

## Thin chemisorbed polyaniline film on cobalt oxide as an electrode for hybrid energy storage devices

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### ABSTRACT

Electrical charge storing electrodes and their surface modification are intensively investigated to improve the charge storability indicators in electrochemical energy storage devices. Here, the effects of a thin chemisorbed polyaniline (PANI) film on the charge storage behavior of rod-shaped spinal-type cobalt oxide ( $\text{Co}_3\text{O}_4$ ) nanorods ( $\text{PANI@Co}_3\text{O}_4$ ) are detailed for fabrication of battery–supercapacitor hybrid (BSH) devices. The  $\text{PANI@Co}_3\text{O}_4$  showed larger surface area and optimum porosity properties, which contributed to ~50 % enhanced specific charge than that in the  $\text{Co}_3\text{O}_4$ . The deconvoluted total charge storage gain showed more contribution to the bulk-diffusion controlled process (battery-type), lower ion transport resistance and Warburg impedance in the  $\text{PANI@Co}_3\text{O}_4$  electrode than that in the  $\text{Co}_3\text{O}_4$ . Two-sets of BSH devices are fabricated using  $\text{PANI@Co}_3\text{O}_4$  as a positive electrode and mesoporous carbon (MC) and activated carbon (AC) negative electrodes in an aqueous electrolyte and benchmarked with symmetric supercapacitors fabricated using the two carbons. The  $\text{PANI@Co}_3\text{O}_4//\text{MC}$  device showed nearly two-fold higher specific energy ( $E_s$ ) than that of  $\text{PANI@Co}_3\text{O}_4//\text{AC}$ . Interestingly,  $\text{AC//AC}$  symmetric supercapacitors showed two-fold higher  $E_s$  than the  $\text{MC//MC}$  device. Origin of differences in the charge storage behavior of the two types of devices are systematically analyzed and reported.

### KEYWORDS

Chemisorption; Electrochemical energy storage; Mesoporous materials; Polymer-nanocomposites; Pseudocapacitors

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