

Catalytic gasification of empty palm fruit bunches using charcoal and bismuth oxide for syngas production

Minhaj Uddin Monir^{a,b}, Azrina Abd Aziz^c, Kaykobad Md. Rezaul Karim^d, Fatema Khatun^{b,c}, Mostafa Tarek^e, Abu Yousuf^f & Dai-Viet N. Vo^g

^a Department of Petroleum and Mining Engineering, Jashore University of Science and Technology, Jashore, 7408, Bangladesh

^b Energy Conversion Laboratory, Department of Petroleum and Mining Engineering, Jashore University of Science and Technology, Jashore, 7408, Bangladesh

^c Faculty of Civil Engineering Technology, Universiti Malaysia Pahang, Pahang, Gambang, Kuantan, 26300, Malaysia

^d Chemistry Discipline, Khulna University, Khulna, 9208, Bangladesh

^e Department of Chemical Engineering, College of Engineering, Universiti Malaysia Pahang, Gambang, Pahang, 26300, Malaysia

^f Department of Chemical Engineering and Polymer Science, Shahjalal University of Science and Technology, Sylhet, 3114, Bangladesh

^g Center of Excellence for Green Energy and Environmental Nanomaterials (CE@GrEEN), Nguyen Tat Thanh University, 300A Nguyen Tat Thanh, District 4, Ho Chi Minh City, 755414, Viet Nam

ABSTRACT

The purpose of this research is to evaluate the intent of empty fruit bunches of palm oil (EFB_{palm oil}) to catalytic gasification of wood produced charcoal (Wood_{charcoal}) in order to notify the large-scale application of Wood_{charcoal} as a possible gasification feedstock. In this study, co-catalyst of bismuth oxide (Bi₂O₃) was also used to obtain syngas. The raw samples were characterized by proximate and ultimate analyses, X-ray diffraction (XRD), field emission scanning electron microscope (FESEM), transmission electron microscopy (TEM) and X-ray photoelectron spectroscopy (XPS) analyses. The produced syngas was analyzed by online portable gas analyzer and gas chromatography-thermal conductivity detector (GC-TCD). The syngas composition of H₂ increased from 3.91 to 4.70% (increased 20.20%), CO increased from 5.73 to 6.30% (increased 10.53%), whereas CO₂ decreased from 20.60 to 12.67% (decreased 38.50%) and CH₄ concentration increased insignificantly from 0.35 to 0.37% (increased 5.7%) which was happened due to the use of Wood_{Charcoal} and Bi₂O₃ with EFB_{palm oil} during gasification. According to the findings, carbon is abundant in Wood_{Charcoal}, which may considerably boost the gasification reactivity with Bi₂O₃. The yield of syngas (H₂ and CO) increased when Wood_{Charcoal} and Bi₂O₃ were used instead of single EFB_{ppo} gasification, indicating that catalyst (Wood_{Charcoal}) and co-catalyst (Bi₂O₃) have a high potential for thermal decomposition and dehydrogenation of volatile matter. Therefore, catalytic gasification of empty palm fruit bunches will be the prospective energy sources for the production of syngas with the utilization of Wood_{Charcoal} and Bi₂O₃.

KEYWORDS

Bi₂O₃; Catalytic gasification; EFB_{palm oil}; Syngas; Wood_{Charcoal}

ACKNOWLEDGEMENTS

The authors want to acknowledge the funding of Malaysia Ministry of Higher Education via Fundamental Research Grant Scheme (FRGS) No. FRGS/1/2017/TK02/UMP/02/20. The authors would like to thank Universiti Malaysia Pahang for providing lab facilities. The authors would also like to thank Dr. Rajesh Chanda and Dr. Saiful Islam for their kind help during the revision of this article.