

PERFORMANCE OF COAGULATION,  
ADSORPTION, ULTRAFILTRATION AND  
HYBRID TREATMENT SYSTEM ON  
KAOLIN/HUMIC ACID REMOVAL

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MASTER OF SCIENCE

UNIVERSITI MALAYSIA PAHANG



## **SUPERVISOR'S DECLARATION**

I hereby declare that I have checked this thesis and in my opinion, this thesis is adequate in terms of scope and quality for the award of the Master of Science

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## **STUDENT'S DECLARATION**

I hereby declare that the work in this thesis is based on my original work except for quotations and citations which have been duly acknowledged. I also declare that it has not been previously or concurrently submitted for any other degree at Universiti Malaysia Pahang or any other institutions.

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## ABSTRAK

Proses rawatan air konvensional yang sedia ada tidak dapat memenuhi kualiti air minuman. Penggunaan bahan kimia dalam proses rawatan akan menjejaskan kualiti air kerana ia menghasilkan produk sampingan. Selaras dengan ini, penyelidikan ini bertujuan untuk mengurangkan penggunaan bahan kimia dalam proses pengentalan, maka proses hibrid atau proses gabungan telah dinilai dalam projek penyelidikan ini. Larutan kaolin dan asid humik telah digunakan sebagai bahan ketidaktulenen yang biasanya didapati dalam sumber air semulajadi dan proses rawatan air ini dinilai dalam bentuk penyingkiran kekeruhan (kaolin) dan penyingkiran warna pada  $UV_{254}$  (asid humik). Dalam kajian ini, tiga proses individu yang berbeza (pengentalan, penjerapan, dan membran penurasan ultra) dan tiga proses hibrid (pengentalan-penurasan ultra, pengentalan-penjerapan, dan pengentalan-penjerapan-penurasan ultra) telah dilaksanakan. Bagi proses penjerapan, bahan tumbuhan semulajadi iaitu kulit manggis telah dipilih dalam penyelidikan ini sebagai bahan penjerap. Proses yang digunakan dalam kajian ini adalah ujian balang dan proses pemisahan membran untuk pembekuan / penjerapan dan ultra penurasan masing-masing. Air kaolin dan air HA-kaolin digunakan sebagai air ujian dalam kajian ini. Parameter yang dikaji mengandungi kekeruhan dan  $UV_{254}$  dan ciri-ciri membran yang digunakan dan kulit manggis. Dalam proses rawatan individu, proses pengentalan didapati sangat berkesan menyingkirkan kekotoran di dalam air pada nilai pH yang tinggi sehingga mencapai kecekapan penyingkiran 97% kekeruhan dan 93%  $UV_{254}$ . Manakala, penggunaan kulit manggis dalam penjerapan kurang berkesan dalam menyingkirkan kekeruhan dan warna  $UV_{254}$ . Membran penurasan ultra didapati dapat menyingkirkan kedua-dua parameter kekeruhan dan warna  $UV_{254}$  sehingga melebihi 99%. Walau bagaimanapun, masalah pengotoran (fouling) telah menyebabkan fluks air terawat menurun cepat dalam 6 minit. Proses rawatan hibrid telah menunjukkan bahawanya adalah satu kaedah rawatan air yang lebih berkesan dan mampan berbanding rawatan individu. Untuk rawatan air melalui pengentalan-membran penurasan ultra, fluks relatif meningkat sehingga 12 dan 30 kali ganda kepada air kaolin dan air HA-kaolin apabila dibandingkan dengan proses membran penurasan ultra. Proses pengentalan-penjerapan pada pH yang tinggi menunjukkan kualiti air terawat yang lebih baik. Pada pH 8 dengan 0.1 g penjerap kulit manggis dapat mencapai penyingkiran sehingga 99% kekeruhan dan  $UV_{254}$ . Bagi proses gabungan semua kaedah, air yang dirawat selepas proses pengentalan-penjerapan yang melalui proses membran penurasan ultra menunjukkan fluks relatif yang lebih tinggi iaitu 16 dan 34 kali ganda tinggi bagi air kaolin dan air HA-kaolin. Dengan morfologi penjerap kulit manggis yang unik dan jumlah permukaan penjerapan yang besar turut mempercepatkan masa pemendapan. Secara amnya, rawatan air melalui sistem hibrid menunjukkan prestasi yang lebih baik berbanding rawatan individu dan ini dapat mengurangkan penggunaan bahan kimia terutamanya di dalam proses pengentalan, maka ini dapat mengurangkan kesan ke atas alam sekitar terhadap isu sisa di dalam loji rawatan air.

## ABSTRACT

Conventional water treatment processes are not capable to meet the drinking water quality. With the aid of chemical use in the treatment process, it deteriorates the water quality by producing by-product that could affect the treated water. In line with this, this research aims to reduce the chemical use in the coagulation process. Therefore, hybrid or combination process is evaluated in this research project. Kaolin solution and humic acid were used as impurities normally found in the surface water and the process was evaluated in terms of turbidity (kaolin) and  $UV_{254}$  (humic acid) removal. In this study, three different individual processes (coagulation, adsorption, and ultrafiltration membrane) and three hybrid processes (coagulation-ultrafiltration, coagulation-adsorption, and coagulation-adsorption-ultrafiltration) are performed. For the adsorption study, natural plant-based material namely, mangosteen pericarp (MP) was used in this research as an adsorbent. The process used in the research experiment were jar test and membrane separation process for coagulation/adsorption and ultrafiltration respectively. Synthetic kaolin and humic kaolin water was used as test water in this research experiment. The parameter analyzed in this research include turbidity,  $UV_{254}$  and physical characterization on membrane and MP adsorbent. In the individual treatment process, the coagulation process could effectively remove the impurity in the water at pH 10 value up to 97% turbidity removal and 93%  $UV_{254}$  removal efficiency. Meanwhile, the solely mangosteen pericarp adsorbent used in the adsorption process did not effectively remove both turbidity and  $UV_{254}$ . The ultrafiltration was able to remove the turbidity and  $UV_{254}$  up to 99% removal, however, the fouling would caused the flux permeate rate sharply reduced within first 6 minutes. The hybrid treatment process has shown higher removal efficiency and membrane relative flux. For the coagulation/ultrafiltration, its relative flux increased up to 12 and 30 times higher for the kaolin test water and HA-kaolin test water, respectively, as compared to the solely ultrafiltration process. The coagulation/adsorption at pH 8 shows better treated water quality. At pH 8 with 0.1 g of mangosteen pericarp adsorbent can harvest up to 99% of turbidity and  $UV_{254}$ . Further subsequent ultrafiltration after coagulation/adsorption shows even higher relative flux which were 16 and 34 times higher for the kaolin test water and HA-kaolin test water respectively. The unique mangosteen pericarp adsorbent morphology and its surface adsorption volume has boosted hastening the sedimentation time consumed. In general, the hybrid system was shown a better performance than the individual process and it could reduce the chemical usage especially in coagulation, hence reduce the environmental impact on the waste issue in the treatment plant.

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## LIST OF ABBREVIATIONS

BOD	Biochemical Oxygen Demand
DO	Dissolved Oxygen
NOM	Natural Organic Matters
SOCs	Synthetic Organic Chemicals
VOCs	Volatile Organic Chemicals
SEM	Scanning Electron Microscope
FTIR	Fourier Transform Infrared Spectroscopy
BET	Brunauer-Emmett-Teller
SDWA	Safe Drinking Water Act and Amendments
WHO	World Health Organization
SPAN	Suruhanjaya Perkhidmatan Air Negara
Alum	Aluminum Sulfate
DI	Deionised Water
RF	Relative Flux
NTU	Nephelometric Turbidity Units

## REFERENCES

- Ahmad, A.L, S Sumathi, and B. H Hameed. 2006. "Coagulation of Residue Oil and Suspended Solid in Palm Oil Mill Effluent by Chitosan, Alum and PAC." *Chemical Engineering Journal* 118 (1–2): 99–105. doi:10.1016/j.ccej.2006.02.001.
- Ahmad, U K, Z Ulang, Z Yusop, and T L Fong. 2002. "Fluorescence Technique for the Characterization of Natural Organic Matter in River Water." *Water Science and Technology : A Journal of the International Association on Water Pollution Research* 46 (9): 117–25.
- Ahmad Husaini Sulaiman. 2012. "Water Resources Agenda in Malaysia." In *Malaysia Water Resources Management Forum 2012*.
- Al-Mudhaf, Humood F., Faisal a. Alsharifi, and Abdel Sattar I Abu-Shady. 2009. "A Survey of Organic Contaminants in Household and Bottled Drinking Waters in Kuwait." *Science of the Total Environment* 407 (5). Elsevier B.V.: 1658–68. doi:10.1016/j.scitotenv.2008.10.057.
- Ali, Eman N, Sabreen R Alfarra, Mashita Mohd Yusoff, and Lutfor Rahman. 2015. "Environmentally Friendly Biosorbent from Moringa Oleifera Leaves for Water Treatment" 6 (3). doi:10.7763/IJESD.2015.V6.582.
- Alves, Vanessa N., and Nivia M M Coelho. 2013. "Selective Extraction and Preconcentration of Chromium Using Moringa Oleifera Husks as Biosorbent and Flame Atomic Absorption Spectrometry." *Microchemical Journal* 109. Elsevier B.V.: 16–22. doi:10.1016/j.microc.2012.05.030.
- Alwi, Habsah, Juferi Idris, Mohibah Musa, and Ku Halim Ku Hamid. 2013. "A Preliminary Study of Banana Stem Juice as a Plant-Based Coagulant for Treatment of Spent Coolant Wastewater." *Journal of Chemistry* 2013: 1–7. doi:10.1155/2013/165057.
- Amagloh, Francis Kweku, and Amos Benang. 2009. "Effectiveness of Moringa Oleifera Seed as Coagulant for Water Purification" 4 (February): 119–23.
- Amneera, W a, Nor Wahidatul Azura Zainon Najib, Siti Rawdhoh Mohd Yusof, and S Ragunathan. 2013. "Water Quality Index of Perlis River , Malaysia." *International Journal of Civil & Environmental Engineering* 13 (2): 1–6.
- Ang, Wei Lun, Abdul Wahab Mohammad, Nidal Hilal, and Choe Peng Leo. 2014. "A Review on the Applicability of Integrated/hybrid Membrane Processes in Water Treatment and Desalination Plants." *Desalination* 363. Elsevier B.V.: 2–18.



doi:10.1016/j.desal.2014.03.008.

Araújo, Cleide S T, Vanessa N. Alves, Hélen C. Rezende, Ione L S Almeida, Rosana M N De Assunção, César R T Tarley, Mariana G. Segatelli, and Nivia M Melo Coelho. 2010. "Characterization and Use of Moringa Oleifera Seeds as Biosorbent for Removing Metal Ions from Aqueous Effluents." *Water Science and Technology* 62 (9): 2198–2203. doi:10.2166/wst.2010.419.

Araya-Farias, Monica, and Laurent Bazinet. 2006. "Effect of Calcium and Carbonate Concentrations on Anionic Membrane Fouling during Electrodialysis." *Journal of Colloid and Interface Science* 296 (1): 242–47. doi:10.1016/j.jcis.2005.08.040.

Asatekin, Ayse, Seoktae Kang, Menachem Elimelech, and Anne M Mayes. 2007. "Anti-Fouling Ultrafiltration Membranes Containing Polyacrylonitrile- Graft -Poly ( Ethylene Oxide ) Comb Copolymer Additives" 298: 136–46. doi:10.1016/j.memsci.2007.04.011.

Asati, A R. 2013. "Treatment Of Waste Water From Parboiled Rice Mill Unit By Coagulation / Flocculation." *Life Sciences Biotechnology and Pharma Research*.

Ashery, Ahamed Fadel, Kamal Radwan, and Mohamed I Gar Al-alm. 2010. "The Effect of pH Control on Turbidity and NOM Removal in Conventional Water Treatment." *International Water Technology Journal* 1 (2): 1–16.

Averett, Rc, Ja Leenheer, Dm McKnight, and Ka Thorn. 1994. "Humic Substances in the Suwannee River, Georgia: Interactions, Properties and Proposed Structures." *US Government Printing Office. Washington, DC, ...*, 224. <http://scholar.google.com/scholar?hl=en&btnG=Search&q=intitle:Humic+Substances+in+the+Suwannee+River,+Georgia:+Interactions,+Properties,+and+Proposed+Structures#0>.

Bacchin, P., P. Aimar, and R. W. Field. 2006. "Critical and Sustainable Fluxes: Theory, Experiments and Applications." *Journal of Membrane Science* 281 (1–2): 42–69. doi:10.1016/j.memsci.2006.04.014.

Barghi, Seyed Hamed, Theodore T. Tsotsis, and Muhammad Sahimi. 2014. "Chemisorption, Physisorption and Hysteresis during Hydrogen Storage in Carbon Nanotubes." *International Journal of Hydrogen Energy* 39 (3). Elsevier Ltd: 1390–97. doi:10.1016/j.ijhydene.2013.10.163.

Belfer, S., R. Fainchtain, Y. Purinson, and O. Kedem. 2000. "Surface Characterization by FTIR-ATR Spectroscopy of Polyethersulfone Membranes-Unmodified, Modified and Protein Fouled." *Journal of Membrane Science* 172 (1–2): 113–24. doi:10.1016/S0376-7388(00)00316-1.

- Bose, Purnendu, and David a. Reckhow. 2007. "The Effect of Ozonation on Natural Organic Matter Removal by Alum Coagulation." *Water Research* 41 (7): 1516–24. doi:10.1016/j.watres.2006.12.027.
- Cao, Bai Chuan, Bao Yu Gao, Chun Hua Xu, Ying Fu, and Xin Liu. 2010. "Effects of pH on Coagulation Behavior and Floc Properties in Yellow River Water Treatment Using Ferric Based Coagulants." *Chinese Science Bulletin* 55 (14): 1382–87. doi:10.1007/s11434-010-0087-5.
- Cao, Yuhe, Keliang Wang, Xiaoming Wang, Zhengrong Gu, William Gibbons, and Han Vu. 2015. "Butanol Vapor Adsorption Behavior on Active Carbons and Zeolite Crystal." *Applied Surface Science* 349. Elsevier B.V.: 1–7. doi:10.1016/j.apsusc.2015.05.005.
- Changmai, Murchana, Poulami Banerjee, Karan Nahar, and Mihir K. Purkait. 2018. "A Novel Adsorbent from Carrot, Tomato and Polyethylene Terephthalate Waste as a Potential Adsorbent for Co (II) from Aqueous Solution: Kinetic and Equilibrium Studies." *Journal of Environmental Chemical Engineering* 6 (1). Elsevier: 246–57. doi:10.1016/j.jece.2017.12.009.
- Chen, Xiangrong, Jianquan Luo, Benkun Qi, Weifeng Cao, and Yinhua Wan. 2015a. "Journal of Water Process Engineering NOM Fouling Behavior during Ultrafiltration : Effect of Membrane Hydrophilicity" 7. Elsevier Ltd: 1–10.
- . 2015b. "NOM Fouling Behavior during Ultrafiltration: Effect of Membrane Hydrophilicity." *Journal of Water Process Engineering* 7. Elsevier Ltd: 1–10. doi:10.1016/j.jwpe.2015.04.009.
- Chen, Yandan, Biao Huang, Mingjie Huang, and Biqiong Cai. 2011. "On the Preparation and Characterization of Activated Carbon from Mangosteen Shell." *Journal of the Taiwan Institute of Chemical Engineers* 42 (5). Taiwan Institute of Chemical Engineers: 837–42. doi:10.1016/j.jtice.2011.01.007.
- Cheng, Wei, Seyed A. Dastgheib, and Tanju Karanfil. 2005. "Adsorption of Dissolved Natural Organic Matter by Modified Activated Carbons." *Water Research* 39 (11): 2281–90. doi:10.1016/j.watres.2005.01.031.
- Childress, Amy E., and Shivaji S. Deshmukh. 1998. "89107.pdf." *Desalination* 118: 167–74.
- Choy, Sook Yan, Krishna Murthy Nagendra Prasad, Ta Yeong Wu, Mavinakere Eshwaraiah Raghunandan, and Ramakrishnan Nagasundara Ramanan. 2014. "Utilization of Plant-Based Natural Coagulants as Future Alternatives towards Sustainable Water Clarification." *Journal of Environmental Sciences* 26 (11).

Elsevier B.V.: 2178–89. doi:10.1016/j.jes.2014.09.024.

Cleophas, Nony, Feona Isidore, Lee Ka Han, and Kawi Bidin. 2013. “Water Quality Status of Liwagu River.” *Journal of Tropical Biology and Conservation* 10 (April): 67–73.

Cornel, P., A. Meda, and S. Bieker. 2011. “4.12 – Wastewater as a Source of Energy, Nutrients, and Service Water.” In *Treatise on Water Science*, 337–75. doi:10.1016/B978-0-444-53199-5.00092-0.

Crist, John T, Yuniati Zevi, John F Mccarthy, James A Throop, and Tammo S Steenhuis. 2005. “Transport and Retention Mechanisms of Colloids in Partially Saturated Porous Media” 4: 184–95.

Cui, Jihong, Wen Hu, Zhanjun Cai, Yingxue Liu, Siyuan Li, Wucheng Tao, and Hui Xiang. 2010. “New Medicinal Properties of Mangostins: Analgesic Activity and Pharmacological Characterization of Active Ingredients from the Fruit Hull of *Garcinia Mangostana* L.” *Pharmacology Biochemistry and Behavior* 95 (2). Elsevier Inc.: 166–72. doi:10.1016/j.pbb.2009.12.021.

Daoud, Hadj, Oumessaâd Benturki, Nouredine Bouras, Mouloud Attou, André Donnot, André Merlin, Fatima Addoun, and Michael D Holtz. 2015. “The Use of an Agricultural Waste Material from *Ziziphus Jujuba* as a Novel Adsorbent for Humic Acid Removal from Aqueous Solutions.” *Journal of Molecular Liquids* 211. Elsevier B.V.: 1039–46. doi:10.1016/j.molliq.2015.08.028.

De Angelis, Gabriele, Laura Medeghini, Aida Maria Conte, and Silvano Mignardi. 2017. “Recycling of Eggshell Waste into Low-Cost Adsorbent for Ni Removal from Wastewater.” *Journal of Cleaner Production* 164. Elsevier Ltd: 1497–1506. doi:10.1016/j.jclepro.2017.07.085.

Department of Irrigation and Drainage. 2009. “STUDY ON THE RIVER WATER QUALITY TRENDS STUDY AND INDEXES IN PENINSULAR MALAYSIA.” Malaysia.

Detector, Monochromator. 2010. “Copyright © 1999 – 2010 by Mark Brandt, Ph.D. 12,” 12–19. doi:10.1503/cmaj.109-4864.

Dong, Hongyu, Baoyu Gao, Qinyan Yue, Yan Wang, and Qian Li. 2015. “Effect of pH on Floc Properties and Membrane Fouling in Coagulation – Ultrafiltration Process with Ferric Chloride and Polyferric Chloride.” *Chemosphere* 130. Elsevier Ltd: 90–97. doi:10.1016/j.chemosphere.2015.03.049.

- Duan, Jinming, Jianhui Wang, Nigel Graham, and Francis Wilson. 2002. "Coagulation of Humic Acid by Aluminium Sulphate in Saline Water Conditions." *Desalination* 150 (1): 1–14. doi:10.1016/S0011-9164(02)00925-6.
- "Electronegativities of the Elements ( Pauling Scale )." 2012. *Princeton School*.  
[http://www.princeton-schools.net/site/handlers/filedownload.ashx?moduleinstanceid=716&dataid=477&FileName=electronegativity table.pdf](http://www.princeton-schools.net/site/handlers/filedownload.ashx?moduleinstanceid=716&dataid=477&FileName=electronegativity%20table.pdf).
- Engelhardt, Terry L. 2014. "Coagulation, Flocculation and Clarification of Drinking Water." *Hach*, no. LIT2141: 1–57.
- EPA. 2003. "Wastewater Technology Fact Sheet - Disinfection for Small Systems."
- Exall, Kirsten. 2005. "Coagulation and Flocculation," July.  
 doi:10.1002/047147844X.pc1505.
- Feng, Chung Liao. 2013. "Measurements of Charge in Heated Water." *Journal of Electrostatics* 71 (3). Elsevier Ltd: 551–53. doi:10.1016/j.elstat.2012.11.016.
- Feng, Lijuan, Wenyu Wang, Ruiqi Feng, Shuang Zhao, Hongyu Dong, Shenglei Sun, Baoyu Gao, and Qinyan Yue. 2015a. "Coagulation Performance and Membrane Fouling of Different Aluminum Species during Coagulation/ultrafiltration Combined Process." *Chemical Engineering Journal* 262. Elsevier B.V.: 1161–67. doi:10.1016/j.cej.2014.10.078.
- . 2015b. "Coagulation Performance and Membrane Fouling of Different Aluminum Species during Coagulation/ultrafiltration Combined Process." *Chemical Engineering Journal* 262 (February). Elsevier B.V.: 1161–67. doi:10.1016/j.cej.2014.10.078.
- Garfí, Marianna, Erasmo Cadena, David Sanchez-Ramos, and Ivet Ferrer. 2016. "Life Cycle Assessment of Drinking Water: Comparing Conventional Water Treatment, Reverse Osmosis and Mineral Water in Glass and Plastic Bottles." *Journal of Cleaner Production* 137. Elsevier Ltd: 997–1003. doi:10.1016/j.jclepro.2016.07.218.
- Geng, Yi. 2005. "Application of Flocc Analysis for Coagulation Optimization at the Split Lake Water Treatment Plant."
- Gouttefangeas, F, J Le Lannic, and P Rabiller. 2012. "Coupling of SEM-EDX and FTIR-ATR to ( Quantitatively ) Investigate Organic Fouling on Porous Organic Composite Membranes," 1066–76.

- Guo, H., Y. Wyart, J. Perot, F. Nauleau, and P. Moulin. 2010. "Low-Pressure Membrane Integrity Tests for Drinking Water Treatment: A Review." *Water Research* 44 (1). Elsevier Ltd: 41–57. doi:10.1016/j.watres.2009.09.032.
- Hesami, Farid, Bijan Bina, and Afshin Ebrahimi. 2014. "The Effectiveness of Chitosan as Coagulant Aid in Turbidity Removal from Water." *International Journal of Environmental Health Engineering* 3 (1): 8. doi:10.4103/2277-9183.131814.
- Hsiung, Chia En, Hsing Lung Lien, Alexander Edward Galliano, Chia Shen Yeh, and Yang Hsin Shih. 2016. "Effects of Water Chemistry on the Destabilization and Sedimentation of Commercial TiO<sub>2</sub> Nanoparticles: Role of Double-Layer Compression and Charge Neutralization." *Chemosphere* 151: 145–51. doi:10.1016/j.chemosphere.2016.02.046.
- Hu, Jin-Jia, Yi-Hsuan Hsieh, and Jeng-Shiung Jan. 2015. "Polyelectrolyte Complex-Silica Hybrid Colloidal Particles Decorated with Different Polyelectrolytes." *Journal of Colloid and Interface Science* 438 (January). Elsevier Inc.: 94–101. doi:10.1016/j.jcis.2014.09.063.
- Huang, Y.K., S.Y. Ang, K.M. Lee, and T.S. Lee. 2015. "Quality of Water Resources in Malaysia." *Research and Practices in Water Quality*, 65–94. doi:10.5772/58969.
- Hwang, Kuo-jen, Chien-yao Liao, and Kuo-lun Tung. 2008. "Effect of Membrane Pore Size on the Particle Fouling in Membrane Filtration." *DES* 234 (1–3). Elsevier B.V.: 16–23. doi:10.1016/j.desal.2007.09.065.
- Idris, Juferi, Ayub Md Som, Mohibah Musa, Ku Halim Ku Hamid, Rafidah Husen, and Miradatul Najwa Muhd Rodhi. 2013. "Dragon Fruit Foliage Plant-Based Coagulant for Treatment of Concentrated Latex Effluent: Comparison of Treatment with Ferric Sulfate." *Journal of Chemistry* 2013: 1–7. doi:10.1155/2013/230860.
- Katsoufidou, K., S. G. Yiantsios, and a. J. Karabelas. 2005. "A Study of Ultrafiltration Membrane Fouling by Humic Acids and Flux Recovery by Backwashing: Experiments and Modeling." *Journal of Membrane Science* 266 (1–2): 40–50. doi:10.1016/j.memsci.2005.05.009.
- Kim, Hyun-Chul. 2015. "High-Rate MIEX Filtration for Simultaneous Removal of Phosphorus and Membrane Foulants from Secondary Effluent." *Water Research* 69. Elsevier Ltd: 40–50. doi:10.1016/j.watres.2014.11.012.
- Kodape, Shyam, and A Acclimatization Stage. 2014. "Study on Performance of Membrane Bioreactor ( MBR ) System at Various Temperatures for Wastewater Treatment" 1 (7): 108–16.

- Kourde-hanafi, Yamina, Patrick Loulergue, Anthony Szymczyk, and Bart Van Der Bruggen. 2017. "Influence of PVP Content on Degradation of PES / PVP Membranes : Insights from Characterization of Membranes with Controlled Composition." *Journal of Membrane Science*. Elsevier B.V. doi:10.1016/j.memsci.2017.03.050.
- Lai, Chun-Hsi, Yung-Chen Chou, and Hsuan-Hsien Yeh. 2015. "Assessing the Interaction Effects of Coagulation Pretreatment and Membrane Material on UF Fouling Control Using HPSEC Combined with Peak-Fitting." *Journal of Membrane Science* 474. Elsevier: 207–14. doi:10.1016/j.memsci.2014.09.052.
- Law, Bee Bee. 2005. "The Usage of Domestic Water Filtration Systems in Malaysia." University of Southern Queensland.
- Lee, Jieun, Sanghyun Jeong, Yun Ye, Vicki Chen, Saravanamuthu Vigneswaran, TorOve Leiknes, and Zongwen Liu. 2016. "Protein Fouling in Carbon Nanotubes Enhanced Ultrafiltration Membrane: Fouling Mechanism as a Function of pH and Ionic Strength." *Separation and Purification Technology*. Elsevier B.V. doi:http://dx.doi.org/10.1016/j.seppur.2016.10.061.
- Leverenz, H L, and T Asano. 2011. "4.03 - Wastewater Reclamation and Reuse System." *Treatise on Water Science*, 63–71. doi:http://dx.doi.org/10.1016/B978-0-444-53199-5.00076-2.
- Li, Chengliang, Anne E. Berns, Andreas Schäffer, Jean Marie Séquaris, Harry Vereecken, Rong Ji, and Erwin Klumpp. 2011. "Effect of Structural Composition of Humic Acids on the Sorption of a Branched Nonylphenol Isomer." *Chemosphere* 84 (4). Elsevier Ltd: 409–14. doi:10.1016/j.chemosphere.2011.03.057.
- Li, Kai, Tinglin Huang, Fangshu Qu, Xing Du, An Ding, Guibai Li, and Heng Liang. 2016. "Performance of Adsorption Pretreatment in Mitigating Humic Acid Fouling of Ultrafiltration Membrane under Environmentally Relevant Ionic Conditions." *Desalination* 377. Elsevier B.V.: 91–98. doi:10.1016/j.desal.2015.09.016.
- Li, Meng Fei. 2014. "Effects of Natural Organic Matter on Contaminant Removal by Superfine Powdered Activated Carbon Coupled with Microfiltration Membrane," no. May.
- Lim, Victor. K. E. 2007. "Infectious Diarrhoea." *Medicine (United Kingdom)* 43 (5): 253–58. doi:10.1016/j.mpmed.2015.02.005.
- Lin, Jr Lin, Chihpin Huang, Ching Ju M Chin, and Jill R. Pan. 2008. "Coagulation Dynamics of Fractal Flocs Induced by Enmeshment and Electrostatic Patch

- Mechanisms.” *Water Research* 42 (17). Elsevier Ltd: 4457–66.  
doi:10.1016/j.watres.2008.07.043.
- Lin, Tao, ZiJian Lu, and Wei Chen. 2015. “Interaction Mechanisms of Humic Acid Combined with Calcium Ions on Membrane Fouling at Different Conditions in an Ultrafiltration System.” *Desalination* 357 (February). Elsevier B.V.: 26–35.  
doi:10.1016/j.desal.2014.11.007.
- Liu, G., M. C. Lut, J. Q J C Verberk, and J. C. Van Dijk. 2013. “A Comparison of Additional Treatment Processes to Limit Particle Accumulation and Microbial Growth during Drinking Water Distribution.” *Water Research* 47 (8). Elsevier Ltd: 2719–28. doi:10.1016/j.watres.2013.02.035.
- Liu, Jing, Jan D. Miller, Xihui Yin, Vishal Gupta, and Xuming Wang. 2014. “Influence of Ionic Strength on the Surface Charge and Interaction of Layered Silicate Particles.” *Journal of Colloid and Interface Science* 432. Elsevier Inc.: 270–77.  
doi:10.1016/j.jcis.2014.06.028.
- Liu, Sean X., and Larry a. Glasgow. 1995. “Conformation of Adsorbed Macromolecules and Solid-Liquid Separation in Coagulation Processes.” *Separations Technology* 5 (3): 139–45. doi:10.1016/0956-9618(94)00119-D.
- Lu, Xiaoqiao, Zuliang Chen, and Xinhao Yang. 1999. “Spectroscopic Study of Aluminium Speciation in Removing Humic Substances by Al Coagulation.” *Water Research* 33 (15): 3271–80. doi:10.1016/S0043-1354(99)00047-0.
- Ma, Zhun, Jian-jun Qin, Cui-xian Liou, Lifeng Zhang, and Suresh Valiyaveetil. 2012. “Effects of Coagulation pH and Mixing Conditions on Characteristics of Flocs in Surface Water Treatment.”
- Malherbe, François, Claude Forano, and Jean-Pierre Besse. 1997. “Use of Organic Media to Modify the Surface and Porosity Properties of Hydrotalcite-like Compounds.” *Microporous Materials* 10 (1–3): 67–84. doi:10.1016/S0927-6513(96)00123-X.
- Malkoc, Emine, and Yasar Nuhoglu. 2005. “Investigations of nickel(II) Removal from Aqueous Solutions Using Tea Factory Waste.” *Journal of Hazardous Materials* 127 (1–3): 120–28. doi:10.1016/j.jhazmat.2005.06.030.
- Mamun, Abdullah a, and Zaki Zainudin. 2013. “Sustainable River Water Quality Management in Malaysia.” *IJUM Engineering Journal* 14 (1): 29–42.
- Mänttäre, Mika, Liisa Puro, Jutta Nuortila-Jokinen, and Marianne Nyström. 2000.

- “Fouling Effects of Polysaccharides and Humic Acid in Nanofiltration.” *Journal of Membrane Science* 165 (1): 1–17. doi:10.1016/S0376-7388(99)00215-X.
- Manurakchinakorn, S., Y. Chainarong, and C. Sawatpadungkit. 2016. “Quality of Mangosteen Juice Colored with Mangosteen Pericarp.” *International Food Research Journal* 23 (3): 1033–39.
- Meghzili, Bachir, Brakchi Souad, and Azzouz Abdelkrim. 2016. “Risk of Residual Aluminum in Treated Waters with Aluminum Sulphate.” *Advances in Research* 6 (5): 1–8. doi:10.9734/AIR/2016/24059.
- Mehrpavar, Alisa, Ahmad Rahimpour, and Mohsen Jahanshahi. 2014. “Modified Ultrafiltration Membranes for Humic Acid Removal.” *Journal of the Taiwan Institute of Chemical Engineers* 45 (1). Taiwan Institute of Chemical Engineers: 275–82. doi:10.1016/j.jtice.2013.06.003.
- Microbac. 2011. FTIR and SEM / EDS in Identification of Unknown Materials FTIR and SEM / EDS in Identification of Unknown Materials, issued 2011.
- Mimoso, Joao, Wouter Pronk, Eberhard Morgenroth, and Frederik Hammes. 2015. “Bacterial Growth in Batch-Operated Membrane Filtration Systems for Drinking Water Treatment.” *Separation and Purification Technology* 156. Elsevier B.V.: 165–74. doi:10.1016/j.seppur.2015.09.070.
- Mittal, Arun. 2011. “Biological Wastewater Treatment.” *Biological Wastewater Treatment*. <http://www.loc.gov/catdir/enhancements/fy0647/98037263-d.html>.
- Mohd Khalid, M. Z, and M Masri. 1996. “Mangosteen Breeding,” no. November: 213–16.
- Muisa, Norah, Zvikomborero Hoko, and Portia Chifamba. 2011. “Impacts of Alum Residues from Morton Jaffray Water Works on Water Quality and Fish, Harare, Zimbabwe.” *Physics and Chemistry of the Earth, Parts A/B/C* 36 (14–15). Elsevier Ltd: 853–64. doi:10.1016/j.pce.2011.07.047.
- Mukherjee, Sumona, Agamuthu Pariatamby, Jaya Narayan Sahu, and Bhaskar Sen Gupta. 2013. “Clarification of Rubber Mill Wastewater by a Plant Based Biopolymer - Comparison with Common Inorganic Coagulants.” *Journal of Chemical Technology and Biotechnology* 88 (10): 1864–73. doi:10.1002/jctb.4041.
- Muthuraman, G., and S. Sasikala. 2014. “Removal of Turbidity from Drinking Water Using Natural Coagulants.” *Journal of Industrial and Engineering Chemistry* 20 (4). The Korean Society of Industrial and Engineering Chemistry: 1727–31.



doi:10.1016/j.jiec.2013.08.023.

- National Water Quality Inventory. 1996. "4.0 Environmental Assessment." In *Environmental Assessment*, 1–49.  
[http://www.google.com.my/url?sa=t&rct=j&q=&esrc=s&source=web&cd=1&ved=0CBwQFjAA&url=http://water.epa.gov/scitech/wastetech/guide/stormwater/upload/2006\\_10\\_31\\_guide\\_stormwater\\_usw\\_b.pdf&ei=j2dkVLuxEdGIuATJ9IKgCw&usg=AFQjCNFi5N3Nx8TNiIXxirUHZd0DoLt1ZQ&sig2=9Y0VkJKXYSO-W1t74sr73w&bvm=bv.79189006,d.c2E](http://www.google.com.my/url?sa=t&rct=j&q=&esrc=s&source=web&cd=1&ved=0CBwQFjAA&url=http://water.epa.gov/scitech/wastetech/guide/stormwater/upload/2006_10_31_guide_stormwater_usw_b.pdf&ei=j2dkVLuxEdGIuATJ9IKgCw&usg=AFQjCNFi5N3Nx8TNiIXxirUHZd0DoLt1ZQ&sig2=9Y0VkJKXYSO-W1t74sr73w&bvm=bv.79189006,d.c2E).
- Naubi, Irena, Noorul Hassan Zardari, Sharif Moniruzzaman Shirazi, Nurul Farahen, Binti Ibrahim, and Lavania Baloo. 2016. "Effectiveness of Water Quality Index for Monitoring Malaysian River Water Quality." *Pol. J. Environ. Stud.* 25 (1): 231–39.  
doi:10.15244/pjoes/60109.
- Ndlangamandla, Nqobile G., Alex T. Kuvarega, Titus A.M. Msagati, Bhekie B. Mamba, and Thabo T.I. Nkambule. 2018. "A Novel Photodegradation Approach for the Efficient Removal of Natural Organic Matter (NOM) from Water." *Physics and Chemistry of the Earth*, no. April 2017. Elsevier: 1–10.  
doi:10.1016/j.pce.2018.05.011.
- Ness, Gd. 2008. "World Population Growth, in The Global Environment: Science, Technology and Management." *The Global Environment: Science, Technology and ...* <http://onlinelibrary.wiley.com/doi/10.1002/9783527619658.ch38/summary>.
- Nilsen, Tom O, Lars O E Ebbesson, Ole G Kverneland, Frode Kroglund, Bengt Finstad, and Sigurd O Stefansson. 2010. "Effects of Acidic Water and Aluminum Exposure on Gill Na(+), K(+)-ATPase Alpha-Subunit Isoforms, Enzyme Activity, Physiology and Return Rates in Atlantic Salmon (*Salmo Salar* L.)." *Aquatic Toxicology (Amsterdam, Netherlands)* 97 (3). Elsevier B.V.: 250–59.  
doi:10.1016/j.aquatox.2009.12.001.
- Oladoja, N.a., Y.B Alliu, a.E Ofomaja, and I.E. Unuabonah. 2011. "Synchronous Attenuation of Metal Ions and Colour in Aqua Stream Using Tannin–alum Synergy." *Desalination* 271 (1–3). Elsevier B.V.: 34–40.  
doi:10.1016/j.desal.2010.12.008.
- Osman, Mohamad Bin, and Abd Rahman Milan. 2006. *Mangosteen: Garcinia Mangostana L. Postharvest Biology and Technology of Tropical and Subtropical Fruits: Volume 4 : Mangosteen to White Sapote*. Woodhead Publishing Limited.  
doi:10.1533/9780857092618.1.
- Palapol, Y., S. Ketsa, D. Stevenson, J. M. Cooney, A. C. Allan, and I. B. Ferguson. 2009. "Colour Development and Quality of Mangosteen (*Garcinia Mangostana* L.)

Fruit during Ripening and after Harvest.” *Postharvest Biology and Technology* 51 (3): 349–53. doi:10.1016/j.postharvbio.2008.08.003.

Patrick, Di Martino, Houari Ahmed, and Laboratoire Errmece Ea. 2010. “Assessment of UV Pre-Treatment to Reduce Fouling of NF Membranes.”

Pedraza-chaverri, José, Noemí Cárdenas-rodríguez, Marisol Orozco-ibarra, and Jazmin M Pérez-rojas. 2008. “Medicinal Properties of Mangosteen ( *Garcinia Mangostana* ).” *Food and Chemical Toxicology* 46 (10). Elsevier Ltd: 3227–39. doi:10.1016/j.fct.2008.07.024.

Pedraza-Chaverri, José, Noemí Cárdenas-Rodríguez, Marisol Orozco-Ibarra, and Jazmin M. Pérez-Rojas. 2008. “Medicinal Properties of Mangosteen (*Garcinia Mangostana*).” *Food and Chemical Toxicology* 46 (10). Elsevier Ltd: 3227–39. doi:10.1016/j.fct.2008.07.024.

Perkins, Tracy L., Karen Perrow, Paulina Rajko-Nenow, Colin F. Jago, Davey L. Jones, Shelagh K. Malham, and James E. McDonald. 2016. “Decay Rates of Faecal Indicator Bacteria from Sewage and Ovine Faeces in Brackish and Freshwater Microcosms with Contrasting Suspended Particulate Matter Concentrations.” *Science of The Total Environment*. Elsevier B.V. doi:10.1016/j.scitotenv.2016.03.076.

Prime Minister’s Department. 2010. *Tenth Malaysia Plan 2011-2015*.

Ripszam, M., C.M.J. Gallampois, Å. Berglund, H. Larsson, a. Andersson, M. Tysklind, and P. Haglund. 2015. “Effects of Predicted Climatic Changes on Distribution of Organic Contaminants in Brackish Water Mesocosms.” *Science of The Total Environment* 517. Elsevier B.V.: 10–21. doi:10.1016/j.scitotenv.2015.02.051.

Robinson, T., B. Chandran, and P. Nigam. 2002. “Removal of Dyes from a Synthetic Textile Dye Effluent by Biosorption on Apple Pomace and Wheat Straw.” *Water Research* 36 (11): 2824–30. doi:10.1016/S0043-1354(01)00521-8.

Rodrigues, Alexandrina, António Brito, Peter Janknecht, Maria Fernanda Proença, and Regina Nogueira. 2009. “Quantification of Humic Acids in Surface Water: Effects of Divalent Cations, pH, and Filtration.” *Journal of Environmental Monitoring : JEM* 11 (2): 377–82. doi:10.1039/b811942b.

Roheim, John. 2012. “Low Cost Automated On Site System For Growing And Dispensing Vegetative Bacteria.”

Santos, Philipe, Ana C Aguiar, Gerardo F Barbero, Camila A Rezende, and Julian

- Martínez. 2015. "Ultrasonics Sonochemistry Supercritical Carbon Dioxide Extraction of Capsaicinoids from Malagueta Pepper ( *Capsicum Frutescens* L .) Assisted by Ultrasound." *Ultrasonics - Sonochemistry* 22. Elsevier B.V.: 78–88. doi:10.1016/j.ultsonch.2014.05.001.
- Schulze-makuch, Dirk. 2005. "Advection, Dispersion, Sorption, Degradation, Attenuation." Vol. II. <https://www.eolss.net/Sample-Chapters/C07/E2-09-03-05.pdf>.
- Shen, Meng, Sinan Keten, and Richard M Lueptow. 2016. "Rejection Mechanisms for Contaminants in Polymeric Reverse Osmosis Membranes," 36–47.
- Sheng, Chenguang, A G Agwu Nnanna, Yanghe Liu, and John D Vargo. 2016. "Removal of Trace Pharmaceuticals from Water Using Coagulation and Powdered Activated Carbon as Pretreatment to Ultrafiltration Membrane System." *Science of The Total Environment* 550. Elsevier B.V.: 1075–83. doi:<http://dx.doi.org/10.1016/j.scitotenv.2016.01.179>.
- Shi, Meixia, Galina Printsypar, Oleg Iliev, Victor M. Calo, Gary L. Amy, and Suzana P. Nunes. 2015. "Water Flow Prediction for Membranes Using 3D Simulations with Detailed Morphology." *Journal of Membrane Science* 487. Elsevier: 19–31. doi:10.1016/j.memsci.2015.03.036.
- Shockravi, Abbas, Vahid Vatanpour, Zahra Najjar, Saedeh Bahadori, and Ali Javadi. 2017. "A New High Performance Polyamide as an Effective Additive for Modification of Antifouling Properties and Morphology of Asymmetric PES Blend Ultrafiltration Membranes." *Microporous and Mesoporous Materials* 246. Elsevier Ltd: 24–36. doi:10.1016/j.micromeso.2017.03.013.
- Simate, Geoffrey S, Sunny E Iyuke, Sehliselo Ndlovu, Mike Heydenrych, and Lubinda F Walubita. 2012. "Human Health Effects of Residual Carbon Nanotubes and Traditional Water Treatment Chemicals in Drinking Water." *Environment International* 39 (1). Elsevier Ltd: 38–49. doi:10.1016/j.envint.2011.09.006.
- Srisurichan, Surapit, Ratana Jiratananon, and a. G. Fane. 2005. "Humic Acid Fouling in the Membrane Distillation Process." *Desalination* 174 (1): 63–72. doi:10.1016/j.desal.2004.09.003.
- Subramonian, Wennie, Ta Yeong Wu, and Siang-Piao Chai. 2014. "A Comprehensive Study on Coagulant Performance and Floc Characterization of Natural Cassia Obtusifolia Seed Gum in Treatment of Raw Pulp and Paper Mill Effluent." *Industrial Crops and Products* 61. Elsevier B.V.: 317–24. doi:10.1016/j.indcrop.2014.06.055.

- Sun, Shenglei, Zhonglian Yang, Xin Huang, Fan Bu, Defang Ma, Hongyu Dong, Baoyu Gao, Qinyan Yue, Yan Wang, and Qian Li. 2015. "Coagulation Performance and Membrane Fouling of Polyferric Chloride/epichlorohydrin–dimethylamine in Coagulation/ultrafiltration Combined Process." *Desalination* 357. Elsevier B.V.: 163–70. doi:10.1016/j.desal.2014.11.031.
- Suthammarak, Wichit, Pornpayom Numpraphrut, Ratiya Charoensakdi, Neelobol Neungton, Vachara Tunrungruangtavee, Nattapon Jaisupa, Suwit Charoensak, Primchanien Moongkarndi, and Weerasak Muangpaisan. 2016. "Antioxidant-Enhancing Property of the Polar Fraction of Mangosteen Pericarp Extract and Evaluation of Its Safety in Humans." *Oxidative Medicine and Cellular Longevity* 2016. doi:10.1155/2016/1293036.
- Tang, Hongxiao, Feng Xiao, and Dongsheng Wang. 2015. "Speciation, Stability, and Coagulation Mechanisms of Hydroxyl Aluminum Clusters Formed by PACl and Alum: A Critical Review." *Advances in Colloid and Interface Science* 226. Elsevier B.V.: 78–85. doi:10.1016/j.cis.2015.09.002.
- Tatsuzawa, Fumi, Shun Ito, Motoki Sato, Hiroki Muraoka, Kazuhisa Kato, Yoshihito Takahata, and Satoshi Ogawa. 2013. "A Tetra-Acylated Cyanidin 3-Sophoroside-5-Glucoside from the Purple-Violet Flowers of *Moricandia Arvensis* (L.) DC. (Brassicaceae)." *Phytochemistry Letters* 6 (2). Phytochemical Society of Europe: 170–73. doi:10.1016/j.phytol.2012.12.007.
- Tehrani-Bagha, a.R., H. Nikkar, N.M. Mahmoodi, M. Markazi, and F.M. Menger. 2011. "The Sorption of Cationic Dyes onto Kaolin: Kinetic, Isotherm and Thermodynamic Studies." *Desalination* 266 (1–3). Elsevier B.V.: 274–80. doi:10.1016/j.desal.2010.08.036.
- Teow, Y.H., B.S. Ooi, and A.L. Ahmad. 2016. "Fouling Behaviours of PVDF-TiO<sub>2</sub> Mixed-Matrix Membrane Applied to Humic Acid Treatment." *Journal of Water Process Engineering* 15. Elsevier Ltd: 89–98. doi:10.1016/j.jwpe.2016.03.005.
- Thermo Nicolet Corporation. 2001. Introduction to Fourier Transform Infrared Spectrometry, issued 2001.
- Thevannan, Ayyasamy, Rubeena Mungroo, and Hui Catherine Niu. 2010. "Biosorption of Nickel with Barley Straw." *Bioresource Technology* 101 (6). Elsevier Ltd: 1776–80. doi:10.1016/j.biortech.2009.10.035.
- Tian, Qing-hua, Yun-tao Xin, Li Yang, Xue-hai Wang, and Xue-yi Guo. 2016. "Theoretical Simulation and Experimental Study of Hydrolysis Separation of SbCl<sub>3</sub> in Complexation – Precipitation System." *Transactions of Nonferrous Metals Society of China* 26 (10). The Nonferrous Metals Society of China: 2746–53.

doi:10.1016/S1003-6326(16)64370-4.

- UNESCO. 2017. *Wastewater - The Untapped Resource*.  
<http://unesdoc.unesco.org/images/0024/002471/247153e.pdf>.
- Veshchunov, M.S. 2010. "A New Approach to the Brownian Coagulation Theory." *Journal of Aerosol Science* 41 (10). Elsevier: 895–910.  
doi:10.1016/j.jaerosci.2010.07.001.
- Villet, Michael C., and George R. Gavalas. 2007. "Measurement of Concentration-Dependent Gas Diffusion Coefficients in Membranes from a Pseudo-Steady State Permeation Run." *Journal of Membrane Science* 297 (1–2): 199–205.  
doi:10.1016/j.memsci.2007.03.045.
- von Wandruszka, Ray. 2000. "Humic Acids: Their Detergent Qualities and Potential Uses in Pollution Remediation." *Geochemical Transactions* 1 (2): 10.  
doi:10.1039/b001869o.
- Wahab, Mohamed Ali, Hatem Boubakri, Salah Jellali, and Naceur Jedidi. 2012. "Characterization of Ammonium Retention Processes onto Cactus Leaves Fibers Using FTIR, EDX and SEM Analysis." *Journal of Hazardous Materials* 241–242. Elsevier B.V.: 101–9. doi:10.1016/j.jhazmat.2012.09.018.
- Wang, Chao, Ji Dai, Chii Shang, and Guanghao Chen. 2013. "Removal of Aqueous Fullerene nC60 from Wastewater by Alum-Enhanced Primary Treatment." *Separation and Purification Technology* 116 (September). Elsevier B.V.: 61–66.  
doi:10.1016/j.seppur.2013.05.035.
- Wang, Jing J, Qing H Shi, Wei Zhang, and Barbara J S Sanderson. 2012. "Anti-Skin Cancer Properties of Phenolic-Rich Extract from the Pericarp of Mangosteen (*Garcinia mangostana* Linn.)." *Food and Chemical Toxicology : An International Journal Published for the British Industrial Biological Research Association* 50 (9). Elsevier Ltd: 3004–13. doi:10.1016/j.fct.2012.06.003.
- Wang, Lingling, Changseok Han, Mallikarjuna N Nadagouda, and Dionysios D Dionysiou. 2016. "An Innovative Zinc Oxide-Coated Zeolite Adsorbent for Removal of Humic Acid." *Journal of Hazardous Materials* 313. Elsevier B.V.: 283–90. doi:10.1016/j.jhazmat.2016.03.070.
- Wihelm, L.R, Dwayne A Suter, and G.H Brusewitz. 2005. "Drying and Dehydration." *Process Engineering*, no. m: 259–84.
- Wilf, Mark. 2008. "Membrane Types and Factors Affecting Membrane Performance."

In *Membrane Types and Factors Affecting Membrane Performance*, 1–92.

- Winston, Gary, Shlomo Lerman, Shalom Goldberger, Malcolm Collins, and Alex Leventhal. 2003. “A Tap Water Turbidity Crisis in Tel Aviv, Israel, due to Technical Failure: Toxicological and Risk Management Issues.” *International Journal of Hygiene and Environmental Health* 206 (3): 193–200. doi:10.1078/1438-4639-00206.
- Woo, YC, JJ Lee, LD Tijing, HK Shon, Minwei Yao, and HS Kim. 2015. “Characteristics of Membrane Fouling by Consecutive Chemical Cleaning in Pressurized Ultrafiltration as Pre-Treatment of Seawater Desalination.” *Desalination* 369. Elsevier B.V.: 51–61. doi:10.1016/j.desal.2015.04.030.
- Woo, Yun Chul, Jeong Jun Lee, Leonard D. Tijing, Ho Kyong Shon, Minwei Yao, and Han-Seung Kim. 2015. “Characteristics of Membrane Fouling by Consecutive Chemical Cleaning in Pressurized Ultrafiltration as Pre-Treatment of Seawater Desalination.” *Desalination* 369 (August). Elsevier B.V.: 51–61. doi:10.1016/j.desal.2015.04.030.
- World Health Organization. 2006. *Guidelines for Drinking-Water Quality*. World Health. Vol. 1. World Health Organization.
- Xiang, Jun, Zongli Xie, Manh Hoang, Derrick Ng, and Kaisong Zhang. 2014. “Effect of Ammonium Salts on the Properties of Poly(piperazineamide) Thin Film Composite Nanofiltration Membrane.” *Journal of Membrane Science* 465. Elsevier: 34–40. doi:10.1016/j.memsci.2014.03.074.
- Yao, Meng, Jun Nan, and Ting Chen. 2014. “Effect of Particle Size Distribution on Turbidity under Various Water Quality Levels during FI Occulation Processes.” *Des* 354. Elsevier B.V.: 116–24. doi:10.1016/j.desal.2014.09.029.
- Yin, Jun, and Baolin Deng. 2015. “Polymer-Matrix Nanocomposite Membranes for Water Treatment.” *Journal of Membrane Science* 479. Elsevier: 256–75. doi:10.1016/j.memsci.2014.11.019.
- Yu, Wenzheng, and Nigel. J.D. Graham. 2015. “Application of Fe(II)/K<sub>2</sub>MnO<sub>4</sub> as a Pre-Treatment for Controlling UF Membrane Fouling in Drinking Water Treatment.” *Journal of Membrane Science* 473. Elsevier: 283–91. doi:10.1016/j.memsci.2014.08.060.
- Zaini, Muhamad, Zawati Harun, Hatijah Basri, and Ahmad Fauzi. 2014. “Studies on Fouling by Natural Organic Matter ( NOM ) on Polysulfone Membranes : Effect of Polyethylene Glycol ( PEG ).” *DES* 333 (1). Elsevier B.V.: 36–44. doi:10.1016/j.desal.2013.11.019.

- ZEO Health Ltd. 2008. "Effects of Humic Acid on Animals and Humans Literature Review and Current Research," January.  
www.zeohealth.com/HumicAcidbenefits.pdf.
- Zhan, Xiao, Baoyu Gao, Qinyan Yue, Bin Liu, Xing Xu, and Qian Li. 2010. "Removal Natural Organic Matter by Coagulation-Adsorption and Evaluating the Serial Effect through a Chlorine Decay Model." *Journal of Hazardous Materials* 183 (1–3). Elsevier B.V.: 279–86. doi:10.1016/j.jhazmat.2010.06.132.
- Zhang, Xiaoxiao, Zhonglian Yang, Yan Wang, Bao Yu Gao, and Qinyan Yue. 2012. "The Removal Efficiency and Reaction Mechanism of Aluminum Coagulant on Organic Functional Groups-Carboxyl and Hydroxyl." *Chemical Engineering Journal* 211–212. Elsevier B.V.: 186–94. doi:10.1016/j.cej.2012.09.056.
- Zhang, Xiaozhen, Jianping Hu, Qibing Chang, Yongqing Wang, Jian-er Zhou, and Tianguai Zhao. 2015. "Influences of Internal Coagulant Composition on Microstructure and Properties of Porous YSZ Hollow Fibre Membranes for Water Treatment." *Separation and Purification Technology* 147. Elsevier B.V.: 1–9. doi:10.1016/j.seppur.2015.01.027.
- Zhang, Xuliang, Changfa Xiao, Xiaoyu Hu, Xin Jin, and Qianqian Bai. 2013. "Study on the Interfacial Bonding State and Fouling Phenomena of Polyvinylidene Fluoride Matrix-Reinforced Hollow Fiber Membranes during Microfiltration." *Desalination* 330. Elsevier B.V.: 49–60. doi:10.1016/j.desal.2013.09.022.
- Zhao, Shan, Guohe Huang, Guanhui Cheng, Yafei Wang, and Haiyan Fu. 2014. "Hardness , COD and Turbidity Removals from Produced Water by Electrocoagulation Pretreatment prior to Reverse Osmosis Membranes." *DES* 344. Elsevier B.V.: 454–62. doi:10.1016/j.desal.2014.04.014.
- Zhao, Shuang, Baoyu Gao, Shenglei Sun, Qinyan Yue, Hongyu Dong, and Wen Song. 2015. "Coagulation Efficiency, Floc Properties and Membrane Fouling of Polyaluminum Chloride in Coagulation-ultrafiltration System: The Role of Magnesium." *Colloids and Surfaces A: Physicochemical and Engineering Aspects* 469. Elsevier B.V.: 235–41. doi:10.1016/j.colsurfa.2015.01.036.
- Zhao, Shuang, Baoyu Gao, Qinyan Yue, Wuchang Song, Ruibao Jia, and Pan Liu. 2015a. "Evaluation of Floc Properties and Membrane Fouling in Coagulation-Ultrafiltration System: The Role of Enteromorpha Polysaccharides." *Desalination* 367. Elsevier B.V.: 126–33. doi:10.1016/j.desal.2015.03.041.
- . 2015b. "Evaluation of Floc Properties and Membrane Fouling in Coagulation-ultrafiltration System: The Role of Enteromorpha Polysaccharides." *Desalination* 367. Elsevier B.V.: 126–33. doi:10.1016/j.desal.2015.03.041.

- . 2015c. “Evaluation of Floc Properties and Membrane Fouling in Coagulation–ultrafiltration System: The Role of Enteromorpha Polysaccharides.” *Desalination* 367 (July). Elsevier B.V.: 126–33. doi:10.1016/j.desal.2015.03.041.
- Zhao, Shuang, Baoyu Gao, Qinyan Yue, Shenglei Sun, Wuchang Song, and Ruibao Jia. 2015. “Influence of Enteromorpha Polysaccharides on Variation of Coagulation Behavior, Flocs Properties and Membrane Fouling in Coagulation–ultrafiltration Process.” *Journal of Hazardous Materials* 285. Elsevier B.V.: 294–303. doi:10.1016/j.jhazmat.2014.12.017.
- Zhao, Song, Zhi Wang, Jixiao Wang, and Shichang Wang. 2014. “The Effect of pH of Coagulation Bath on Tailoring the Morphology and Separation Performance of Polysulfone / Polyaniline Ultra Filtration Membrane.” *Journal of Membrane Science* 469. Elsevier: 316–25. doi:10.1016/j.memsci.2014.06.054.
- Zhou, Youlian, Yuanbo Zhang, Guanghui Li, Yudong Wu, and Tao Jiang. 2015. “A Further Study on Adsorption Interaction of Humic Acid on Natural Magnetite, Hematite and Quartz in Iron Ore Pelletizing Process: Effect of the Solution pH Value.” *Powder Technology* 271. Elsevier B.V.: 155–66. doi:10.1016/j.powtec.2014.10.045.
- Zularisam, a. W., a. F. Ismail, and Razman Salim. 2006. “Behaviours of Natural Organic Matter in Membrane Filtration for Surface Water Treatment - a Review.” *Desalination* 194 (1–3): 211–31. doi:10.1016/j.desal.2005.10.030.
- Zularisam, A W, A F Ismail, M R Salim, Mimi Sakinah, and O Hiroaki. 2007. “Fabrication , Fouling and Foulant Analyses of Asymmetric Polysulfone ( PSF ) Ultrafiltration Membrane Fouled with Natural Organic Matter ( NOM ) Source Waters” 299: 97–113. doi:10.1016/j.memsci.2007.04.030.
- Hendricks. D. W. 2005. *Water treatment unit processes: Physical and chemical*. US: CRC press.