

WASTE-TO-WEALTH: FROM PRINTER TONER WASTE INTO VALUABLE MAGNETIC MATERIALS

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Product Background



- **Objective**
 - To recover magnetite (Fe2O3) by magnetic separation technique and convert into Fe2O3 using reduction technique.
 - To synthesize Nickel Zinc Ferrite (NZF) using extracted iron oxide from waste toner powder.
 - To study the structural, microstructural and magnetic properties of Nickel Zinc Ferrite (NZF) prepared using extracted Fe2O3 as main component.

Environmental Impact

- Valuable metals from waste toner were recovered.
- The number of discarded toner cartridge can be reduced.
- Reduce leaching of toxic chemicals from toner powder into landfill.

1) The residual toner powder contains about 8% for an end-of-life toner for each toner cartridge.

2) The synthesizing of magnetite iron oxide can be extracted by using magnetic separation and reduction technique .

3) The extraction is done via simple and lowcost alternative.

State of the Art/ Methods





Graphical Results





SEM image of a) waste toner and b) Fe₂O₃ at magnification of 3000x.



Conclusion

Based on XRD analysis, the major composition in waste toner was iron oxide (Fe2O3).

- The addition of PET to magnetite powder acts as a reducing agent for reduction process from magnetite (Fe3O4) to hematite (Fe2O3).
- As for saturation magnetization, Fe3O4 sample had the highest value with 3.4878 emu/g ٠ compared to Fe2O3 sample with 0.23313 emu/g while for NZF was 0.16075 emu/g.
- The area for all samples have smaller coercive forces which mean that they have smaller ٠ hysteresis loss, hence the samples were easily magnetized and demagnetized.
- As for squareness ratio, the NZF sample had larger value with 0.45722 compared to Fe2O3 ٠ and Fe3O4 with 0.43661 and 0.14325 respectively.

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Status of Innovation

- TRL Level 1
- Status of Finished Product: Proof of concept
- Functionality of Product: Replacement of hematite (Fe₂O₃) used in ferrites