMed* menunjukkan salutan tebal dan bebas retakan. Berdasarkan keputusan kakisan, ianya terbukti bahawa kehadiran lapisan oksida dan salutan HAp mampu meningkatkan sifat kakisan terhadap 316L *SLMed*. Kadar kakisan bagi substrat 316L SS yang bersalut ternyata paling rendah berbanding keadaan substrat yang lain. Ternyata, kehadiran lapisan-lapisan oksida bagi substrat bersalut terma teroksida menunjukkan perbezaan ketara dalam sifat kakisan kerana kehadiran Cr_2O_3 dan Fe_2O_3 sebagai lapisan oksida. Kesimpulannya, lapisan-lapisan oksida dan salutan HAp yang terbentuk pada substrat 316L SS terma teroksida membantu dalam meningkatkan sifat kakisan untuk substrat 316L SS *SLMed*. Dengan kemunculan kaedah fabrikasi yang baru untuk biomaterial logam bersama lapisan oksida dan penyalutan HAp ternyata mampu membaik pulih kadar kakisan 316L SS. Nilai-nilai kadar kakisan bagi semua keadaan 316L *SLMed* juga boleh diterima berbanding dengan kaedah fabrikasi konvensional yang lain.

ABSTRACT

316L Stainless Steel (316L SS) is one of the metallic biomaterials that commonly applied as a metal implant in the biomedical field as substitute or function restoration of degenerated tissues or organs. In this study, 316L SS taken as the metal substrate and fabricated through to additive manufacturing (AM) technology which is via selective laser melting (SLM) process. Concerning the corrosion issue of 316L SS, thermal oxidation (TO) and hydroxyapatite (HAp) coating were introduced in this study. The TO was conducted at 700 °C for 150, 200 and 250 h to develop oxide layer formation. The thermally oxidised SLMed 316L SS substrates were investigated based on a surface characterisation and a corrosion behaviour. The optimum condition for TO then continues with the HAp coating application. Sol-gel dip coating coated the non-oxidised and thermally oxidised SLMed 316L SS substrates and sintered at 500 °C for 1 h. The effect of TO and HAp coating were examined on the SLMed 316L SS substrates. The weight percentage of the SLMed substrates and the thickness of oxide layer increased with the increment of soaking time during TO. The iron oxide (Fe_2O_3) was detected as the oxide layer on thermally oxidised substrates. However, chromium oxide (Cr_2O_3) was detected on thermally oxidised for 150 h. In this study, only TO for 150 h showed improvement in corrosion behaviour due to the presence of Fe_2O_3 and Cr_2O_3 layers. Prolonged soaking time shows no improvement in corrosion behaviour. Consequently, the thermally oxidised SLMed 316L SS substrates for condition 150 h have proceeded to coat with HAp. The HAp coating on thermally oxidised SLMed 316L SS substrates exhibited thicker coating deposition and free crack coatings surface compared to coated non-oxidised SLMed substrates. Based on corrosion result, it confirmed the formation of the oxide layer and HAp coating able to enhance corrosion resistance of the SLMed 316L SS. The corrosion rate for coated thermally oxidised SLMed substrates is the lowest compared to other substrate conditions. The presence of oxide layers on the coated thermally oxidised SLMed 316L SS substrates show a significant difference in the corrosion behaviour due to the presence of Cr_2O_3 and Fe_2O_3 layers. It concluded that the oxide layers and HAp coating formed on thermally oxidised SLMed substrate gave full support to enhance corrosion behaviour of the SLMed substrate. Introducing new fabrication method for metallic biomaterials with presence of oxide layer and HAp coating was able to improve corrosion behaviour of 316L SS. The values of corrosion rate for all SLMed 316L conditions were also acceptable and tolerable compared to others conventional fabrication methods.

REFERENCES

- Abdulhameed, Z. N. (2011). *Corrosion Behavior of Some Implant Alloys in Simulated Human Body Environment*. (Degree of Master in Science in Materials Engineering), Department of Materials engineering, University of Technology.
- Agarwal, S., Curtin, J., Duffy, B., & Jaiswal, S. (2016). Biodegradable magnesium alloys for orthopaedic applications: A review on corrosion, biocompatibility and surface modifications. *Materials Science and Engineering: C*.
- Agrawal, K., Singh, G., Prakash, S., & Puri, D. (2011). *Synthesis of HA by various sol-gel techniques and their comparison: a review*. Paper presented at the National Conference on Advancements and Futuristic Trends in Mechanical and Materials Engineering.
- Albayrak, O., El-Atwani, O., & Altintas, S. (2008). Hydroxyapatite coating on titanium substrate by electrophoretic deposition method: Effects of titanium dioxide inner layer on adhesion strength and hydroxyapatite decomposition. *Surface and Coatings Technology*, 202(11), 2482-2487.
- Antunes, R. A., & de Oliveira, M. C. L. (2012). Corrosion fatigue of biomedical metallic alloys: mechanisms and mitigation. *Acta Biomaterialia*, 8(3), 937-962.
- Asri, R., Harun, W., Hassan, M., Ghani, S., & Buyong, Z. (2016). A review of hydroxyapatite-based coating techniques: Sol-gel and electrochemical depositions on biocompatible metals. *Journal of the mechanical behavior of biomedical materials*, 57, 95-108.
- Ayu, H. M., Izman, S., Daud, R., Krishnamurithy, G., Shah, A., Tomadi, S. H., & Salwani, M. S. (2017). Surface Modification on CoCrMo Alloy to Improve the Adhesion Strength of Hydroxyapatite Coating. *Procedia Engineering*, 184, 399-408.
- Baino, F., & Potestio, I. (2016). Orbital implants: State-of-the-art review with emphasis on biomaterials and recent advances. *Materials Science and Engineering: C*, 69, 1410-1428.
- Bakan, F., Laçin, O., & Sarac, H. (2013). A novel low temperature sol-gel synthesis process for thermally stable nano crystalline hydroxyapatite. *Powder Technology*, 233, 295-302.
- Baker, R. M., Tseng, L.-F., Iannolo, M. T., Oest, M. E., & Henderson, J. H. (2016). Self-deploying shape memory polymer scaffolds for grafting and stabilizing complex bone defects: A mouse femoral segmental defect study. *Biomaterials*, 76, 388-398.
- Ballarre, J., Seltzer, R., Mendoza, E., Orellano, J. C., Mai, Y.-W., García, C., & Ceré, S. M. (2011). Morphologic and nanomechanical characterization of bone tissue growth around bioactive sol-gel coatings containing wollastonite particles applied on stainless steel implants. *Materials Science and Engineering: C*, 31(3), 545-552.

- Bandyopadhyay, A., Gualtieri, T., & Bose, S. (2015). Global Engineering and Additive Manufacturing. *Additive Manufacturing*, 1, 9-11.
- Baroux, B. (2002). Further Insights on the Pitting Corrosion of Stainless Steels. *Corrosion Technology-New York And Basel*, 17, 311-348.
- Bauer, S., Schmuki, P., von der Mark, K., & Park, J. (2013). Engineering biocompatible implant surfaces: Part I: Materials and surfaces. *Progress in Materials Science*, 58(3), 261-326.
- Behnamian. (2017). Corrosion behavior of alloy 316L stainless steel after exposure to supercritical water at 500 °C for 20,000 h. *The Journal of Supercritical Fluids*, 127, 191-199.
- Behnamian, Mostafaei, A., Kohandehghan, A., Amirkhiz, B. S., Li, J., Zheng, W., Luo, J. L. (2017). Characterization of oxide layer and micro-crack initiation in alloy 316L stainless steel after 20,000 h exposure to supercritical water at 500 °C. *Materials Characterization*, 131, 532-543.
- Ben-Nissan, B., Choi, A. H., Roest, R., Latella, B. A., & Bendavid, A. (2015). 2 - Adhesion of hydroxyapatite on titanium medical implants. In M. Mucalo (Ed.), *Hydroxyapatite (HAp) for Biomedical Applications* (pp. 21-51): Woodhead Publishing.
- Besra, L., & Liu, M. (2007). A review on fundamentals and applications of electrophoretic deposition (EPD). *Progress in Materials Science*, 52(1), 1-61.
- Bettini, E., Leygraf, C., & Pan, J. (2013). Nature of current increase for a CoCrMo alloy:" transpassive" dissolution vs. water oxidation. *INTERNATIONAL JOURNAL OF ELECTROCHEMICAL SCIENCE*, 8(10), 11791-11804.
- Bhola, R., Bhola, S. M., Mishra, B., & Olson, D. L. (2011). Corrosion in titanium dental implants/prostheses—a review. *Trends in Biomaterials and Artificial Organs*, 25(1), 34-46.
- Bigi, A., Boanini, E., Bracci, B., Facchini, A., Panzavolta, S., Segatti, F., & Sturba, L. (2005). Nanocrystalline hydroxyapatite coatings on titanium: a new fast biomimetic method. *Biomaterials*, 26(19), 4085-4089.
- Bobbert, F. S. L., Lietaert, K., Eftekhari, A. A., Pouran, B., Ahmadi, S. M., Weinans, H., & Zadpoor, A. A. (2017). Additively manufactured metallic porous biomaterials based on minimal surfaces: A unique combination of topological, mechanical, and mass transport properties. *Acta Biomaterialia*, 53, 572-584.
- Boopathi, S., & Kumar, A. R. (2014). Synthesis and Characterization of Nanostructured Hydroxyapatite using Sol-gel Method. *International Journal of ChemTech Research*, 6(3).
- Breme, H., Biehl, V., Reger, N., & Gawalt, E. (2016). a Metallic Biomaterials: Introduction *Handbook of Biomaterial Properties* (pp. 151-158): Springer.

- Bruschi, S., Bertolini, R., Bordin, A., Medea, F., & Ghiotti, A. (2016). Influence of the machining parameters and cooling strategies on the wear behavior of wrought and additive manufactured Ti6Al4V for biomedical applications. *Tribology International*, *102*, 133-142.
- Bryant, M., Farrar, R., Freeman, R., Brummitt, K., & Neville, A. (2013). Fretting corrosion characteristics of polished collarless tapered stems in a simulated biological environment. *Tribology International*, *65*, 105-112.
- Bryington, M. S., Hayashi, M., Kozai, Y., Vandeweghe, S., Andersson, M., Wennerberg, A., & Jimbo, R. (2013). The influence of nano hydroxyapatite coating on osseointegration after extended healing periods. *Dental Materials*, *29*(5), 514-520.
- Buscail, H., Messki, S. E., Riffard, F., Perrier, S., Cueff, R., Caudron, E., & Issartel, C. (2008). Characterization of the oxides formed at 1000°C on the AISI 316L stainless steel—Role of molybdenum. *Materials Chemistry and Physics*, *111*(2), 491-496.
- Calin, M., Gebert, A., Ghinea, A. C., Gostin, P. F., Abdi, S., Mickel, C., & Eckert, J. (2013). Designing biocompatible Ti-based metallic glasses for implant applications. *Materials Science and Engineering: C*, *33*(2), 875-883.
- Cao, H., & Liu, X. (2013). Plasma-Sprayed Ceramic Coatings for Osseointegration. *International Journal of Applied Ceramic Technology*, *10*(1), 1-10.
- Carayon, M., & Lacout, J. (2003). Study of the Ca/P atomic ratio of the amorphous phase in plasma-sprayed hydroxyapatite coatings. *Journal of Solid State Chemistry*, *172*(2), 339-350.
- Case, E., Smith, I., & Baumann, M. (2005). Microcracking and porosity in calcium phosphates and the implications for bone tissue engineering. *Materials Science and Engineering: A*, *390*(1-2), 246-254.
- Catauro, Papale, F., & Bollino, F. (2015). Characterization and biological properties of TiO₂/PCL hybrid layers prepared via sol-gel dip coating for surface modification of titanium implants. *Journal of Non-Crystalline Solids*, *415*(0), 9-15.
- Catauro, M., Bollino, F., Papale, F., Giovanardi, R., & Veronesi, P. (2014). Corrosion behavior and mechanical properties of bioactive sol-gel coatings on titanium implants. *Materials Science and Engineering: C*, *43*, 375-382.
- Catauro, M., Papale, F., Sapio, L., & Naviglio, S. (2016). Biological influence of Ca/P ratio on calcium phosphate coatings by sol-gel processing. *Materials Science and Engineering: C*, *65*, 188-193.
- Chaturvedi, T. (2009). An overview of the corrosion aspect of dental implants (titanium and its alloys). *Indian Journal of Dental Research*, *20*(1), 91.

- Chen, Guillemot, G., Gandin, C.-A., & Bellet, M. (2017). Three-dimensional finite element thermomechanical modeling of additive manufacturing by selective laser melting for ceramic materials. *Additive Manufacturing*, 16(Supplement C), 124-137. doi: <https://doi.org/10.1016/j.addma.2017.02.005>
- Chen, & Liu, X. (2016). Advancing biomaterials of human origin for tissue engineering. *Progress in polymer science*, 53, 86-168.
- Chen, & Thouas, G. A. (2015). Metallic implant biomaterials. *Materials Science and Engineering: R: Reports*, 87(0), 1-57.
- Chen, Yang, H.-Y., Abbott, T., Easton, M., & Birbilis, N. (2014). Corrosion protection of magnesium and its alloys by metal phosphate conversion coatings. *Surface Engineering*, 30(12), 871-879.
- Cleries, L., Martinez, E., Fernandez-Pradas, J., Sardin, G., Esteve, J., & Morenza, J. (2000). Mechanical properties of calcium phosphate coatings deposited by laser ablation. *Biomaterials*, 21(9), 967-971.
- Costa, D. O., Dixon, S. J., & Rizkalla, A. S. (2012). One-and three-dimensional growth of hydroxyapatite nanowires during sol-gel-hydrothermal synthesis. *ACS applied materials & interfaces*, 4(3), 1490-1499.
- Cvetkovski, K., Ahlström, J., Norell, M., & Persson, C. (2014). Analysis of wear debris in rolling contact fatigue cracks of pearlitic railway wheels. *Wear*, 314(1-2), 51-56.
- Dai, J., Zhu, J., Chen, C., & Weng, F. (2016). High temperature oxidation behavior and research status of modifications on improving high temperature oxidation resistance of titanium alloys and titanium aluminides: A review. *Journal of Alloys and Compounds*, 685, 784-798.
- Davis, J. R. (2001). Surface engineering for corrosion and wear resistance (Vol. 1): ASM international.
- Deal, B. E., & Grove, A. S. (1965). General Relationship for the Thermal Oxidation of Silicon. *Journal of Applied Physics*, 36(12), 3770.
- Delaunay, C., Petit, I., Learmonth, I., Oger, P., & Vendittoli, P. (2010). Metal-on-metal bearings total hip arthroplasty: the cobalt and chromium ions release concern. *Orthopaedics & Traumatology: Surgery & Research*, 96(8), 894-904.
- Demnati, I., Grossin, D., Errassifi, F., Combes, C., Rey, C., & Le Bolay, N. (2014). Synthesis of fluor-hydroxyapatite powder for plasma sprayed biomedical coatings: Characterization and improvement of the powder properties. *Powder Technology*, 255(0), 23-28.

- Di, W., Yongqiang, Y., Xubin, S., & Yonghua, C. (2012). Study on energy input and its influences on single-track, multi-track, and multi-layer in SLM. *The International Journal of Advanced Manufacturing Technology*, 58(9), 1189-1199.
- Dorozhkin, S. V. (2014). Calcium orthophosphate coatings on magnesium and its biodegradable alloys. *Acta Biomaterialia*, 10(7), 2919-2934.
- Espallargas, N., Torres, C., & Muñoz, A. I. (2015). A metal ion release study of CoCrMo exposed to corrosion and tribocorrosion conditions in simulated body fluids. *Wear*(0).
- Fathi, M. H., Salehi, M., Saatchi, A., Mortazavi, V., & Moosavi, S. (2003). In vitro corrosion behavior of bioceramic, metallic, and bioceramic–metallic coated stainless steel dental implants. *Dental Materials*, 19(3), 188-198.
- Faustini, M., Louis, B., Albouy, P. A., Kuemmel, M., & Grosso, D. (2010). Preparation of Sol–Gel Films by Dip-Coating in Extreme Conditions. *The Journal of Physical Chemistry C*, 114(17), 7637-7645.
- Fernandez-Pradas, J., Cleries, L., Martinez, E., Sardin, G., Esteve, J., & Morenza, J. (2001). Influence of thickness on the properties of hydroxyapatite coatings deposited by KrF laser ablation. *Biomaterials*, 22(15), 2171-2175.
- Freire, W. P., Fook, M. V. L., Barbosa, E. F., dos S Araújo, C., Barbosa, R. C., & Pinheiro, Í. M. (2015). *Biocompatibility of dental restorative materials*. Paper presented at the Materials Science Forum.
- Furko, M., Jiang, Y., Wilkins, T. A., & Balázsi, C. (2016). Electrochemical and morphological investigation of silver and zinc modified calcium phosphate bioceramic coatings on metallic implant materials. *Materials Science and Engineering: C*, 62, 249-259.
- Gao, W., Zhang, Y., Ramanujan, D., Ramani, K., Chen, Y., Williams, C. B., Zavattieri, P. D. (2015). The status, challenges, and future of additive manufacturing in engineering. *Computer-Aided Design*, 69, 65-89.
- García, C., Ceré, S., & Durán, A. (2004). Bioactive coatings prepared by sol–gel on stainless steel 316L. *Journal of Non-Crystalline Solids*, 348(0), 218-224.
- Geetha, M., Singh, A. K., Asokamani, R., & Gogia, A. K. (2009). Ti based biomaterials, the ultimate choice for orthopaedic implants – A review. *Progress in Materials Science*, 54(3), 397-425.
- Gepreel, M. A.-H., & Niinomi, M. (2013). Biocompatibility of Ti-alloys for long-term implantation. *Journal of the mechanical behavior of biomedical materials*, 20, 407-415.
- Gopi, Collins Arun Prakash, V., Kavitha, L., Kannan, S., Bhalaji, P. R., Shinyjoy, E., & Ferreira, J. M. F. (2011). A facile electrodeposition of hydroxyapatite onto borate passivated surgical grade stainless steel. *Corrosion Science*, 53(6), 2328-2334.

- Gopi, Karthika, A., Nithiya, S., & Kavitha, L. (2014). In vitro biological performance of minerals substituted hydroxyapatite coating by pulsed electrodeposition method. *Materials Chemistry and Physics*, 144(1), 75-85.
- Gopi, Prakash, V. C. A., & Kavitha, L. (2009). Evaluation of hydroxyapatite coatings on borate passivated 316L SS in Ringer's solution. *Materials Science and Engineering: C*, 29(3), 955-958.
- Gopi, Ramya, S., Rajeswari, D., & Kavitha, L. (2013). Corrosion protection performance of porous strontium hydroxyapatite coating on polypyrrole coated 316L stainless steel. *Colloids and Surfaces B: Biointerfaces*, 107(0), 130-136.
- Gray, J., & Luan, B. (2002). Protective coatings on magnesium and its alloys—a critical review. *Journal of Alloys and Compounds*, 336(1), 88-113.
- Gross, K., Chai, C., Kannagara, G., Ben-Nissan, B., & Hanley, L. (1998). Thin hydroxyapatite coatings via sol-gel synthesis. *Journal of Materials Science: Materials in Medicine*, 9(12), 839-843.
- Guillamet, R., Lopitiaux, J., Hannyer, B., & Lenglet, M. (1993). Oxidation of stainless steels (AISI 304 and 316) at high temperature. Influence on the metallic substratum. *Le Journal de Physique IV*, 3(C9), C9-349-C349-356.
- Güteryüz, H., & Çimenoglu, H. (2004). Effect of thermal oxidation on corrosion and corrosion-wear behaviour of a Ti-6Al-4V alloy. *Biomaterials*, 25(16), 3325-3333.
- Guoxin, H., Lixiang, Z., Yunliang, F., & Yanhong, L. (2008). Fabrication of high porous NiTi shape memory alloy by metal injection molding. *Journal of Materials Processing Technology*, 206(1-3), 395-399.
- Habib, K. A., Damra, M., Saura, J., Cervera, I., & Bellés, J. (2011). Breakdown and evolution of the protective oxide scales of AISI 304 and AISI 316 stainless steels under high-temperature oxidation. *International Journal of Corrosion*, 2011.
- Hanawa. (1999). In vivo metallic biomaterials and surface modification. *Materials Science and Engineering: A*, 267(2), 260-266.
- Hanawa, Asami, K., & Asaoka, K. (1998). Repassivation of titanium and surface oxide film regenerated in simulated bioliquid. *Journal of biomedical materials research*, 40(4), 530-538.
- Harun, W., Asri, R., Romlay, F., Sharif, S., Jan, N., & Tsumori, F. (2018). Surface characterisation and corrosion behaviour of oxide layer for SLMed-316L stainless steel. *Journal of Alloys and Compounds*.
- Hedberg, Y. S., Qian, B., Shen, Z., Virtanen, S., & Wallinder, I. O. (2014). In vitro biocompatibility of CoCrMo dental alloys fabricated by selective laser melting. *Dental Materials*, 30(5), 525-534.

- Hołowacz, I., Podbielska, H., Bauer, J., & Ulatowska-Jarża, A. (2005). Viscosity, surface tension and refractive index of tetraethylorthosilicate-based sol-gel materials depending on ethanol content. *Optica applicata*, 35(4).
- Hosseinalipour, S. M., Ershad-langroudi, A., Hayati, A. N., & Nabizade-Haghighi, A. M. (2010). Characterization of sol-gel coated 316L stainless steel for biomedical applications. *Progress in Organic Coatings*, 67(4), 371-374.
- Hou, P., Shi, C., Wu, L., & Hou, X. (2016). Chitosan/hydroxyapatite/Fe₃O₄ magnetic composite for metal-complex dye AY220 removal: Recyclable metal-promoted Fenton-like degradation. *Microchemical Journal*, 128, 218-225.
- Huang, C. H., Lai, J. J., Huang, J. C., Lin, C. H., & Jang, J. S. C. (2016). Effects of Cu content on electrochemical response in Ti-based metallic glasses under simulated body fluid. *Materials Science and Engineering: C*, 62, 368-376.
- Huang, C. H., Lai, J. J., Wei, T. Y., Chen, Y. H., Wang, X., Kuan, S. Y., & Huang, J. C. (2015). Improvement of bio-corrosion resistance for Ti₄₂Zr₄₀Si₁₅Ta₃ metallic glasses in simulated body fluid by annealing within supercooled liquid region. *Materials Science and Engineering: C*, 52, 144-150.
- Hussain, H. D., Ajith, S. D., & Goel, P. (2016). Nickel release from stainless steel and nickel titanium archwires – An in vitro study. *Journal of Oral Biology and Craniofacial Research*, 6(3), 213-218.
- Izman, S., Hassan, M., Kadir, M. R. A., Abdullah, M., Anwar, M., Shah, A., & Daud, R. (2012). *Effect of Pretreatment Process on Thermal Oxidation of Biomedical Grade Cobalt Based Alloy*. Paper presented at the Advanced Materials Research.
- Jamesh, M., Kumar, S., & Narayanan, T. S. (2012). Effect of thermal oxidation on corrosion resistance of commercially pure titanium in acid medium. *Journal of materials engineering and performance*, 21(6), 900-906.
- Jamesh, M., Narayanan, T. S., & Chu, P. K. (2013). Thermal oxidation of titanium: Evaluation of corrosion resistance as a function of cooling rate. *Materials Chemistry and Physics*, 138(2), 565-572.
- Ji, X., Lou, W., Wang, Q., Ma, J., Xu, H., Bai, Q., Liu, J. (2012). Sol-gel-derived hydroxyapatite-carbon nanotube/titania coatings on titanium substrates. *International journal of molecular sciences*, 13(4), 5242-5253.
- Jirarungsatian, C., & Prateepasen, A. (2010). Pitting and uniform corrosion source recognition using acoustic emission parameters. *Corrosion Science*, 52(1), 187-197.
- Kaluderović, M. R., Schreckenbach, J. P., & Graf, H.-L. (2016). Titanium dental implant surfaces obtained by anodic spark deposition – From the past to the future. *Materials Science and Engineering: C*.

- Kamachimudali, U., Sridhar, T., & Raj, B. (2003). Corrosion of bio implants. *Sadhana*, 28(3-4), 601-637.
- Kasemo, B., & Lausmaa, J. (1986). Surface science aspects on inorganic biomaterials. *CRC Crit. Rev. Clin. Neurobiol.;*(United States), 4.
- Kessler, V. G., Spijksma, G. I., Seisenbaeva, G. A., Håkansson, S., Blank, D. H., & Bouwmeester, H. J. (2006). New insight in the role of modifying ligands in the sol-gel processing of metal alkoxide precursors: A possibility to approach new classes of materials. *Journal of sol-gel science and technology*, 40(2-3), 163-179.
- Kim, I.-S., & Kumta, P. N. (2004). Sol-gel synthesis and characterization of nanostructured hydroxyapatite powder. *Materials Science and Engineering: B*, 111(2-3), 232-236.
- Kirk, P., & Pilliar, R. (1999). The deformation response of sol-gel-derived zirconia thin films on 316L stainless steel substrates using a substrate straining test. *Journal of materials science*, 34(16), 3967-3975.
- Koutsopoulos, S. (2002). Synthesis and characterization of hydroxyapatite crystals: a review study on the analytical methods. *Journal of biomedical materials research*, 62(4), 600-612.
- Koziolok, M., Grimm, M., Becker, D., Jordanov, V., Zou, H., Shimizu, J., Weitschies, W. (2015). Investigation of pH and Temperature Profiles in the GI Tract of Fasted Human Subjects Using the Intellicap® System. *Journal of Pharmaceutical Sciences*, 104(9), 2855-2863.
- Krawiec, H., Vignal, V., & Latkiewicz, M. (2016). Structure and electrochemical behaviour in the Ringer's solution at 25 °C of electrodeposited Co-Mo nanocrystalline coating. *Materials Chemistry and Physics*, 183, 121-130.
- Krawiec, H., Vignal, V., Loch, J., & Erasmus-Vignal, P. (2015). Influence of plastic deformation on the microstructure and corrosion behaviour of Ti-10Mo-4Zr and Ti-6Al-4V alloys in the Ringer's solution at 37 °C. *Corrosion Science*, 96, 160-170.
- Kuijjer, R., Jansen, E. J. P., Emans, P. J., Bulstra, S. K., Riesle, J., Pieper, J., Busscher, H. J. (2007). Assessing infection risk in implanted tissue-engineered devices. *Biomaterials*, 28(34), 5148-5154.
- Kumar, S., Narayanan, T. S. N. S., Ganesh Sundara Raman, S., & Seshadri, S. K. (2010). Surface modification of CP-Ti to improve the fretting-corrosion resistance: Thermal oxidation vs. anodizing. *Materials Science and Engineering: C*, 30(6), 921-927.
- Kumar, S., Narayanan, T. S. N. S., Raman, S. G. S., & Seshadri, S. K. (2009). Thermal oxidation of CP-Ti: Evaluation of characteristics and corrosion resistance as a function of treatment time. *Materials Science and Engineering: C*, 29(6), 1942-1949.

- Kuphasuk, C., Oshida, Y., Andres, C. J., Hovijitra, S. T., Barco, M. T., & Brown, D. T. (2001). Electrochemical corrosion of titanium and titanium-based alloys. *The Journal of prosthetic dentistry*, 85(2), 195-202.
- Kurtz, S. M., & Devine, J. N. (2007). PEEK biomaterials in trauma, orthopedic, and spinal implants. *Biomaterials*, 28(32), 4845-4869.
- Kwok, C. T., Wong, P. K., Cheng, F. T., & Man, H. C. (2009). Characterization and corrosion behavior of hydroxyapatite coatings on Ti6Al4V fabricated by electrophoretic deposition. *Applied Surface Science*, 255(13–14), 6736-6744.
- Li, Liao, H., & Hermansson, L. (1996). Sintering of partially-stabilized zirconia and partially-stabilized zirconia—hydroxyapatite composites by hot isostatic pressing and pressureless sintering. *Biomaterials*, 17(18), 1787-1790.
- Li, Qin, N., Zhang, L., Lv, J., Li, Q., & Luo, Y. (2016). Effects of different concentrations of metal ions on degradation of adenosine triphosphate in common carp (*Cyprinus carpio*) fillets stored at 4 °C: An in vivo study. *Food Chemistry*, 211, 812-818.
- Li, Shi, Y., Wang, Z., Wang, L., Liu, J., & Jiang, W. (2010). Densification behavior of gas and water atomized 316L stainless steel powder during selective laser melting. *Applied Surface Science*, 256(13), 4350-4356.
- Lidwell, O. M. (1993). Sir John Charnley, Surgeon (1911–1982): the control of infection after total joint replacement. *Journal of Hospital Infection*, 23(1), 5-15.
- Ling, R. S. M. (1989). Sir John Charnley memorial lecture: Mechanical factors in implant fixation. *Current Orthopaedics*, 3(3), 168-175.
- Liu, Troczynski, T., & Tseng, W. J. (2001). Water-based sol–gel synthesis of hydroxyapatite: process development. *Biomaterials*, 22(13), 1721-1730.
- Liu, Yang, Y., Mai, S., Wang, D., & Song, C. (2015). Investigation into spatter behavior during selective laser melting of AISI 316L stainless steel powder. *Materials & Design*, 87, 797-806.
- Liu, X., Poon, R. W., Kwok, S. C., Chu, P. K., & Ding, C. (2004). Plasma surface modification of titanium for hard tissue replacements. *Surface and Coatings Technology*, 186(1), 227-233.
- Ljungblad, U. (2014). Method for additive manufacturing: Google Patents.
- Luo, H., Dong, C., Xiao, K., & Li, X. (2011). Characterization of passive film on 2205 duplex stainless steel in sodium thiosulphate solution. *Applied Surface Science*, 258(1), 631-639.
- Man, H., Chiu, K., Cheng, F., & Wong, K. (2009). Adhesion study of pulsed laser deposited hydroxyapatite coating on laser surface nitrided titanium. *Thin Solid Films*, 517(18), 5496-5501.

- Manivasagam, G., Dhinasekaran, D., & Rajamanickam, A. (2010). Biomedical implants: Corrosion and its prevention-a review. *Recent Patents on Corrosion Science*, 2(1), 40-54.
- Marticorena, M., Corti, G., Olmedo, D., Guglielmotti, M., & Duhalde, S. (2007). *Laser surface modification of Ti implants to improve osseointegration*. Paper presented at the Journal of Physics: Conference Series.
- Matusiewicz, H. (2014). Potential release of in vivo trace metals from metallic medical implants in the human body: From ions to nanoparticles – A systematic analytical review. *Acta Biomaterialia*, 10(6), 2379-2403.
- Milošev, I., & Strehblow, H.-H. (2003). The composition of the surface passive film formed on CoCrMo alloy in simulated physiological solution. *Electrochimica Acta*, 48(19), 2767-2774.
- Mohd , M. F., Abdul Kadir, M. R., Iqbal, N., Hassan, M. A., & Hussain, R. (2014). Dipcoating of poly (ϵ -caprolactone)/hydroxyapatite composite coating on Ti6Al4V for enhanced corrosion protection. *Surface and Coatings Technology*, 245(0), 102-107.
- Mohseni, E., Zalnezhad, E., & Bushroa, A. R. (2014). Comparative investigation on the adhesion of hydroxyapatite coating on Ti–6Al–4V implant: A review paper. *International Journal of Adhesion and Adhesives*, 48(0), 238-257.
- Montanaro, L., Campoccia, D., & Arciola, C. R. (2007). Advancements in molecular epidemiology of implant infections and future perspectives. *Biomaterials*, 28(34), 5155-5168.
- Morais, L. S., Serra, G. G., Muller, C. A., Andrade, L. R., Palermo, E. F., Elias, C. N., & Meyers, M. (2007). Titanium alloy mini-implants for orthodontic anchorage: immediate loading and metal ion release. *Acta Biomaterialia*, 3(3), 331-339.
- Muley, S. V., Vidvans, A. N., Chaudhari, G. P., & Udainiya, S. (2016). An assessment of ultra fine grained 316L stainless steel for implant applications. *Acta Biomaterialia*, 30, 408-419.
- Mullins, M. M., Norbury, W., Dowell, J. K., & Heywood-Waddington, M. (2007). Thirty-Year Results of a Prospective Study of Charnley Total Hip Arthroplasty by the Posterior Approach. *The Journal of arthroplasty*, 22(6), 833-839.
- Munoz, A. I., Schwiesau, J., Jolles, B., & Mischler, S. (2015). In vivo electrochemical corrosion study of a CoCrMo biomedical alloy in human synovial fluids. *Acta Biomaterialia*, 21, 228-236.
- Nakai, M., Niinomi, M., Cho, K., & Narita, K. (2015). Enhancing functionalities of metallic materials by controlling phase stability for use in orthopedic implants *Interface Oral Health Science 2014* (pp. 79-91): Springer.

- Narayanan, T. S., Park, I. S., & Lee, M. H. (2014). Strategies to improve the corrosion resistance of microarc oxidation (MAO) coated magnesium alloys for degradable implants: prospects and challenges. *Progress in Materials Science*, 60, 1-71.
- Nasab, M. B., & Hassan, M. R. (2010). Metallic biomaterials of knee and hip-A review. *Trends Biomater. Artif. Organs*, 24(1), 69-82.
- Nayak, A. K. (2010). Hydroxyapatite synthesis methodologies: an overview. *International Journal of ChemTech Research*, 2(2), 903-907.
- Nercessian, O. A., Martin, G., Joshi, R. P., Su, B. W., & Eftekhari, N. S. (2005). A 15- to 25-Year Follow-Up Study of Primary Charnley Low-Friction Arthroplasty: A Single Surgeon Series. *The Journal of arthroplasty*, 20(2), 162-167.
- Niinomi, M., Nakai, M., & Hieda, J. (2012). Development of new metallic alloys for biomedical applications. *Acta Biomaterialia*, 8(11), 3888-3903.
- Oh, S., Han, M.-H., Im, W.-B., Kim, S. Y., Kim, K. H., You, C., & Ong, J. L. (2010). Surface characterization and dissolution study of biodegradable calcium metaphosphate coated by sol-gel method. *Journal of sol-gel science and technology*, 53(3), 627-633.
- Ohtsuka, T., Nishikata, A., Sakairi, M., & Fushimi, K. (2018). *Electrochemistry for Corrosion Fundamentals*: Springer.
- Okazaki, Y., & Gotoh, E. (2005). Comparison of metal release from various metallic biomaterials in vitro. *Biomaterials*, 26(1), 11-21.
- Onoki, T., & Hashida, T. (2006). New method for hydroxyapatite coating of titanium by the hydrothermal hot isostatic pressing technique. *Surface and Coatings Technology*, 200(24), 6801-6807.
- Ou, S.-F., Chiou, S.-Y., & Ou, K.-L. (2013). Phase transformation on hydroxyapatite decomposition. *Ceramics International*, 39(4), 3809-3816.
- Patel, B., Favaro, G., Inam, F., Reece, M. J., Angadji, A., Bonfield, W., Edirisinghe, M. (2012). Cobalt-based orthopaedic alloys: Relationship between forming route, microstructure and tribological performance. *Materials Science and Engineering: C*, 32(5), 1222-1229.
- Qiu, C., Yue, S., Adkins, N. J. E., Ward, M., Hassanin, H., Lee, P. D., Attallah, M. M. (2015). Influence of processing conditions on strut structure and compressive properties of cellular lattice structures fabricated by selective laser melting. *Materials Science and Engineering: A*, 628, 188-197.
- Qu, H., & Wei, M. (2008). Improvement of bonding strength between biomimetic apatite coating and substrate. *Journal of Biomedical Materials Research Part B: Applied Biomaterials*, 84(2), 436-443.

- Rafieerad, A. R., Ashra, M. R., Mahmoodian, R., & Bushroa, A. R. (2015). Surface characterization and corrosion behavior of calcium phosphate-base composite layer on titanium and its alloys via plasma electrolytic oxidation: A review paper. *Materials Science and Engineering: C*, 57, 397-413.
- Rajabi-Zamani, A. H., Behnamghader, A., & Kazemzadeh, A. (2008). Synthesis of nanocrystalline carbonated hydroxyapatite powder via nonalkoxide sol-gel method. *Materials Science and Engineering: C*, 28(8), 1326-1329.
- Rajesh, P., Muraleedharan, C., Komath, M., & Varma, H. (2011). Pulsed laser deposition of hydroxyapatite on titanium substrate with titania interlayer. *Journal of Materials Science: Materials in Medicine*, 22(3), 497-505.
- Rath, P. C., Besra, L., Singh, B. P., & Bhattacharjee, S. (2012). Titania/hydroxyapatite bi-layer coating on Ti metal by electrophoretic deposition: Characterization and corrosion studies. *Ceramics International*, 38(4), 3209-3216.
- Robin, A., Silva, G., & Rosa, J. L. (2013). Corrosion behavior of HA-316L SS biocomposites in aqueous solutions. *Materials Research*, 16(6), 1254-1259.
- Roest, R., Latella, B. A., Heness, G., & Ben-Nissan, B. (2011). Adhesion of sol-gel derived hydroxyapatite nanocoatings on anodised pure titanium and titanium (Ti6Al4V) alloy substrates. *Surface and Coatings Technology*, 205(11), 3520-3529.
- Rojaee, R., Fathi, M., Raeissi, K., & Sharifnabi, A. (2014). Biodegradation assessment of nanostructured fluoridated hydroxyapatite coatings on biomedical grade magnesium alloy. *Ceramics International*, 40(9), 15149-15158.
- Runa, M., Mathew, M., & Rocha, L. (2013). Tribocorrosion response of the Ti6Al4V alloys commonly used in femoral stems. *Tribology International*, 68, 85-93.
- Ryan, G., Pandit, A., & Apatsidis, D. P. (2006). Fabrication methods of porous metals for use in orthopaedic applications. *Biomaterials*, 27(13), 2651-2670.
- Sakamoto, A. (1991). Study of Furnace Atmosphere for Vacuum-Inert Gas Partial-Pressure Brazing. *Welding journal*, 11, 311.
- Sallica-Leva, E., Jardini, A., & Fogagnolo, J. (2013). Microstructure and mechanical behavior of porous Ti-6Al-4V parts obtained by selective laser melting. *Journal of the mechanical behavior of biomedical materials*, 26, 98-108.
- Sanchez, A. H. M., Luthringer, B. J. C., Feyerabend, F., & Willumeit, R. (2015). Mg and Mg alloys: How comparable are in vitro and in vivo corrosion rates? A review. *Acta Biomaterialia*, 13, 16-31.
- Sato, N. (2011). Basics of corrosion chemistry. *Green corrosion chemistry and engineering: Opportunities and challenges*, 1-32.

- Shabir, G. A. (2003). Validation of high-performance liquid chromatography methods for pharmaceutical analysis: Understanding the differences and similarities between validation requirements of the US Food and Drug Administration, the US Pharmacopeia and the International Conference on Harmonization. *Journal of chromatography A*, 987(1), 57-66.
- Shadanbaz, S., & Dias, G. J. (2012). Calcium phosphate coatings on magnesium alloys for biomedical applications: A review. *Acta Biomaterialia*, 8(1), 20-30.
- Shih, C.-C., Shih, C.-M., Su, Y.-Y., Su, L. H. J., Chang, M.-S., & Lin, S.-J. (2004). Effect of surface oxide properties on corrosion resistance of 316L stainless steel for biomedical applications. *Corrosion Science*, 46(2), 427-441.
- Sidane, Khireddine, H., Yala, S., Ziani, S., Bir, F., & Chicot, D. (2015). Morphological and mechanical properties of hydroxyapatite bilayer coatings deposited on 316L SS by sol-gel method. *Metallurgical and Materials Transactions B*, 46(5), 2340-2347.
- Sidane, Rammal, H., Beljebbar, A., Gangloff, S. C., Chicot, D., Velard, F., Kerdjoudj, H. (2017). Biocompatibility of sol-gel hydroxyapatite-titania composite and bilayer coatings. *Materials Science and Engineering: C*, 72, 650-658.
- Simka, W., Kaczmarek, M., Baron-Wiecheć, A., Nawrat, G., Marciniak, J., & Żak, J. (2010). Electropolishing and passivation of NiTi shape memory alloy. *Electrochimica Acta*, 55(7), 2437-2441.
- Simonelli, M., Tuck, C., Aboulkhair, N. T., Maskery, I., Ashcroft, I., Wildman, R. D., & Hague, R. (2015). A Study on the Laser Spatter and the Oxidation Reactions During Selective Laser Melting of 316L Stainless Steel, Al-Si10-Mg, and Ti-6Al-4V. *Metallurgical and materials transactions A*, 46(9), 3842-3851.
- Singh, & Lien Lo, S. (2016). Catalytic performance of hierarchical metal oxides for peroxidative degradation of pyridine in aqueous solution. *Chemical Engineering Journal*.
- Singh, Singh, S., & Prakash, S. (2011). Surface characterization of plasma sprayed pure and reinforced hydroxyapatite coating on Ti6Al4V alloy. *Surface and Coatings Technology*, 205(20), 4814-4820.
- Sivakumar, B., Singh, R., & Pathak, L. C. (2015). Corrosion behavior of titanium boride composite coating fabricated on commercially pure titanium in Ringer's solution for bioimplant applications. *Materials Science and Engineering: C*, 48, 243-255.
- Song, Shan, D. Y., & Han, E. H. (2008). Electrodeposition of hydroxyapatite coating on AZ91D magnesium alloy for biomaterial application. *Materials Letters*, 62(17-18), 3276-3279.
- Song, Wang, H.-d., Xu, B.-s., & Xing, Z.-g. (2016). Effect of fretting wear on very high cycle bending fatigue behaviors of FV520B steel. *Tribology International*, 103, 132-138.

- Souier, T., Martin, F., Bataillon, C., & Cousty, J. (2010). Local electrical characteristics of passive films formed on stainless steel surfaces by current sensing atomic force microscopy. *Applied Surface Science*, 256(8), 2434-2439.
- Sridhar, Vinodhini, S., Kamachi Mudali, U., Venkatachalapathy, B., & Ravichandran, K. (2016). Load-bearing metallic implants: electrochemical characterisation of corrosion phenomena. *Materials Technology*, 1-14.
- Sridhar, T. M., Kamachi Mudali, U., & Subbaiyan, M. (2003). Preparation and characterisation of electrophoretically deposited hydroxyapatite coatings on type 316L stainless steel. *Corrosion Science*, 45(2), 237-252.
- Subramanian, B. (2015). In vitro corrosion and biocompatibility screening of sputtered Ti₄₀Cu₃₆Pd₁₄Zr₁₀ thin film metallic glasses on steels. *Materials Science and Engineering: C*, 47, 48-56.
- Sun, Huang, Y., Fan, H., Wang, Y., Ning, Z., Liu, F., Chen, J. J. J. (2015). In vitro and in vivo biocompatibility of an Ag-bearing Zr-based bulk metallic glass for potential medical use. *Journal of Non-Crystalline Solids*, 419, 82-91.
- Sun, M., Wu, X., Zhang, Z., & Han, E.-H. (2009). Oxidation of 316 stainless steel in supercritical water. *Corrosion Science*, 51(5), 1069-1072.
- Surmenev, R. A. (2012). A review of plasma-assisted methods for calcium phosphate-based coatings fabrication. *Surface and Coatings Technology*, 206(8), 2035-2056.
- Tahmasbi, Armin, Solati-Hashjin, M., Osman, N. A. A., & Faghihi, S. (2014). Improved biophysical performance of hydroxyapatite coatings obtained by electrophoretic deposition at dynamic voltage. *Ceramics International*, 40(8, Part B), 12681-12691.
- Toque, J. A., Herliansyah, M. K., Hamdi, M., Ide-Ektessabi, A., & Sopyan, I. (2010). Adhesion failure behavior of sputtered calcium phosphate thin film coatings evaluated using microscratch testing. *Journal of the mechanical behavior of biomedical materials*, 3(4), 324-330.
- Trepanier, C., Tabrizian, M., Yahia, L. H., Bilodeau, L., & Piron, D. L. (1999). Effect of modification of oxide layer on NiTi stent corrosion resistance. *Journal of biomedical materials research*, 48(1), 96-98.
- Underwood, E. (2012). Trace elements in human and animal nutrition: Elsevier.
- Unsal, T., Ilhan-Sungur, E., Arkan, S., & Cansever, N. (2016). Effects of Ag and Cu ions on the microbial corrosion of 316L stainless steel in the presence of *Desulfovibrio* sp. *Bioelectrochemistry*, 110, 91-99.
- Vaithilingam, J., Prina, E., Goodridge, R. D., Hague, R. J. M., Edmondson, S., Rose, F. R. A. J., & Christie, S. D. R. (2016). Surface chemistry of Ti6Al4V components fabricated using

- selective laser melting for biomedical applications. *Materials Science and Engineering: C*, 67, 294-303.
- Visan, S., & Popescu, R. F. (2011). Biomaterials. The Behavior of Stainless Steel as a Biomaterial. *Economia. Seria Management*, 14, 177-183.
- Wang. (2015). Study on torsional fretting wear behavior of a ball-on-socket contact configuration simulating an artificial cervical disk. *Materials Science and Engineering: C*, 55, 22-33.
- Wang, Wang, L.-f., Zhu, M.-l., Zhang, J.-s., & Lei, M.-k. (2006). Influence of high-intensity pulsed ion beams irradiation on oxidation behavior of 316L stainless steel at 700°C. *Transactions of Nonferrous Metals Society of China*, 16, s676-s680.
- Wang, Wu, G., Lin, X., & Liu, Y. (2015). Coatings for osseointegration of metallic biomaterials. *Surface Coating and Modification of Metallic Biomaterials*, 345.
- Wang, Xu, S., Zhou, S., Xu, W., Leary, M., Choong, P., Xie, Y. M. (2016). Topological design and additive manufacturing of porous metals for bone scaffolds and orthopaedic implants: A review. *Biomaterials*, 83, 127-141.
- Wang, Yang, Y., Yi, Z., & Su, X. (2013). Research on the fabricating quality optimization of the overhanging surface in SLM process. *The International Journal of Advanced Manufacturing Technology*, 65(9-12), 1471-1484.
- Wang, Z.-L., & Zeng, R.-C. (2010). Comparison in characterization of composite and sol-gel coating on AZ31 magnesium alloy. *Transactions of Nonferrous Metals Society of China*, 20, s665-s669.
- Wataha, J. C. (2000). Biocompatibility of dental casting alloys: a review. *The Journal of prosthetic dentistry*, 83(2), 223-234.
- Wataha, J. C. (2002). Alloys for prosthodontic restorations. *The Journal of prosthetic dentistry*, 87(4), 351-363.
- Williams, J. A. (2005). Wear and wear particles—some fundamentals. *Tribology International*, 38(10), 863-870.
- Xiao, & Liu, R. F. (2006). Effect of suspension stability on electrophoretic deposition of hydroxyapatite coatings. *Materials Letters*, 60(21), 2627-2632.
- Xiao, M., Chen, Y. M., Biao, M. N., Zhang, X. D., & Yang, B. C. (2017). Bio-functionalization of biomedical metals. *Materials Science and Engineering: C*, 70, 1057-1070.
- Yan, C., Hao, L., Hussein, A., & Young, P. (2015). Ti-6Al-4V triply periodic minimal surface structures for bone implants fabricated via selective laser melting. *Journal of the mechanical behavior of biomedical materials*, 51, 61-73.

- Yan, C., Hao, L., Hussein, A., Young, P., & Raymont, D. (2014). Advanced lightweight 316L stainless steel cellular lattice structures fabricated via selective laser melting. *Materials & Design*, 55, 533-541. doi: <http://dx.doi.org/10.1016/j.matdes.2013.10.027>
- Yang, Kim, K.-H., & Ong, J. L. (2005). A review on calcium phosphate coatings produced using a sputtering process—an alternative to plasma spraying. *Biomaterials*, 26(3), 327-337.
- Yang, Ong, J. L., & Tian, J. (2003). Deposition of highly adhesive ZrO₂ coating on Ti and CoCrMo implant materials using plasma spraying. *Biomaterials*, 24(4), 619-627. doi: [http://dx.doi.org/10.1016/S0142-9612\(02\)00376-9](http://dx.doi.org/10.1016/S0142-9612(02)00376-9)
- Yang, & Ren, Y. (2010). Nickel-free austenitic stainless steels for medical applications. *Science and Technology of Advanced Materials*, 11(1), 014105.
- Yang, H., Yang, K., & Zhang, B. (2007). Pitting corrosion resistance of La added 316L stainless steel in simulated body fluids. *Materials Letters*, 61(4), 1154-1157.
- Yang, Y. C., & Chou, B. Y. (2007). Bonding strength investigation of plasma-sprayed HA coatings on alumina substrate with porcelain intermediate layer. *Materials Chemistry and Physics*, 104(2-3), 312-319.
- Yu, X., Jiang, Z., Wei, D., Zhou, C., Huang, Q., & Yang, D. (2013). Tribological properties of magnetite precipitate from oxide scale in hot-rolled microalloyed steel. *Wear*, 302(1-2), 1286-1294.
- Yuan, Q., & Golden, T. D. (2009). Electrochemical study of hydroxyapatite coatings on stainless steel substrates. *Thin Solid Films*, 518(1), 55-60.
- Yun, Y., Dong, Z., Lee, N., Liu, Y., Xue, D., Guo, X., Heineman, W. (2009). Revolutionizing biodegradable metals. *Materials Today*, 12(10), 22-32.
- Zeng, R.-c., Zhang, J., Huang, W.-j., Dietzel, W., Kainer, K., Blawert, C., & Wei, K. (2006). Review of studies on corrosion of magnesium alloys. *Transactions of Nonferrous Metals Society of China*, 16, s763-s771.
- Zhang, Guan, R. F., & Zhang, X. P. (2011). Synthesis and characterization of sol-gel hydroxyapatite coatings deposited on porous NiTi alloys. *Journal of Alloys and Compounds*, 509(13), 4643-4648.
- Zhang, Xianting, Z., Yongsheng, W., Kui, C., & Wenjian, W. (2006). Adhesion strength of sol-gel derived fluoridated hydroxyapatite coatings. *Surface and Coatings Technology*, 200(22), 6350-6354.
- Zhao, J., Xu, D., Shahzad, M. B., Kang, Q., Sun, Y., Sun, Z., Yang, K. (2016). Effect of surface passivation on corrosion resistance and antibacterial properties of Cu-bearing 316L stainless steel. *Applied Surface Science*.

Zhao, Y., Wong, S. M., Wong, H. M., Wu, S., Hu, T., Yeung, K. W., & Chu, P. K. (2013). Effects of carbon and nitrogen plasma immersion ion implantation on in vitro and in vivo biocompatibility of titanium alloy. *ACS applied materials & interfaces*, 5(4), 1510-1516.