

SYNTHESIS AND CHARACTERIZATION OF  
TIN BASED HYBRID NANOFIBERS AND  
NANOFLOWERS AS PHOTOELECTRODE IN  
DYE-SENSITIZED SOLAR CELLS

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We hereby declare that we have checked this thesis and in our opinion, this thesis is adequate in terms of scope and quality for the award of the degree of Doctor of Philosophy.

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I hereby declare that the work in this thesis is based on my original work except for quotations and citations which have been duly acknowledged. I also declare that it has not been previously or concurrently submitted for any other degree at Universiti Malaysia Pahang or any other institutions.

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## **DEDICATION**

This thesis is dedicated to my beloved parents, husband, brothers, sisters, and kids.

Without whom none of my success would be possible

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## ABSTRAK

Fotoanod memainkan peranan penting dalam operasi sel suria peka-pewarna kerana pelbagai fungsi termasuk: (1) menyediakan permukaan untuk penjerapan bahan pewarna dan (2) menerima elektron yang teruja dari pewarna melalui suntikan sinaran cahaya, seterusnya mengalirkannya ke litar luar untuk menghasilkan arus elektrik. Penyelidikan yang terkandung dalam tesis ini menerangkan sintesis komposit struktur nano oksida logam serta binari oksida logam tulen serta penilaian kesesuaian bahan sebagai fotoanod dalam sel solar peka-pewarna. Oleh kerana sifat-sifat seperti tahap konduksi yang baik dalam  $\text{TiO}_2/\text{ZnO}$ , mobiliti elektron yang tinggi ( $150 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$  untuk kristal nano) dan kelebihan kekonduksian elektrik  $\text{SnO}_2$ ; oksida logam komposit dalam sistem  $\text{SnO}_2\text{-TiO}_2$  dan  $\text{SnO}_2\text{-ZnO}$  dipilih untuk dikaji dalam kerja semasa. Kaedah elektrospinning digunakan untuk mensintesis komposit nano pelbagai morfologi disebabkan keupayaannya untuk pembangunan dan penghasilan bahan nano yang berskala besar. Penghasilan komposit telah ditentukan oleh pembelauan sinar-X, penyebaran tenaga X-ray dan analisis dari spektroskopi X-ray foto elektron. Morfologi pula diuji oleh mikroskop pengimbas emisi elektron dan mikroskop transmisi dengan serakan elektron di kawasan terpilih. Kajian morfologi menunjukkan bahawa  $\text{SnO}_2\text{-TiO}_2$  terbentuk dalam dua struktur, fiber nano dan nanoflowers apabila kepekatan prekursor diselaraskan manakala  $\text{SnO}_2\text{-ZnO}$  memberikan morfologi fiber nano. Sinergi dalam sifat optik, elektronik dan komposit elektrik fiber nano ditunjukkan oleh voltametri siklik, Mott Schottky, dan spektroskopi penyerapan. Voltan-arus, kemerosotan voltan litar terbuka dan pengukuran impedans elektrokimia menunjukkan bahawa struktur nano komposit mempunyai sifat-sifat yang beerti apabila digunakan sebagai fotoanod dalam sel suria peka-pewarna disebabkan oleh kecekapan pengubahan cahaya (PCE  $\sim 5.65\%$  untuk fiber nano komposit  $\text{SnO}_2\text{-ZnO}$ ,  $7.4\%$  untuk komposit nanoflowers  $\text{SnO}_2\text{-TiO}_2$  dan  $\sim 8.5\%$  untuk fiber nano komposit  $\text{SnO}_2\text{-TiO}_2$ ) berbanding dengan logam oksida tulen  $\text{SnO}_2$  ( $\sim 3.90\%$ ),  $\text{ZnO}$  ( $\sim 1.38\%$ ) dan  $\text{TiO}_2$  fiber nano ( $\sim 5.1\%$ ). Hasil kajian ini menunjukkan kaedah ringkas dan berskala ringkas mampu untuk menghasilkan reka bentuk komposit elektrod bagi meningkatkan prestasi fungsi peranti akhir.

## ABSTRACT

Photoanode plays a crucial role in the operation of dye-sensitized solar cells due to its many functions: (1) provide a surface for the adsorption of the dye and (2) accepts photoinjected electrons from the excited dye and conducts them to the external circuit to produce an electric current. The research embodied in this thesis describes the synthesis of composite metal oxides nanostructure as well as pure binary metal oxides and evaluate their suitability as a photoanode in the dye-sensitized solar cells. Due to the advantageous properties such as favorable conduction band level of  $\text{TiO}_2/\text{ZnO}$ , high electron mobility ( $150 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$  for nanocrystals) and high electrical conductivity of  $\text{SnO}_2$ , attention is devoted to these three materials. Composite metal oxides in the  $\text{SnO}_2\text{-TiO}_2$  and  $\text{SnO}_2\text{-ZnO}$  nanostructures are chosen for the current work to overcome the limitation of the single metal oxide; such as low conduction band edge of  $\text{SnO}_2$ , low charge mobility of  $\text{TiO}_2$  and lack of stability of  $\text{ZnO}$  based photoanode in dye-sensitized solar cells. Electrospinning method is adopted for the synthesis of composite nanostructures morphologies due to the feasibility of this method for developing nanoscale materials in large scales. The composites formation is confirmed by X-ray diffraction, energy dispersive X-ray and X-ray photoelectron spectroscopy analyses. The morphology is examined by field emission scanning electron microscopy and transmission electron microscopy with selected area electron diffraction. Morphological studies show that  $\text{SnO}_2\text{-TiO}_2$  formed in two structures, nanofibers and nanoflowers by adjusting the precursor's concentration whereas  $\text{SnO}_2\text{-ZnO}$  gave nanofibers morphology. Synergy in the optical, electronic and electrical properties of the composite nanofibers is demonstrated by cyclic voltammetry, Mott-Schottky, and absorption spectroscopy. Current-voltage, Open-circuit voltage decay and electrochemical impedance measurements revealed that the composite nanostructures offering valuable properties when utilized as a photoanode in dye-sensitized solar cells in terms of photoconversion efficiency (PCE  $\sim 5.65\%$  for  $\text{SnO}_2\text{-ZnO}$  composite nanofibers,  $7.40\%$  for  $\text{SnO}_2\text{-TiO}_2$  composite nanoflowers and  $\sim 8.50\%$  for  $\text{SnO}_2\text{-TiO}_2$  composite nanofibers) compared to its binary counterparts  $\text{SnO}_2$  ( $\sim 3.90\%$ ),  $\text{ZnO}$  ( $\sim 1.38\%$ ) and  $\text{TiO}_2$  nanofibers ( $\sim 5.10\%$ ). Results of this research revealed a facile and scalable method to fabricate a simple composite electrode design for enhancing the functional performances of the final device.



## TABLE OF CONTENT

<b>DECLARATION</b>	
<b>TITLE PAGE</b>	
<b>DEDICATION</b>	
<b>ACKNOWLEDGEMENTS</b>	<b>ii</b>
<b>ABSTRAK</b>	<b>iii</b>
<b>ABSTRACT</b>	<b>iv</b>
<b>TABLE OF CONTENT</b>	<b>v</b>
<b>LIST OF TABLES</b>	<b>xi</b>
<b>LIST OF FIGURES</b>	<b>xii</b>
<b>LIST OF SYMBOLS AND UNITS</b>	<b>xvi</b>
<b>LIST OF ABBREVIATIONS</b>	<b>xviii</b>
<b>CHAPTER 1 INTRODUCTION</b>	<b>1</b>
1.1 Research Background	1
1.2 Problem Statement	2
1.3 Research Objectives	4
1.4 Research Scope	5
1.5 Statement of Contribution	6
1.6 Thesis Outline	7
<b>CHAPTER 2 LITERATURE REVIEW</b>	<b>8</b>
2.1 Chapter Overview	8
2.2 Brief History of Photovoltaics	8

2.3	Working Principle of DSSCs	10
2.4	Charge Transfer and Charge Transport in DSSCs	11
2.5	Device Structure of DSSCs	14
2.5.1	Transparent Conducting Electrode	14
2.5.2	Nanostructured Wide-Band Gap Semiconductor	15
2.5.3	Dye (Sensitizer)	17
2.5.4	Electrolyte (Hole Transport Layer)	18
2.5.5	Photocathode (Counter Electrode)	19
2.6	Photovoltaic Parameters	19
2.7	Strategies to Improve Power Conversion Efficiency and Decrease Energy Loss	20
2.8	One Dimensional Nanostructure Based Photoanode	21
2.8.1	One Dimensional TiO <sub>2</sub> Based DSSCs	23
2.8.2	One Dimensional ZnO Based DSSCs	24
2.8.3	One Dimensional SnO <sub>2</sub> Based DSSCs	27
2.9	Tailoring Charge Transport through Metals Oxide Semiconductors	31
2.9.1	Doped Photoanode	31
2.9.2	Core-Shell Photoanode	33
2.9.3	Composite Photoanode	35
2.10	Conclusions	37
<b>CHAPTER 3 METHODOLOGY</b>		<b>38</b>
3.1	Chapter Overview	38
3.2	Research Methodology	38
3.3	Synthesis of Nanofibers and Nanoflowers Composite Metals Oxide Using Electrospinning Technique	39
3.4	Characterization of the Nanomaterials	43

3.4.1	Thermogravimetric Analysis	43
3.4.2	X-Rays Diffraction Analysis	43
3.4.3	X-ray Photoelectron Spectroscopy	44
3.4.4	N <sub>2</sub> Adsorption-Desorption (Brunauer-Emmett–Teller) Analysis	45
3.4.5	Field Emission Scanning Electron Microscopy	46
3.4.6	Transmission Electron Microscopy & Selected Area Electron Diffraction Analyse	46
3.4.7	UV-Vis-NIR Spectroscopy Analysis	47
3.4.8	Cyclic Voltammetry Analysis	48
3.4.9	Mott-Schottky Analysis	48
3.4.10	Two Probe Method	49
3.5	Fabrication Steps of the Dye-Sensitized Solar Cells	49
3.5.1	Paste Preparation	50
3.5.2	Solar Fabrication	50
3.6	Characterization of the Complete Dye-Sensitized Solar Cells	51
3.6.1	Current-Voltage Measurement	51
3.6.2	Open Circuit Voltage Decay	52
3.6.3	Electrochemical Impedance Spectroscopy	52
3.7	Summary	56

**CHAPTER 4 SYNTHESIS OF COMPOSITE METAL OXIDES AND THEIR  
COUNTERPARTS** **57**

4.1	Chapter Overview	57
4.2	Precursor Concentration vs. Viscosity of the Solutions	57
4.3	Thermogravimetric Analysis	58
4.3.1	SnO <sub>2</sub> -TiO <sub>2</sub> Nanofibers and Nanoflowers	58
4.3.2	SnO <sub>2</sub> -ZnO Nanofibers	59

4.4	X-Ray Diffraction Analysis	60
4.4.1	SnO <sub>2</sub> -TiO <sub>2</sub> Nanofibers and Nanoflowers	60
4.4.2	SnO <sub>2</sub> -ZnO Nanofibers	63
4.5	X-Ray Photoelectron Spectroscopy	64
4.5.1	SnO <sub>2</sub> -TiO <sub>2</sub> Nanofibers and Nanoflowers	64
4.5.2	SnO <sub>2</sub> -ZnO Nanofibers	67
4.6	Energy-Dispersive X-Ray Spectrometry Analysis	68
4.6.1	SnO <sub>2</sub> -TiO <sub>2</sub> Nanofibers and Nanoflowers	68
4.6.2	SnO <sub>2</sub> -ZnO Nanofibers	70
4.7	Field Emission Scanning Electron Microscopy	72
4.7.1	SnO <sub>2</sub> -TiO <sub>2</sub> Nanofibers and Nanoflowers	72
4.7.2	SnO <sub>2</sub> -ZnO Nanofibers	74
4.8	Transmission Electron Microscopy & Selected Area Electron Diffraction Analyses	75
4.8.1	SnO <sub>2</sub> -TiO <sub>2</sub> Nanofibers and Nanoflowers	75
4.8.2	SnO <sub>2</sub> -ZnO Nanofibers	77
4.9	Surface Area and Pore Size Distribution Analysis	78
4.10	Conclusions	80
 <b>CHAPTER 5 OPTICAL AND ELECTROCHEMICAL PROPERTIES OF THE COMPOSITE ELECTRODES</b>		<b>81</b>
5.1	Chapter Overview	81
5.2	Optical Properties of the Composite Electrodes	81
5.2.1	SnO <sub>2</sub> -TiO <sub>2</sub> Nanofibers and Nanoflowers	81
5.2.2	SnO <sub>2</sub> -ZnO Nanofibers	82
5.3	Cycling Voltammetry	83
5.3.1	SnO <sub>2</sub> -TiO <sub>2</sub> Nanofibers and Nanoflowers	83

5.3.2	SnO <sub>2</sub> -ZnO Nanofibers	85
5.4	Two Probe Method	87
5.5	Mott–Schottky Analysis	88
5.5.1	SnO <sub>2</sub> -TiO <sub>2</sub> Nanofibers and Nanoflowers	88
5.5.2	SnO <sub>2</sub> -ZnO Nanofibers	90
5.6	Conclusions	92
 <b>CHAPTER 6 COMPOSITE NANOSTRUCTURES PHOTOANODES BASED DYE-SENSITIZED SOLAR CELLS</b>		<b>93</b>
6.1	Chapter Overview	93
6.2	Morphologies and Thickness of the Electrodes	93
6.3	Light Scattering Properties of the Dye-Anchored Electrodes	94
6.3.1	SnO <sub>2</sub> -TiO <sub>2</sub> Nanofibers and Nanoflowers	94
6.3.2	SnO <sub>2</sub> -ZnO Nanofibers	97
6.4	Current-Voltage Measurement	100
6.4.1	SnO <sub>2</sub> -TiO <sub>2</sub> Nanofibers and Nanoflowers	100
6.4.2	SnO <sub>2</sub> -ZnO Nanofibers	101
6.5	Open Circuit Voltage Decay	103
6.5.1	SnO <sub>2</sub> -TiO <sub>2</sub> Nanofibers and Nanoflowers	103
6.5.2	SnO <sub>2</sub> -ZnO Nanofibers	105
6.6	Electrochemical Impedance Spectroscopy	107
6.6.1	SnO <sub>2</sub> -TiO <sub>2</sub> Nanofibers and Nanoflowers	107
6.6.2	SnO <sub>2</sub> -ZnO Nanofibers	114
6.7	Conclusions	120
 <b>CHAPTER 7 SUMMARY AND RECOMMENDATIONS</b>		<b>121</b>
7.1	Summary	121

7.1.1	Composite Preparation and Characterization	121
7.1.2	Evaluation of Electrical, Optical and Electrochemical Properties	123
7.1.3	Dye-Sensitized Solar Cells Fabrication and Characterization	123
7.2	Recommendations	124
	<b>REFERENCES</b>	<b>126</b>
	<b>APPENDIX A PHOTOANODES MORPHOLOGIES</b>	<b>151</b>
	<b>APPENDIX B PHOTOANODES THICKNESS</b>	<b>152</b>
	<b>APPENDIX C CHARGE DENSITY CALCULATIONS</b>	<b>153</b>
	<b>APPENDIX D DYE LOADING TIME</b>	<b>154</b>
	<b>LIST OF PUBLICATIONS</b>	<b>155</b>

## LIST OF TABLES

Table 2.1	Fabrication techniques of TiO <sub>2</sub> in different morphologies and its photovoltaic parameters as photoanode based DSSCs	25
Table 2.2	Fabrication techniques of ZnO in different morphologies as photoanode based DSSCs	28
Table 2.3	Photovoltaic parameters of different morphologies of SnO <sub>2</sub> as photoanode based DSSCs	32
Table 2.4	Photovoltaic parameters for doped and composite SnO <sub>2</sub> nanostructures based on photoanode DSSCs	36
Table 4.1	Characteristics parameters of the electrospinning polymeric solutions such as precursor's concentrations and viscosities	58
Table 4.2	The values of d spacing obtained from XRD of SnO <sub>2</sub> and TiO <sub>2</sub>	61
Table 4.3	The values of d spacing obtained from XRD of SnO <sub>2</sub> -TiO <sub>2</sub> nanofibers and SnO <sub>2</sub> -TiO <sub>2</sub> nanoflowers	62
Table 4.4	Estimated unit cell parameters from the XRD patterns of the TiO <sub>2</sub> , SnO <sub>2</sub> , SnO <sub>2</sub> -TiO <sub>2</sub> nanofibers, and SnO <sub>2</sub> -TiO <sub>2</sub> nanoflowers	62
Table 4.5	The values of d spacing obtained from XRD of SnO <sub>2</sub> and ZnO and SnO <sub>2</sub> -ZnO nanofibers	64
Table 4.6	Unit cell parameters of the SnO <sub>2</sub> -ZnO nanofibers and its counterparts calculated from the XRD patterns	64
Table 4.7	Elements observed in XPS analysis of SnO <sub>2</sub> -TiO <sub>2</sub> composite nanofibers and nanoflowers and their binding energies	67
Table 4.8	Elements observed in XPS analysis of SnO <sub>2</sub> -ZnO composite nanofibers and their binding energies	68
Table 4.9	EDX results of electrospun SnO <sub>2</sub> -TiO <sub>2</sub> composite nanofibers using spectra collected from three samples locations	69
Table 4.10	EDX weight ratios of electrospun SnO <sub>2</sub> -TiO <sub>2</sub> composite nanoflowers using three spectrums focused on three distinct areas	70
Table 4.11	Elemental ratios (both atomic and weight) of electrospun SnO <sub>2</sub> -ZnO composite nanofibers collected from three distinct areas	71
Table 4.12	Surface properties of the composite metal oxides and constituent metal oxides from gas adsorption analysis	78
Table 6.1	Photovoltaics parameters of the DSSCs fabricated using pure and composite materials	101
Table 6.2	PVs parameters for ZnO, SnO <sub>2</sub> , and SnO <sub>2</sub> -ZnO composite nanofibers based DSSCs under 1 sun conditions	103
Table 6.3	Electron lifetime using OCVD measurement for the pure and composite DSSCs	105
Table 6.4	Electron lifetime using OCVD measurement for the SnO <sub>2</sub> , ZnO and SnO <sub>2</sub> -ZnO composite DSSCs	107

## LIST OF FIGURES

Figure 2.1	Structure of DSSCs based liquid electrolyte	10
Figure 2.2	Electron transfer mechanism in DSSCs based liquid electrolyte	12
Figure 2.3	Typical time rates for the favourable reactions (green) and unfavourable reactions (red) in a Ru-DSSCs with iodide/triiodide electrolyte under working conditions ( $100 \text{ mW cm}^{-2}$ )	13
Figure 2.4	Simulated I-V curve (red) and simulated power curve (green) showing performance parameters of DSSCs including $I_{SC}$ , $V_{OC}$ , and maximum power point	20
Figure 2.5	Schematic illustration of electron diffuse transport in (a) a conventional nanoparticle electrode (zigzag pathway) and (b) a 1D electrode (direct pathway)	22
Figure 2.6	Core-shell mechanism in DSSCs: (a) the edge of CB of the shell material is lower than the LUMO of dye and (b) the edge of CB of the shell material is higher the LUMO and higher than the CB edge of the core material	34
Figure 3.1	Schematics of research methodology	40
Figure 3.2	(a) Electrospinning technique for producing nanofibers with (b) FESEM image for annealed nanofibers	41
Figure 3.3	Schematic representation of the synthesis of $\text{SnO}_2\text{-TiO}_2$ nanoflowers and nanofibers by changing metal precursors concentration via electrospinning technique	42
Figure 3.4	The set up for two probe measurement	49
Figure 3.5	Fabrication procedures of the DSSCs	50
Figure 3.6	Transmission line model circuit of a complete DSSCs for the extraction of charge transport parameters	53
Figure 3.7	Equivalent circuit of complete DSSCs to fit the EIS spectra on Z-view, where DX1 represents (b) the transmission line for diffusion-recombination at the photoelectrode	56
Figure 4.1	TGA for as prepared $\text{SnO}_2\text{-TiO}_2\text{/PVP}$ composite material	58
Figure 4.2	TGA for as prepared $\text{SnO}_2\text{-TiO}_2\text{/PVAc}$ composite material	59
Figure 4.3	TGA for as prepared $\text{SnO}_2\text{-ZnO/PVP}$ composite nanofibers	60
Figure 4.4	XRD patterns of the electrospun hybrids $\text{SnO}_2\text{-TiO}_2$ along with $\text{SnO}_2$ and $\text{TiO}_2$ nanofibers	61
Figure 4.5	X-ray diffraction patterns of $\text{SnO}_2\text{-ZnO}$ , $\text{SnO}_2$ and $\text{ZnO}$ nanofibers	63
Figure 4.6	(a) Full XPS survey scan spectrum of $\text{SnO}_2\text{-TiO}_2$ composite nanofibers; XPS spectra of (b) Ti 2p showing two fitted peaks; (c) fitted Sn 3d spectrum displaying two distinct peaks; (d) deconvoluted peak corresponding to O 1s	65



Figure 4.7	(a) Full XPS survey scan spectrum of SnO <sub>2</sub> -TiO <sub>2</sub> composite nanoflowers, XPS spectra of (b) Ti 2p showing two fitted peaks (inset shows a comparison between nanofibers and nanoflowers), (c) fitted Sn 3d spectrum displaying two distinct peaks, and (d) deconvoluted peak corresponding to O 1s	66
Figure 4.8	XPS full survey scan spectrum of SnO <sub>2</sub> -ZnO composite nanofibers; XPS spectra of (b) fitted Sn 3d spectrum displaying two distinct peaks; (c) Zn 2p showing two fitted peaks and; (d) deconvoluted peak of O 1s	68
Figure 4.9	EDX spectra of SnO <sub>2</sub> -TiO <sub>2</sub> composite nanofibers from three selected areas	69
Figure 4.10	EDX spectra of SnO <sub>2</sub> -TiO <sub>2</sub> composite nanoflowers from three selected areas	70
Figure 4.11	EDX spectra of SnO <sub>2</sub> -ZnO composite nanofibers from three selected area	71
Figure 4.12	FESEM surface morphology (a-d) and cross section (e, f) of SnO <sub>2</sub> -TiO <sub>2</sub> composite nanofibers	72
Figure 4.13	FESEM surface morphology (a, b) and high resolution (c, d) of SnO <sub>2</sub> -TiO <sub>2</sub> composite nanoflowers	73
Figure 4.14	FESEM images of the annealed SnO <sub>2</sub> -ZnO nanofibers (a-d) the surface morphology and nanofibers diameters and (e, f) cross-section view present the formation of porous nanofibers	74
Figure 4.15	(a, b) bright field TEM images, (c) HRTEM lattice images, (d) SAED pattern of SnO <sub>2</sub> -TiO <sub>2</sub> composite nanofibers	75
Figure 4.16	(a, b) TEM images, (c) HRTEM lattice images, and (d) SAED pattern of SnO <sub>2</sub> -TiO <sub>2</sub> composite nanoflowers	76
Figure 4.17	(a, b) TEM images, (c) HRTEM lattice images, and (d) SAED pattern of SnO <sub>2</sub> -ZnO composite nanofibers	77
Figure 4.18	Nitrogen adsorption-desorption isotherms of (a) SnO <sub>2</sub> , (b) TiO <sub>2</sub> , (c) SnO <sub>2</sub> -TiO <sub>2</sub> nanofibers, (d) SnO <sub>2</sub> -TiO <sub>2</sub> nanoflowers, (e) SnO <sub>2</sub> -ZnO and (f) ZnO nanofibers; (inset shows BJH isotherms of corresponding sample)	79
Figure 5.1	Diffuse reflectance spectra of the composite photoanode and their counterparts presented in terms of plots of $(h\nu F(R))^2$ versus photon energy ( $h\nu$ ): The inset shows details of optical absorption onsets in terms of the reflectance versus wavelength	82
Figure 5.2	Kubelka-Munk plots of the SnO <sub>2</sub> -ZnO composite nanofibers along with their counterparts: The inset presents the UV-Vis diffuse reflectance spectra	83
Figure 5.3	The CV data of the SnO <sub>2</sub> nanofibers, TiO <sub>2</sub> nanofibers, SnO <sub>2</sub> -TiO <sub>2</sub> composite nanofibers and SnO <sub>2</sub> -TiO <sub>2</sub> composite nanoflowers electrodes in 5 M KOH at scan rate 10 mV s <sup>-1</sup>	85

Figure 5.4	CV curves of the SnO <sub>2</sub> nanofibers, ZnO nanofibers and SnO <sub>2</sub> -ZnO composite nanofibers electrodes measured in 5 M KOH at a scan rate of 10 mV s <sup>-1</sup>	86
Figure 5.5	I-V plot of (a) SnO <sub>2</sub> -TiO <sub>2</sub> nanofibers and SnO <sub>2</sub> -TiO <sub>2</sub> nanoflowers along with TiO <sub>2</sub> nanofibers, SnO <sub>2</sub> nanofibers and (b) SnO <sub>2</sub> -ZnO nanofibers with ZnO nanofibers, SnO <sub>2</sub> nanofibers	87
Figure 5.6	Mott-Schottky analysis of SnO <sub>2</sub> , TiO <sub>2</sub> nanofibers and SnO <sub>2</sub> -TiO <sub>2</sub> composite (nanofibers and nanoflowers) electrodes measured in 1 M Na <sub>2</sub> SO <sub>4</sub>	89
Figure 5.7	Mott-Schottky curves of SnO <sub>2</sub> nanofibers, SnO <sub>2</sub> -ZnO composite nanofibers and ZnO nanofibers electrodes measured in 1 M Na <sub>2</sub> SO <sub>4</sub> electrolyte	91
Figure 6.1	Cross section of (a, b) SnO <sub>2</sub> -TiO <sub>2</sub> , (c, d) SnO <sub>2</sub> -ZnO composite nanofibers, and (e, f) SnO <sub>2</sub> -TiO <sub>2</sub> composite nanoflowers photoanodes films at two magnifications	94
Figure 6.2	Absorption spectra of the composites dye anchored photoanodes and its counterparts	95
Figure 6.3	Transmission spectra of the composites dye anchored photoanode and its counterparts	96
Figure 6.4	UV-Vis absorbance spectra of the dyes solutions that were desorbed from SnO <sub>2</sub> , TiO <sub>2</sub> and SnO <sub>2</sub> -TiO <sub>2</sub> nanofibers and nanoflowers photoanodes used in the DSSCs.	97
Figure 6.5	Absorption spectra of the composite dye anchored photoanode and its counterparts	98
Figure 6.6	Transmission spectra of the composite dye anchored photoanode and its counterparts	99
Figure 6.7	UV-Vis absorbance spectra of the solutions of dyes that were desorbed from SnO <sub>2</sub> , ZnO and SnO <sub>2</sub> -ZnO nanofibers photoanodes used in the DSSCs.	99
Figure 6.8	J-V curves of the solar cells fabricated devices using the TiO <sub>2</sub> , SnO <sub>2</sub> , and SnO <sub>2</sub> -TiO <sub>2</sub> nanofibers and SnO <sub>2</sub> -TiO <sub>2</sub> nanoflowers under 1 sun condition. All the photoanodes were sensitized for 24 h	100
Figure 6.9	J-V curves of the cells fabricated using ZnO nanofibers, SnO <sub>2</sub> nanofibers and SnO <sub>2</sub> -ZnO nanofibers under 1 sun condition	102
Figure 6.10	Open circuit voltage decay curves of SnO <sub>2</sub> , TiO <sub>2</sub> , SnO <sub>2</sub> -TiO <sub>2</sub> nanofibers, and SnO <sub>2</sub> -TiO <sub>2</sub> nanoflowers	104
Figure 6.11	An electron lifetime calculated from the OCVD curve at the dark for composite and pure photoanodes based DSSCs	105
Figure 6.12	OCVD curves of SnO <sub>2</sub> -ZnO composite nanofibers with SnO <sub>2</sub> and ZnO nanofibers	106
Figure 6.13	An electron lifetime calculated from the OCVD curve at the dark for composite and pure photoanodes based DSSCs	107

Figure 6.14	Nyquist plots of (a) SnO <sub>2</sub> nanofibers, (b) TiO <sub>2</sub> nanofibers, (c) SnO <sub>2</sub> -TiO <sub>2</sub> nanofibers, (d) SnO <sub>2</sub> -TiO <sub>2</sub> nanoflowers at different applied voltages,(e) for all electrodes at a bias voltage of 0.7 V, and (f) The magnified area of the Nyquist plot at the high frequency region showing the differences in the transport resistance	109
Figure 6.15	Bode plots for the composite electrodes and their counterparts at an applied voltage of 0.7 V (a) magnitude and (b) phase angle	110
Figure 6.16	(a) charge transport resistance, (b) recombination resistance between the semiconductors metal oxides and the acceptor species in the electrolyte, (c) transit time,(d) electron lifetime, and (e) ratio between transit and electron lifetime as function of applied voltage measured under dark conditions for SnO <sub>2</sub> nanofibers, TiO <sub>2</sub> nanofibers, SnO <sub>2</sub> -TiO <sub>2</sub> nanofibers and SnO <sub>2</sub> -TiO <sub>2</sub> nanoflowers photoanodes	113
Figure 6.17	(a) chemical capacitance and (b) $L_n/d$ as a function of applied voltage measured under dark conditions for SnO <sub>2</sub> nanofibers, TiO <sub>2</sub> nanofibers, SnO <sub>2</sub> -TiO <sub>2</sub> nanofibers, and SnO <sub>2</sub> -TiO <sub>2</sub> nanoflowers photoanodes	114
Figure 6.18	Nyquist plot of the (a) SnO <sub>2</sub> nanofibers, (b) ZnO nanofibers, (c) SnO <sub>2</sub> -ZnO nanofibers at a different applied voltage, (d) Nyquist plot for the three electrodes at 0.7 V, and (e) Magnified portion of the Nyquist plot at the high-frequency region showing the differences in the transport resistance	115
Figure 6.19	Bode plots for three electrodes at an applied voltage of 0.7 V (a) magnitude and (b) phase angle	116
Figure 6.20	(a) charge transport resistance, (b) recombination resistance between the semiconductors metal oxides and the acceptor species in the electrolyte, (c) electron transit time and (d) electron lifetime calculated from impedance spectroscopy measurements as a function of applied voltages for SnO <sub>2</sub> -ZnO nanofibers and its counterparts	118
Figure 6.21	(a) chemical capacitance and (b) $L_n/d$ calculated from impedance spectroscopy measurements as a function of applied voltages for SnO <sub>2</sub> -ZnO nanofibers and its counterparts	119

## LIST OF SYMBOLS AND UNITS

A	Absorption
$k_B$	Boltzmann constant
$N_D$	Charge density
cP	Centipoise
J	Current density
cm	Centimeter
$\eta_{cc}$	Charge collection efficiency
$\epsilon_m$	Dielectric constant of the semiconductor
$L_n$	Electron diffuse length
n	Electron density
$D_n$	Electron diffuse coefficient
$\tau_n$	Electron lifetime
$\tau_d$	Electron transit time
$\mu_e$	Electron mobility
$E_g$	Energy gap
F	Farad
$V_{fb}$	Flat-band potential
$\nabla E_F$	Gradient in Fermi levels of a materials interface
$Z''$	Imaginary impedance
k	Kilo
$I_{max}$	Maximum current
$f_{max}$	Maximum frequency
$P_{max}$	Maximum power
$V_{max}$	Maximum voltage
$\mu$	Micro
ms	Millisecond
mV	Millivolt
$\epsilon$	Molar extinction coefficient
$V_{oc}$	Open circuit voltage
$\Omega$	Ohm
d	Photoelectrode thickness

$Z'$	Real impedance
$R_{\text{Rec}}$	Recombination resistance
$R$	Reflectance
$C_{\mu}$	Photoanode chemical capacitance
$C$	Space charge capacitance
$J_{\text{SC}}$	Short circuit current density
$R_t$	Transport resistance
$R_s$	Series resistance
$R_{\text{Sh}}$	Shunt resistance
$T$	Transmittance
$T$	Temperature
$\lambda$	Wavelength
$\epsilon_0$	Vacuum permittivity

## LIST OF ABBREVIATIONS

ALD	Atomic layer deposition
AM	Air mass
ACG	Aqueous chemical growth
BET	Brunauer-Emmett-Teller
CB	Conduction band
CBD	Chemical bath deposition
CPM	Chemical precipitation method
CV	Cyclic voltammetry
CE	Counter electrode
CPE	Constant phase element
CdTe	Cadmium telluride
CIGS	Copper indium gallium diselenide
DCTP	Direct current thermal plasma
DSSCs	Dye-sensitized solar cells
EM	Electrochemical method
EIS	Electrochemical impedance spectroscopy
EDX	Energy dispersive X-ray spectroscopy
EDM	Electrodeposited method
ES	Electrospun method
FSP	Flame spray pyrolysis.
FESEM	Field emission scanning electron microscopy
FF	Fill factor
FTO	Fluorine-doped tin oxide
HTM	Hydrothermal method
HOMO	Highest occupied molecular orbitals
HRTEM	High-resolution transmission electron microscopy
H	Hydrolysis
ITO	Indium-tin-oxide
pI	Isoelectric point
LSV	Linear sweep voltammetry
LPD	liquid-phase deposition

LUMO	Lowest unoccupied molecular orbitals
MOS	Metal oxide semiconductor
MS	Mott–Schottky
NFs	Nanofibers
NPs	Nanoparticle
NRs	Nanorods
NTs	Nanotubes
NWs	Nanowires
1D	One dimensional
OCVD	Open circuit voltage decay
PSCs	Perovskite solar cells
PV	Photovoltaic
PCE	Power conversion efficiency
Pt	Platinum
PCE	Polyethylene naphthalate
SAED	Selected area electron diffraction
S	Sputtering
SC	Spin coating
STM	Solvothermal method
TL	Transmission line
TE	Thermal evaporation
3D	Three-dimensional materials
2D	Two-dimensional materials
TPM	Two probe method
TEM	Transmission electron microscopy
TCO	Transparent conductive oxide
TGA	Thermogravimetric analysis
UV-Vis-NIR	Ultraviolet-visible near-infra-red
VB	Valance band
WE	Working electrode
XRD	X-ray diffraction
XPS	X-ray photoelectron spectroscopy

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