

**THERMOPHYSICAL PROPERTIES  
CHARACTERISATION  
OF WATER BASED MANGO OXIDE  
HYBRID NANOFUID**

**JAMES LAU TZE CHEN**

**Master of Science**

**UNIVERSITI MALAYSIA PAHANG**



### **SUPERVISOR'S DECLARATION**

We hereby declare that we have checked this thesis and in our opinion, this thesis is adequate in terms of scope and quality for the award of the degree of Master of Science.

A handwritten signature in black ink, appearing to read "Ahmed Oumer".

(Supervisor's Signature)

Full Name : DR. AHMED NURYE OUMER

Position : SENIOR LECTURER

Date : 16/12/2020

A handwritten signature in black ink, enclosed within a large circle.

(Co-supervisor's Signature)

Full Name : DR. AZIZUDDIN BIN ABD AZIZ

Position : SENIOR LECTURER

Date : 16/12/2020



### **STUDENT'S DECLARATION**

I hereby declare that the work in this thesis is based on my original work except for quotations and citations which have been duly acknowledged. I also declare that it has not been previously or concurrently submitted for any other degree at Universiti Malaysia Pahang or any other institutions.

A handwritten signature in black ink, appearing to read "James Lau Tze Chen".

---

(Student's Signature)

Full Name : JAMES LAU TZE CHEN

ID Number : MME18002

Date : 16/12/2020

**THERMOPHYSICAL PROPERTIES CHARACTERISATION  
OF WATER BASED MANGO OXIDE  
HYBRID NANOFUID**

**JAMES LAU TZE CHEN**

Thesis submitted in fulfillment of the requirements  
for the award of the degree of  
**Master of Science**

College of Engineering  
**UNIVERSITI MALAYSIA PAHANG**

**DECEMBER 2020**

## **ACKNOWLEDGEMENTS**

First and foremost, I am very grateful to God for giving me wisdom and strength throughout my master by research program. Although many obstacles during my study, I feel blessed that I can complete my study smoothly under His guidance and blessing.

Next, I wish to express my sincere gratitude to my main supervisor, Dr Ahmed Nurye Oumer for giving me for his full support to complete my graduate study. I also would like to thank my co-supervisor, Dr Azizuddin Bin Aziz for his assistance and financial support. Since the day I began my master research, their continuous guidance, encouragement, support and ideas make this research to be successful.

I would also like to thank Universiti Malaysia Pahang for granting me research funding to finish my research as this work was supported by Universiti Malaysia Pahang under Grant RDU180389. I also wish to thank the Faculty of Mechanical Engineering and Institute of Postgraduate Studies for providing help directly or indirectly to finish my studies.

Last but not least, I wish to thank my beloved family members for their moral support and encouragement to attain my goals. I am also thankful to my friends who always giving me support, determination and encouragement to complete my study.

## **ABSTRAK**

Nanofluid boleh dijelaskan sebagai campuran nanopartikel dan bendalir. Ramai penyelidik cuba meningkatkan kekonduksian haba bendalir dengan menambahkan nanopartikel logam dan bukan logam dalam bendalir kerana kekonduksian haba nanopartikel logam dan bukan logam lebih tinggi daripada bendalir. Tetapi, ia juga meningkatkan kelikatan bendalir menyebabkan penurunan tekanan tinggi dalam aplikasi pemindahan haba. Ini menghasilkan kuasa pam yang lebih tinggi diperlukan dan seterusnya meningkatkan kos operasi sistem. Oleh itu, kombinasi nanopartikel logam atau bukan logam dengan nanopartikel bio tersebar dalam bendalir boleh menjadi satu pilihan untuk mengurangkan penurunan tekanan sambil mengekalkan aplikasi pemindahan haba. Ini adalah kerana penambahan nanopartikel bio yang bersifat serat dalam bendalir dapat menghasilkan peningkatan kekonduksian terma yang rendah dengan trend kelikatan yang berkurangan secara mendadak. Tiga objektif dalam kajian ini adalah untuk menyediakan nanofluid hibrid mango / oksida, menilai kestabilan serta sifat termofilik nanofluid hibrid mango / oksida dan menentukan keadaan untuk sifat haba yang lebih baik dan kelikatan yang berkurangan. Nanopartikel kulit mangga (MB) dan daun mangga (ML) mula-mula dihasilkan menggunakan kaedah top-down. Seterusnya, nanopartikel MB dan ML yang disediakan campur dengan nanopartikel oksida dan gantung ke dalam bendalir dengan menggunakan pengaduk dan mandi ultrasonik. Kestabilan nanofluid diukur menggunakan kaedah pemendapan dan analisis UV-Vis. Seterusnya, ketumpatan, kelikatan, kekonduksian haba dan kapasiti haba tertentu nanofluid dinilai menggunakan kaedah teori dan eksperimen. Selepas itu, keadaan untuk sifat haba yang dipertingkatkan dan kelikatan berkurang nanofluid ditentukan menggunakan Kaedah Surface Response (RSM). Dari keputusan, nanofluid mempunyai kestabilan yang sederhana dengan tanda pemendapan selepas 1 hari dan meningkat sedikit selepas 14 hari. Semua nanofluid stabil dan boleh digunakan pada panjang gelombang 300nm dengan purata serapan puncak 2.491. Nanofluid MB/TiO<sub>2</sub> dan ML/TiO<sub>2</sub> mempunyai 12.2% dan 14.7% kelikatan berkurangan berbanding dengan TiO<sub>2</sub> nanofluid. Nanofluid MB / SiO<sub>2</sub> dan ML / SiO<sub>2</sub> mempunyai 3.2% dan 4.4% kelikatan berkurang berbanding dengan SiO<sub>2</sub> nanofluid. Nanofluid MB/TiO<sub>2</sub> dan ML/TiO<sub>2</sub> telah mengekalkan peningkatan kekonduksian haba dengan 4.5% dan 5.4% lebih rendah berbanding dengan TiO<sub>2</sub> nanofluid. Nanofluida MB/SiO<sub>2</sub> dan ML/SiO<sub>2</sub> telah mengekalkan peningkatan kekonduksian haba dengan 1.1% dan 1.6% lebih rendah berbanding dengan nanofluid SiO<sub>2</sub>. Kapasiti dan kapasiti haba tertentu bergantung kepada jenis bahan nanopartikel. Kelikatan menurun dengan suhu dan meningkat dengan kepekatan. Sementara itu, kekonduksian haba meningkat dengan suhu dan kepekatan. Dari hasil RSM, nanofluid MB/oksida mempunyai kekonduksian haba yang paling tinggi dan kelikatan yang paling rendah dengan jenis bahan MB/SiO<sub>2</sub> (12.92% SiO<sub>2</sub> dan 87.08% MB), suhu 70°C dan kepekatan 0.25. Sementara itu, nanofluid ML/oksida mempunyai kekonduksian haba yang paling tinggi dan kelikatan yang paling rendah dengan jenis bahan MB/SiO<sub>2</sub> (42.98% SiO<sub>2</sub> dan 57.02% ML), suhu 70°C dan kepekatan 0.25%. Kesimpulannya, nanofluid hibrid mangga/oksida mempunyai kestabilan yang sederhana dengan kelikatan yang berkurangan seterusnya mengekalkan kekonduksian haba daripada nanofluid berdasarkan bukan bio. Oleh itu, nanofluid hibrid mangga/oksida boleh digunakan untuk mengurangkan penurunan tekanan sambil mengekalkan aplikasi pemindahan haba.

## ABSTRACT

Nanofluid can be described as a mixture of nanoparticles and base fluid. Many researchers attempted to enhance the thermal conductivity of base fluids by adding metallic and non-metallic nanoparticles in fluids. Although metallic and non-metallic nanoparticles can increase the thermal conductivity of base fluids, it also increases the viscosity of base fluids that causes a high pressure drop in the heat transfer application. This results that higher pumping power is needed which in turn increases the system operation cost. Therefore, an alternative solution needs to be devised. For this, hybrid nanofluid with a combination of metallic or non-metallic nanoparticles and bio nanoparticles dispersed in a base fluid can be one option to reduce the pressure drop further while maintaining the heat transfer application. This is because of the addition of bio nanoparticles that have fibre by nature in fluids resulting in low thermal conductivity enhancement with an exponentially decreasing trend of viscosity that might reduce pressure drop. Thus, the three objectives of this research are to determine the stability of mango/oxide hybrid nanofluid, to evaluate thermophysical properties of mango/oxide hybrid nanofluid and to measure the optimum parameters for enhanced thermal properties and reduced viscosity. Mango bark (MB) and mango leaf (ML) nanoparticles were first produced using the top-down method. In the second processing step, the MB and ML nanoparticles that are prepared separately mixed with oxide nanoparticles and suspended into the base fluid using stirrer and an ultrasonic bath. Stability of the nanofluids was measured using sedimentation method and UV-Vis analysis. Then, density, viscosity, thermal conductivity and specific heat capacity of the nanofluids were evaluated using theoretical and experimental method. After that, the optimum parameters for enhanced thermal properties and reduced viscosity of nanofluids were determined using Response Surface Methodology (RSM). From the results, the nanofluids have stability in moderation with the sign of sedimentation after 1 day and slightly increased after 14 days. All nanofluids are stable and applicable at 300nm wavelength with average peak absorbance of 2.491. At 30°C with 1% of volume concentration, MB/TiO<sub>2</sub> and ML/TiO<sub>2</sub> water-based nanofluids have 12.2% and 14.7% reduced viscosity than TiO<sub>2</sub> water-based nanofluid. MB/SiO<sub>2</sub> and ML/SiO<sub>2</sub> water-based nanofluids have 3.2% and 4.4% reduced viscosity than SiO<sub>2</sub> water-based nanofluid. At 30°C with 1% of volume concentration, MB/TiO<sub>2</sub> and ML/TiO<sub>2</sub> water-based nanofluids have maintained thermal conductivity enhancement with 4.5% and 5.4% lower than TiO<sub>2</sub> water-based nanofluid. MB/SiO<sub>2</sub> and ML/SiO<sub>2</sub> water-based nanofluids have maintained thermal conductivity enhancement with 1.1% and 1.6% lower than SiO<sub>2</sub> water-based nanofluid. Density and specific heat capacity are dependent on the material type. The viscosity decreased with temperature and increased with concentration. Meanwhile, the thermal conductivity increased with temperature and concentration. From the results of RSM, MB/oxide nanofluids have the most enhanced thermal conductivity and reduced viscosity with the material type of MB/SiO<sub>2</sub> (12.92% SiO<sub>2</sub> and 87.08% of mango bark), the temperature of 70°C and concentration of 0.25. Meanwhile, ML/oxide nanofluid have the most enhanced thermal conductivity and reduced viscosity with the material type of MB/SiO<sub>2</sub> (42.98% SiO<sub>2</sub> and 57.02% of mango leaf), the temperature of 70°C and concentration of 0.25%. In conclusion, mango/oxide hybrid nanofluids have stability in moderation with reduced viscosity and maintained thermal conductivity enhancement than non-bio based nanofluids that can be used to reduce the pressure drop further while maintaining the heat transfer application.

## **TABLE OF CONTENT**

### **DECLARATION**

### **TITLE PAGE**

<b>ACKNOWLEDGEMENTS</b>	ii
-------------------------	----

<b>ABSTRAK</b>	iii
----------------	-----

<b>ABSTRACT</b>	iv
-----------------	----

<b>TABLE OF CONTENT</b>	v
-------------------------	---

<b>LIST OF TABLES</b>	ix
-----------------------	----

<b>LIST OF FIGURES</b>	xi
------------------------	----

<b>LIST OF SYMBOLS</b>	xiv
------------------------	-----

<b>LIST OF ABBREVIATIONS</b>	xv
------------------------------	----

<b>CHAPTER 1 INTRODUCTION</b>	1
-------------------------------	---

1.1    Background	1
-------------------	---

1.2    Problem Statement	3
--------------------------	---

1.3    Objective	4
------------------	---

1.4    Scope of study	4
-----------------------	---

<b>CHAPTER 2 LITERATURE REVIEW</b>	5
------------------------------------	---

2.1    Introduction	5
---------------------	---

2.2    Preparation of bio-nanoparticles	6
---	---

2.3    Bio/non-bio nanofluid preparation	7
--	---

2.3.1    One-step method	7
--------------------------	---

2.3.2    Two-step method	8
--------------------------	---

2.4	Stability	9
2.4.1	Stability measurement methods	9
2.4.2	Stability enhancement techniques	12
2.5	Thermo-physical properties of bio/metal-nanofluids	13
2.5.1	Density	13
2.5.2	Viscosity	15
2.5.3	Thermal conductivity	18
2.5.4	Specific heat capacity	20
2.6	Factor affecting thermo-physical properties	22
2.6.1	Effect of material type	22
2.6.2	Effect of temperature	25
2.6.3	Effect of concentration	28
2.6.4	Effect of base-fluid	31
2.7	Conclusion	33

<b>CHAPTER 3 METHODOLOGY</b>	<b>34</b>	
3.1	Introduction	34
3.2	Material Preparation	36
3.2.1	Collection of material	36
3.2.2	Drying process	37
3.2.3	Dry grinding process	37
3.2.4	Wet grinding process	38
3.2.5	Particle size characterization	39
3.3	Nanofluid preparation	41
3.4	Stability measurement	42
3.5	Thermophysical properties measurement	43

3.5.1	Density	43
3.5.2	Viscosity	44
3.5.3	Thermal conductivity	47
3.5.4	Specific heat capacity	48
3.6	Design of experiment (DOE)	50
3.6.1	Definition and purpose	50
3.6.2	Response surface methodology	51
<b>CHAPTER 4 RESULTS AND DISCUSSION</b>		<b>54</b>
4.1	Introduction	54
4.2	Material characterization	54
4.2.1	Size of nanoparticles	54
4.2.2	Bulk density measurement	55
4.3	Nanofluid stability	56
4.4	Thermophysical properties	59
4.4.1	Density	60
4.4.2	Viscosity	62
4.4.3	Thermal conductivity	66
4.4.4	Specific heat capacity	71
4.5	Statistical Analysis	73
4.5.1	Analysis of thermal conductivity	75
4.5.2	Analysis of viscosity	81
4.5.3	Optimization Parameters	87
4.5.4	Optimization Responses	92
<b>CHAPTER 5 CONCLUSION</b>		<b>95</b>

5.1	Introduction	95
5.2	Recommendation	97
<b>REFERENCES</b>		<b>98</b>
<b>APPENDIX A SAMPLE APPENDIX 1</b>		<b>111</b>

## REFERENCES

- A. R., A., Satyanand, A., Arvind, P., & Sarit K., D. (2018). Experimental assessment of the thermo-hydraulic performance of automobile radiator with metallic and non-metallic nanofluids. *Heat Transfer Engineering*, 0(0), 1–42. <https://doi.org/10.1080/01457632.2018.1528055>
- Adewumi, G. A., Inambao, F., Sharifpur, M., & Meyer, J. P. (2018a). Investigation of the Viscosity and Stability of Green Nanofluids from Coconut Fibre Carbon Nanoparticles: Effect of Temperature and Mass Fraction. *International Journal of Applied Engineering Research*, 13(10), 8336–8342. Retrieved from <http://www.ripublication.com>
- Adewumi, G. A., Inambao, F., Sharifpur, M., & Meyer, J. P. (2018b). Investigation of the Viscosity and Stability of Green Nanofluids from Coconut Fibre Carbon Nanoparticles: Effect of Temperature and Mass Fraction. *International Journal of Applied Engineering Research*, 13(10), 8336–8342. Retrieved from <http://www.ripublication.com>
- Ahmadi, M., Elmongy, H., Madrakian, T., & Abdel-Rehim, M. (2017). Nanomaterials as sorbents for sample preparation in bioanalysis: A review. *Analytica Chimica Acta*. <https://doi.org/10.1016/j.aca.2016.11.062>
- Ahmed, S., Ahmad, M., Swami, B. L., & Ikram, S. (2016). A review on plants extract mediated synthesis of silver nanoparticles for antimicrobial applications: A green expertise. *Journal of Advanced Research*. <https://doi.org/10.1016/j.jare.2015.02.007>
- Ahmed, S., Annu, Ikram, S., & Yudha, S. (2016). Biosynthesis of gold nanoparticles: A green approach. *Journal of Photochemistry and Photobiology B: Biology*. <https://doi.org/10.1016/j.jphotobiol.2016.04.034>
- Ajitha, B., Divya, A., & Reddy, P. S. (2013). Impact of Ph on the Properties of Spherical Silver Nanoparticles Capped by PVA. *Advanced Materials Manufacturing & Characterization*, 3(1), 1. <https://doi.org/10.11127/ijammc.2013.02.075>
- AL-SHAMMARY, A. A. G., KOUZANI, A. Z., KAYNAK, A., KHOO, S. Y., NORTON, M., & GATES, W. (2018). Soil Bulk Density Estimation Methods: A Review. *Pedosphere*, 28(4), 581–596. [https://doi.org/10.1016/S1002-0160\(18\)60034-7](https://doi.org/10.1016/S1002-0160(18)60034-7)
- Alawi, O. A., Sidik, N. A. C., Xian, H. W., Kean, T. H., & Kazi, S. N. (2018). Thermal conductivity and viscosity models of metallic oxides nanofluids. *International Journal of Heat and Mass Transfer*, 116, 1314–1325.

<https://doi.org/10.1016/j.ijheatmasstransfer.2017.09.133>

Angayarkanni, S. A., & Philip, J. (2015). Review on thermal properties of nanofluids: Recent developments. *Advances in Colloid and Interface Science*. <https://doi.org/10.1016/j.cis.2015.08.014>

Anoop, K. B., Sundararajan, T., & Das, S. K. (2009). Effect of particle size on the convective heat transfer in nanofluid in the developing region. *International Journal of Heat and Mass Transfer*, 52(9–10), 2189–2195. <https://doi.org/10.1016/j.ijheatmasstransfer.2007.11.063>

Askari, S., Lotfi, R., Rashidi, A. M., Koolivand, H., & Koolivand-Salooki, M. (2016). Rheological and thermophysical properties of ultra-stable kerosene-based Fe<sub>3</sub>O<sub>4</sub>/Graphene nanofluids for energy conservation. *Energy Conversion and Management*, 128, 134–144. <https://doi.org/10.1016/j.enconman.2016.09.037>

Awua, J. T., Ibrahim, J. S., Kwaghger, A., Sharifpur, M., & Meyer, J. P. (2016). Investigation Into Thermal Conductivity of Palm Kernel Fibre Nanofluids With Mixture of Ethylene ... Investigation Into Thermal Conductivity of Palm Kernel Fibre. *12th International Conference on Heat Transfer, Fluid Mechanics and Thermodynamics*, (July), 1719–1725. <https://doi.org/10.1016/j.compscitech.2009.05.004>

Azmi, W. H., Sharma, K. V., Mamat, R., Najafi, G., & Mohamad, M. S. (2016). The enhancement of effective thermal conductivity and effective dynamic viscosity of nanofluids - A review. *Renewable and Sustainable Energy Reviews*. <https://doi.org/10.1016/j.rser.2015.09.081>

Babita, Sharma, S. K., & Gupta, S. M. (2016). Preparation and evaluation of stable nanofluids for heat transfer application: A review. *Experimental Thermal and Fluid Science*. <https://doi.org/10.1016/j.expthermflusci.2016.06.029>

Basnet, P., Inakhunbi Chanu, T., Samanta, D., & Chatterjee, S. (2018). A review on bio-synthesized zinc oxide nanoparticles using plant extracts as reductants and stabilizing agents. *Journal of Photochemistry and Photobiology B: Biology*, 183, 201–221. <https://doi.org/10.1016/j.jphotobiol.2018.04.036>

Batmunkh, M., Tanshen, M. R., Nine, M. J., Myekhlai, M., Choi, H., Chung, H., & Jeong, H. (2014). Thermal conductivity of TiO<sub>2</sub> nanoparticles based aqueous nanofluids with an addition of a modified silver particle. *Industrial and Engineering Chemistry Research*, 53(20), 8445–8451. <https://doi.org/10.1021/ie403712f>

Cabaleiro, D., Gracia-Fernández, C., Legido, J. L., & Lugo, L. (2015). Specific heat of metal oxide nanofluids at high concentrations for heat transfer. *International Journal*

Chandrasekar, M., Suresh, S., & Chandra Bose, A. (2010). Experimental investigations and theoretical determination of thermal conductivity and viscosity of Al<sub>2</sub>O<sub>3</sub>/water nanofluid. *Experimental Thermal and Fluid Science*, 34(2), 210–216. <https://doi.org/10.1016/j.expthermflusci.2009.10.022>

Che Sidik, N. A., Mahmud Jamil, M., Aziz Japar, W. M. A., & Muhammad Adamu, I. (2017). A review on preparation methods, stability and applications of hybrid nanofluids. *Renewable and Sustainable Energy Reviews*. <https://doi.org/10.1016/j.rser.2017.05.221>

Chiam, H. W., Azmi, W. H., Usri, N. A., Mamat, R., & Adam, N. M. (2017). Thermal conductivity and viscosity of Al<sub>2</sub>O<sub>3</sub>nanofluids for different based ratio of water and ethylene glycol mixture. *Experimental Thermal and Fluid Science*, 81, 420–429. <https://doi.org/10.1016/j.expthermflusci.2016.09.013>

Chilingar, G. V., & Haroun, M. (2014). Introduction to Electrokinetics.

Dardan, E., Afrand, M., & Meghdadi Isfahani, A. H. (2016). Effect of suspending hybrid nano-additives on rheological behavior of engine oil and pumping power. *Applied Thermal Engineering*, 109, 524–534. <https://doi.org/10.1016/j.applthermaleng.2016.08.103>

Das, P. K. (2017). A review based on the effect and mechanism of thermal conductivity of normal nanofluids and hybrid nanofluids. *Journal of Molecular Liquids*. <https://doi.org/10.1016/j.molliq.2017.05.071>

Das, S. K., Putra, N., & Roetzel, W. (2003). Pool boiling characteristics of nano-fluids. *International Journal of Heat and Mass Transfer*, 46(5), 851–862. [https://doi.org/10.1016/S0017-9310\(02\)00348-4](https://doi.org/10.1016/S0017-9310(02)00348-4)

Dendisová, M., Jeništová, A., Parchaňská-Kokaislová, A., Matějka, P., Prokopec, V., & Švecová, M. (2018). The use of infrared spectroscopic techniques to characterize nanomaterials and nanostructures: A review. *Analytica Chimica Acta*. <https://doi.org/10.1016/j.aca.2018.05.046>

Devendiran, D. K., & Amirtham, V. A. (2016). A review on preparation, characterization, properties and applications of nanofluids. *Renewable and Sustainable Energy Reviews*. <https://doi.org/10.1016/j.rser.2016.01.055>

Dhinesh Kumar, D., & Valan Arasu, A. (2018). A comprehensive review of preparation,

characterization, properties and stability of hybrid nanofluids. *Renewable and Sustainable Energy Reviews*. <https://doi.org/10.1016/j.rser.2017.05.257>

Esfahani, N. N., Toghraie, D., & Afrand, M. (2018). A new correlation for predicting the thermal conductivity of ZnO–Ag (50%–50%)/water hybrid nanofluid: An experimental study. *Powder Technology*, 323, 367–373. <https://doi.org/10.1016/j.powtec.2017.10.025>

Farbod, M., Ahangarpour, A., & Etemad, S. G. (2015). Stability and thermal conductivity of water-based carbon nanotube nanofluids. *Particuology*, 22, 59–65. <https://doi.org/10.1016/j.partic.2014.07.005>

Fuskele, V., & Sarviya, R. M. (2017). Recent developments in Nanoparticles Synthesis, Preparation and Stability of Nanofluids. In *Materials Today: Proceedings* (Vol. 4, pp. 4049–4060). <https://doi.org/10.1016/j.matpr.2017.02.307>

Gangadevi, R., Vinayagam, B. K., & Senthilraja, S. (2018). Effects of sonication time and temperature on thermal conductivity of CuO/water and Al<sub>2</sub>O<sub>3</sub>/water nanofluids with and without surfactant. In *Materials Today: Proceedings* (Vol. 5, pp. 9004–9011). <https://doi.org/10.1016/j.matpr.2017.12.347>

Ghadimi, A., Saidur, R., & Metselaar, H. S. C. (2011). A review of nanofluid stability properties and characterization in stationary conditions. *International Journal of Heat and Mass Transfer*, 54(17–18), 4051–4068. <https://doi.org/10.1016/j.ijheatmasstransfer.2011.04.014>

Ghanbarpour, M., Bitaraf Haghigi, E., & Khodabandeh, R. (2014). Thermal properties and rheological behavior of water based Al<sub>2</sub>O<sub>3</sub>nanofluid as a heat transfer fluid. *Experimental Thermal and Fluid Science*, 53, 227–235. <https://doi.org/10.1016/j.expthermflusci.2013.12.013>

Gu, B., Hou, B., Lu, Z., Wang, Z., & Chen, S. (2013). Thermal conductivity of nanofluids containing high aspect ratio fillers. *International Journal of Heat and Mass Transfer*, 64, 108–114. <https://doi.org/10.1016/j.ijheatmasstransfer.2013.03.080>

Guardia, P., Batlle-Brugal, B., Roca, A. G., Iglesias, O., Morales, M. P., Serna, C. J., ... Batlle, X. (2007). Surfactant effects in magnetite nanoparticles of controlled size. *Journal of Magnetism and Magnetic Materials*, 316(2 SPEC. ISS.), 756–759. <https://doi.org/10.1016/j.jmmm.2007.03.085>

Gupta, M., Singh, V., Kumar, R., & Said, Z. (2017). A review on thermophysical properties of nanofluids and heat transfer applications. *Renewable and Sustainable Energy Reviews*. <https://doi.org/10.1016/j.rser.2017.02.073>

Gupta, M., Singh, V., Kumar, S., Kumar, S., Dilbaghi, N., & Said, Z. (2018). Up to date review on the synthesis and thermophysical properties of hybrid nanofluids. *Journal of Cleaner Production*. <https://doi.org/10.1016/j.jclepro.2018.04.146>

Haddad, Z., Abid, C., Oztop, H. F., & Mataoui, A. (2014). A review on how the researchers prepare their nanofluids. *International Journal of Thermal Sciences*. <https://doi.org/10.1016/j.ijthermalsci.2013.08.010>

Hamid, K. A., Azmi, W. H., Nabil, M. F., & Mamat, R. (2018). Experimental investigation of nanoparticle mixture ratios on TiO<sub>2</sub>–SiO<sub>2</sub>nanofluids heat transfer performance under turbulent flow. *International Journal of Heat and Mass Transfer*, 118, 617–627. <https://doi.org/10.1016/j.ijheatmasstransfer.2017.11.036>

Hamzah, M. H., Sidik, N. A. C., Ken, T. L., Mamat, R., & Najafi, G. (2017). Factors affecting the performance of hybrid nanofluids: A comprehensive review. *International Journal of Heat and Mass Transfer*. <https://doi.org/10.1016/j.ijheatmasstransfer.2017.07.021>

Huminic, G., & Huminic, A. (2018). Hybrid nanofluids for heat transfer applications – A state-of-the-art review. *International Journal of Heat and Mass Transfer*. <https://doi.org/10.1016/j.ijheatmasstransfer.2018.04.059>

Hussein, A. M., Bakar, R. A., Kadirkama, K., & Sharma, K. V. (2013). Experimental measurement of nanofluids thermal properties. *International Journal of Automotive and Mechanical Engineering*, 7(1), 850–863. <https://doi.org/10.15282/ijame.7.2012.5.0070>

Hwang, Y., Lee, J. K., Lee, C. H., Jung, Y. M., Cheong, S. I., Lee, C. G., ... Jang, S. P. (2007). Stability and thermal conductivity characteristics of nanofluids. *Thermochimica Acta*, 455(1–2), 70–74. <https://doi.org/10.1016/j.tca.2006.11.036>

Issa, R. J. (2016). Effect of Nanoparticles Size and Concentration on Thermal and Rheological Properties of AL<sub>2</sub>O<sub>3</sub>-Water Nanofluids. In *Proceedings of the World Congress on Momentum, Heat and Mass Transfer*. <https://doi.org/10.11159/enfht16.101>

Jacyna, J., Kordalewska, M., & Markuszewski, M. J. (2019). Design of Experiments in metabolomics-related studies: An overview. *Journal of Pharmaceutical and Biomedical Analysis*. <https://doi.org/10.1016/j.jpba.2018.11.027>

Jeanmonod, D. J., Rebecca, & et al. Suzuki, K. (2018). Thermophysical Properties of Metal Oxides Nanofluids. *Intech Open*, 2, 64. <https://doi.org/10.5772/32009>

- Kallamu, U. M., Ibrahim, J. S., Sharifpur, M., & Meyer, J. P. (2016). Experimental Investigation on Viscosity of Nanofluids Prepared From Banana Fibre-Nanoparticles. *12th International Conference on Heat Transfer, Fluid Mechanics and Thermodynamics*, (July), 1713–1718. Retrieved from <https://www.researchgate.net/publication/305392536>
- Khan, I., Saeed, K., & Khan, I. (2017). Nanoparticles: Properties, applications and toxicities. *Arabian Journal of Chemistry*. <https://doi.org/10.1016/j.arabjc.2017.05.011>
- Khodadadi, H., Aghakhani, S., Majd, H., Kalbasi, R., Wongwises, S., & Afrand, M. (2018). A comprehensive review on rheological behavior of mono and hybrid nanofluids: Effective parameters and predictive correlations. *International Journal of Heat and Mass Transfer*. <https://doi.org/10.1016/j.ijheatmasstransfer.2018.07.103>
- Kim, H. J., Lee, S. H., Lee, J. H., & Jang, S. P. (2015). Effect of particle shape on suspension stability and thermal conductivities of water-based bohemite alumina nanofluids. *Energy*, 90, 1290–1297. <https://doi.org/10.1016/j.energy.2015.06.084>
- Ko, S., & Huh, C. (2018). Use of nanoparticles for oil production applications. *Journal of Petroleum Science and Engineering*. <https://doi.org/10.1016/j.petrol.2018.09.051>
- Kole, M., & Dey, T. K. (2010). Viscosity of alumina nanoparticles dispersed in car engine coolant. *Experimental Thermal and Fluid Science*, 34(6), 677–683. <https://doi.org/10.1016/j.expthermflusci.2009.12.009>
- Kumar, N., Sonawane, S. S., & Sonawane, S. H. (2018). Experimental study of thermal conductivity, heat transfer and friction factor of Al<sub>2</sub>O<sub>3</sub>based nanofluid. *International Communications in Heat and Mass Transfer*, 90, 1–10. <https://doi.org/10.1016/j.icheatmasstransfer.2017.10.001>
- Kumar, R., Ha, S. K., Verma, K., & Tiwari, S. K. (2018). Recent progress in selected bio-nanomaterials and their engineering applications: An overview. *Journal of Science: Advanced Materials and Devices*. <https://doi.org/10.1016/j.jsamd.2018.05.003>
- Lee, J. S., Lee, J. W., & Kang, Y. T. (2015). CO<sub>2</sub>absorption/regeneration enhancement in DI water with suspended nanoparticles for energy conversion application. *Applied Energy*, 143, 119–129. <https://doi.org/10.1016/j.apenergy.2015.01.020>
- Leong, K. Y., Ong, H. C., Amer, N. H., Norazrina, M. J., Risby, M. S., & Ku Ahmad, K. Z. (2016). An overview on current application of nanofluids in solar thermal collector and its challenges. *Renewable and Sustainable Energy Reviews*, 53, 1092–1105. <https://doi.org/10.1016/j.rser.2015.09.060>

- Li, H., Wang, L., He, Y., Hu, Y., Zhu, J., & Jiang, B. (2014). Experimental investigation of thermal conductivity and viscosity of ethylene glycol based ZnO nanofluids. *Applied Thermal Engineering*, 88, 363–368. <https://doi.org/10.1016/j.applthermaleng.2014.10.071>
- Li, Y., Zhou, J., Tung, S., Schneider, E., & Xi, S. (2009, December). A review on development of nanofluid preparation and characterization. *Powder Technology*. <https://doi.org/10.1016/j.powtec.2009.07.025>
- Mahbubul, I. M. (2019a). Introduction to Nanofluid. *Preparation, Characterization, Properties and Application of Nanofluid*, 1–13. <https://doi.org/10.1016/B978-0-12-813245-6.00001-0>
- Mahbubul, I. M. (2019b). *Preparation of Nanofluid. Preparation, Characterization, Properties and Application of Nanofluid*. <https://doi.org/10.1016/B978-0-12-813245-6.00002-2>
- Mahbubul, I. M. (2019c). *Thermophysical Properties of Nanofluids. Preparation, Characterization, Properties and Application of Nanofluid* (Vol. c). <https://doi.org/10.1016/B978-0-12-813245-6.00004-6>
- Mahbubul, I. M., Shahrul, I. M., Khaleduzzaman, S. S., Saidur, R., Amalina, M. A., & Turgut, A. (2015). Experimental investigation on effect of ultrasonication duration on colloidal dispersion and thermophysical properties of alumina-water nanofluid. *International Journal of Heat and Mass Transfer*, 88, 73–81. <https://doi.org/10.1016/j.ijheatmasstransfer.2015.04.048>
- Mansour, M. A., Siddiq, S., Gorla, R. S. R., & Rashad, A. M. (2018). Effects of heat source and sink on entropy generation and MHD natural convection of Al<sub>2</sub>O<sub>3</sub>-Cu/water hybrid nanofluid filled with square porous cavity. *Thermal Science and Engineering Progress*, 6, 57–71. <https://doi.org/10.1016/j.tsep.2017.10.014>
- Mehrali, M., Sadeghinezhad, E., Latibari, S., Kazi, S., Mehrali, M., Zubir, M. N. B. M., & Metselaar, H. S. (2014). Investigation of thermal conductivity and rheological properties of nanofluids containing graphene nanoplatelets. *Topology and Its Applications*, 157(14), 2225–2239. <https://doi.org/10.1016/j.topol.2010.06.004>
- Mitra, I., Manna, N., Manna, J. S., & Mitra, M. K. (2014). Synthesis of Chlorophyll Entrapped Red Luminescent Silica Nanoparticles for Bioimaging Application. *Procedia Materials Science*, 6, 770–774. <https://doi.org/10.1016/j.mspro.2014.07.093>
- Moldoveanu, G. M., & Minea, A. A. (2019). Specific heat experimental tests of simple and hybrid oxide-water nanofluids: Proposing new correlation. *Journal of Molecular*

*Liquids*, 279, 299–305. <https://doi.org/10.1016/j.molliq.2019.01.137>

Mukherjee, S. (2013). Preparation and Stability of Nanofluids-A Review. *IOSR Journal of Mechanical and Civil Engineering*, 9(2), 63–69. <https://doi.org/10.9790/1684-0926369>

Nabati Shoghl, S., Jamali, J., & Keshavarz Moraveji, M. (2016). Electrical conductivity, viscosity, and density of different nanofluids: An experimental study. *Experimental Thermal and Fluid Science*, 74, 339–346. <https://doi.org/10.1016/j.expthermflusci.2016.01.004>

Nabil, M. F., Azmi, W. H., Abdul Hamid, K., Mamat, R., & Hagos, F. Y. (2017). An experimental study on the thermal conductivity and dynamic viscosity of TiO<sub>2</sub>-SiO<sub>2</sub>nanofluids in water: Ethylene glycol mixture. *International Communications in Heat and Mass Transfer*, 86, 181–189. <https://doi.org/10.1016/j.icheatmasstransfer.2017.05.024>

Nabil, M. F., Azmi, W. H., Hamid, K. A., Zawawi, N. N. M., Priyandoko, G., & Mamat, R. (2017). Thermo-physical properties of hybrid nanofluids and hybrid nanolubricants: A comprehensive review on performance. *International Communications in Heat and Mass Transfer*, 83, 30–39. <https://doi.org/10.1016/j.icheatmasstransfer.2017.03.008>

Namburu, P. K., Kulkarni, D. P., Misra, D., & Das, D. K. (2007). Viscosity of copper oxide nanoparticles dispersed in ethylene glycol and water mixture. *Experimental Thermal and Fluid Science*, 32(2), 397–402. <https://doi.org/10.1016/j.expthermflusci.2007.05.001>

Nguyen, C. T., Desgranges, F., Galanis, N., Roy, G., Maré, T., Boucher, S., & Angue Mintsa, H. (2008). Viscosity data for Al<sub>2</sub>O<sub>3</sub>-water nanofluid-hysteresis: is heat transfer enhancement using nanofluids reliable? *International Journal of Thermal Sciences*, 47(2), 103–111. <https://doi.org/10.1016/j.ijthermalsci.2007.01.033>

Nguyen, C. T., Desgranges, F., Roy, G., Galanis, N., Maré, T., Boucher, S., & Angue Mintsa, H. (2007). Temperature and particle-size dependent viscosity data for water-based nanofluids - Hysteresis phenomenon. *International Journal of Heat and Fluid Flow*, 28(6), 1492–1506. <https://doi.org/10.1016/j.ijheatfluidflow.2007.02.004>

Nikkam, N., & Toprak, M. S. (2018). Fabrication and thermo-physical characterization of silver nanofluids: An experimental investigation on the effect of base liquid. *International Communications in Heat and Mass Transfer*, 91, 196–200. <https://doi.org/10.1016/j.icheatmasstransfer.2017.12.017>

O'Hanley, H., Buongiorno, J., McKrell, T., & Hu, L. W. (2012). Measurement and model

validation of nanofluid specific heat capacity with differential scanning calorimetry. *Advances in Mechanical Engineering*, 2012. <https://doi.org/10.1155/2012/181079>

Raja, M., Vijayan, R., Dineshkumar, P., & Venkatesan, M. (2016). Review on nanofluids characterization, heat transfer characteristics and applications. *Renewable and Sustainable Energy Reviews*. <https://doi.org/10.1016/j.rser.2016.05.079>

Rawat, R. S. (2015). Dense Plasma Focus - From Alternative Fusion Source to Versatile High Energy Density Plasma Source for Plasma Nanotechnology. In *Journal of Physics: Conference Series* (Vol. 591). <https://doi.org/10.1088/1742-6596/591/1/012021>

Reddy, M. C. S., & Rao, V. V. (2013). Experimental studies on thermal conductivity of blends of ethylene glycol-water-based TiO<sub>2</sub> nanofluids. *International Communications in Heat and Mass Transfer*, 46, 31–36. <https://doi.org/10.1016/j.icheatmasstransfer.2013.05.009>

Rizvi, S. A. A., & Saleh, A. M. (2018, January). Applications of nanoparticle systems in drug delivery technology. *Saudi Pharmaceutical Journal*. Elsevier. <https://doi.org/10.1016/j.jsps.2017.10.012>

Sadeghi, R., Etemad, S. G., Keshavarzi, E., & Haghshenasfard, M. (2015). Investigation of alumina nanofluid stability by UV-vis spectrum. *Microfluidics and Nanofluidics*, 18(5–6), 1023–1030. <https://doi.org/10.1007/s10404-014-1491-y>

Said, Z., & Saidur, R. (2017). Thermophysical Properties of Metal Oxides Nanofluids. *Nanofluid Heat and Mass Transfer in Engineering Problems*. <https://doi.org/10.5772/65610>

Sajid, M. U., & Ali, H. M. (2018). Thermal conductivity of hybrid nanofluids: A critical review. *International Journal of Heat and Mass Transfer*. <https://doi.org/10.1016/j.ijheatmasstransfer.2018.05.021>

Sarviya, R. M., & Fuskele, V. (2017). Review on Thermal Conductivity of Nanofluids. In *Materials Today: Proceedings* (Vol. 4, pp. 4022–4031). <https://doi.org/10.1016/j.matpr.2017.02.304>

Shahrul, I. M., Mahbubul, I. M., Khaleduzzaman, S. S., Saidur, R., & Sabri, M. F. M. (2014). A comparative review on the specific heat of nanofluids for energy perspective. *Renewable and Sustainable Energy Reviews*. <https://doi.org/10.1016/j.rser.2014.05.081>

Shahrul, I. M., Mahbubul, I. M., Sohel, M. R., Elias, M. M., Khaleduzzaman, S. S.,

Saidur, R., & Sadeghipour, S. (2014). Experimental investigation on the thermo-physical properties of Al<sub>2</sub>O<sub>3</sub> nanoparticles suspended in car radiator coolant. *International Communications in Heat and Mass Transfer*, 54, 48–53. <https://doi.org/10.1016/j.icheatmasstransfer.2014.03.005>

Shao, X., Chen, Y., Mo, S., Cheng, Z., & Yin, T. (2015). Dispersion Stability of TiO<sub>2</sub>-H<sub>2</sub>O Nanofluids Containing Mixed Nanotubes and Nanosheets. In *Energy Procedia* (Vol. 75, pp. 2049–2054). <https://doi.org/10.1016/j.egypro.2015.07.282>

Sharifpur, M., Solomon, A. B., Meyer, J. P., Ibrahim, J. S., & Immanuel, B. (2017). Thermal Conductivity and Viscosity of Mango Bark / Water Nanofluids. *13th International Conference on Heat Transfer, Fluid Mechanics and Thermodynamics*, (July).

Shin, D., & Banerjee, D. (2014). Specific heat of nanofluids synthesized by dispersing alumina nanoparticles in alkali salt eutectic. *International Journal of Heat and Mass Transfer*, 74, 210–214. <https://doi.org/10.1016/j.ijheatmasstransfer.2014.02.066>

Simpson, S., Schelfhout, A., Golden, C., & Vafaei, S. (2018). Nanofluid Thermal Conductivity and Effective Parameters. *Applied Sciences*, 9(1), 87. <https://doi.org/10.3390/app9010087>

Singh, P., Kim, Y. J., Zhang, D., & Yang, D. C. (2016). Biological Synthesis of Nanoparticles from Plants and Microorganisms. *Trends in Biotechnology*. <https://doi.org/10.1016/j.tibtech.2016.02.006>

Solomon, A. B., Sharifpur, M., Meyer, J. P., Ibrahim, J. S., & Immanuel, B. (2017). Convection heat transfer with water based mango bark nanofluids. In *13th International Conference on Heat Transfer, Fluid Mechanics and Thermodynamics*. Retrieved from <https://www.researchgate.net/publication/318562871>

Spernarth, L., & Magdassi, S. (2007). Preparation of ethyl cellulose nanoparticles from nano-emulsion obtained by inversion at constant temperature. *Micro & Nano Letters*, 2(4), 90–95. <https://doi.org/10.1049/mnl>

Suganthi, K. S., & Rajan, K. S. (2017). Metal oxide nanofluids: Review of formulation, thermo-physical properties, mechanisms, and heat transfer performance. *Renewable and Sustainable Energy Reviews*. <https://doi.org/10.1016/j.rser.2017.03.043>

Sundar, L. S., Sharma, K. V, Singh, M. K., & Sousa, A. C. M. (2017). Hybrid nanofluids preparation, thermal properties, heat transfer and friction factor – A review. *Renewable and Sustainable Energy Reviews*. <https://doi.org/10.1016/j.rser.2016.09.108>

Sundar, L. S., Singh, M. K., & Sousa, A. C. M. (2014). Enhanced heat transfer and friction factor of MWCNT-Fe<sub>3</sub>O<sub>4</sub>/water hybrid nanofluids. *International Communications in Heat and Mass Transfer*, 52, 73–83. <https://doi.org/10.1016/j.icheatmasstransfer.2014.01.012>

Syam Sundar, L., Singh, M. K., & Sousa, A. C. M. (2013). Investigation of thermal conductivity and viscosity of Fe<sub>3</sub>O<sub>4</sub>nanofluid for heat transfer applications. *International Communications in Heat and Mass Transfer*, 44, 7–14. <https://doi.org/10.1016/j.icheatmasstransfer.2013.02.014>

Syam Sundar, L., Venkata Ramana, E., Singh, M. K., & Sousa, A. C. M. (2014). Thermal conductivity and viscosity of stabilized ethylene glycol and water mixture Al<sub>2</sub>O<sub>3</sub>nanofluids for heat transfer applications: An experimental study. *International Communications in Heat and Mass Transfer*, 56, 86–95. <https://doi.org/10.1016/j.icheatmasstransfer.2014.06.009>

Thakkar, K. N., Mhatre, S. S., & Parikh, R. Y. (2010). Biological synthesis of metallic nanoparticles. *Nanomedicine: Nanotechnology, Biology, and Medicine*. <https://doi.org/10.1016/j.nano.2009.07.002>

Thota, S., & Crans, D. C. (2018). *Metal Nanoparticles Synthesis and Applications in Pharmaceutical Sciences*. Retrieved from [https://books.google.com.my/books?hl=en&lr=&id=OYZFDwAAQBAJ&oi=fnd&pg=PA15&dq=bio+nanoparticles+plants+preparation&ots=tpQVaXGPpn&sig=379vA6mC1cMFmLKxll9-ImOoz0I&redir\\_esc=y#v=onepage&q=bio nanoparticles plants preparation&f=false](https://books.google.com.my/books?hl=en&lr=&id=OYZFDwAAQBAJ&oi=fnd&pg=PA15&dq=bio+nanoparticles+plants+preparation&ots=tpQVaXGPpn&sig=379vA6mC1cMFmLKxll9-ImOoz0I&redir_esc=y#v=onepage&q=bio nanoparticles plants preparation&f=false)

Timofeeva, E. V., Yu, W., France, D. M., Singh, D., & Routbort, J. L. (2011). Base fluid and temperature effects on the heat transfer characteristics of SiC in ethylene glycol/H<sub>2</sub>O and H<sub>2</sub>O nanofluids. *Journal of Applied Physics*, 109(1). <https://doi.org/10.1063/1.3524274>

Vajjha, R. S., & Das, D. K. (2009). Specific Heat Measurement of Three Nanofluids and Development of New Correlations. *Journal of Heat Transfer*, 131(7), 071601. <https://doi.org/10.1115/1.3090813>

Vajjha, R. S., & Das, D. K. (2012). A review and analysis on influence of temperature and concentration of nanofluids on thermophysical properties, heat transfer and pumping power. *International Journal of Heat and Mass Transfer*. <https://doi.org/10.1016/j.ijheatmasstransfer.2012.03.048>

Wang, B. X., Zhou, L. P., Peng, X. F., Du, X. Z., & Yang, Y. P. (2010). On the specific heat capacity of CuO nanofluid. *Advances in Mechanical Engineering*, 2010. <https://doi.org/10.1155/2010/172085>

- Wang, X. Q., & Mujumdar, A. S. (2007). Heat transfer characteristics of nanofluids: a review. *International Journal of Thermal Sciences*. <https://doi.org/10.1016/j.ijthermalsci.2006.06.010>
- Wong, K. V, & Leon, O. De. (2014). Applications of Nanofluids : Current and Future Applications of Nanofluids : Current and Future, (January 2010). <https://doi.org/10.1155/2010/519659>
- Xie, H., Yu, W., & Chen, W. (2010). MgO nanofluids: Higher thermal conductivity and lower viscosity among ethylene glycol-based nanofluids containing oxide nanoparticles. *Journal of Experimental Nanoscience*, 5(5), 463–472. <https://doi.org/10.1080/17458081003628949>
- Yagnem, A. R., & Venkatachalamathy, S. (2019). Heat transfer enhancement studies in pool boiling using hybrid nanofluids. *Thermochimica Acta*, 672, 93–100. <https://doi.org/10.1016/j.tca.2018.11.014>
- Yang, D., Sun, B., Li, H., & Fan, X. (2015). Experimental study on the heat transfer and flow characteristics of nanorefrigerants inside a corrugated tube. *International Journal of Refrigeration*, 56, 213–223. <https://doi.org/10.1016/j.ijrefrig.2015.04.011>
- Yang, L., Xu, J., Du, K., & Zhang, X. (2017). Recent developments on viscosity and thermal conductivity of nanofluids. *Powder Technology*. <https://doi.org/10.1016/j.powtec.2017.04.061>
- Yang, Y., Zhang, Z. G., Grulke, E. A., Anderson, W. B., & Wu, G. (2005). Heat transfer properties of nanoparticle-in-fluid dispersions (nanofluids) in laminar flow. *International Journal of Heat and Mass Transfer*, 48(6), 1107–1116. <https://doi.org/10.1016/j.ijheatmasstransfer.2004.09.038>
- Yarmand, H., Gharehkhani, S., Ahmadi, G., Shirazi, S. F. S., Baradaran, S., Montazer, E., ... Dahari, M. (2015). Graphene nanoplatelets-silver hybrid nanofluids for enhanced heat transfer. *Energy Conversion and Management*, 100, 419–428. <https://doi.org/10.1016/j.enconman.2015.05.023>
- Yarmand, H., Gharehkhani, S., Shirazi, S. F. S., Amiri, A., Alehashem, M. S., Dahari, M., & Kazi, S. N. (2016). Experimental investigation of thermo-physical properties, convective heat transfer and pressure drop of functionalized graphene nanoplatelets aqueous nanofluid in a square heated pipe. *Energy Conversion and Management*, 114, 38–49. <https://doi.org/10.1016/j.enconman.2016.02.008>
- Yarmand, H., Gharehkhani, S., Shirazi, S. F. S., Amiri, A., Montazer, E., Arzani, H. K., ... Kazi, S. N. (2016). Nanofluid based on activated hybrid of biomass

carbon/graphene oxide: Synthesis, thermo-physical and electrical properties. *International Communications in Heat and Mass Transfer*, 72, 10–15.  
<https://doi.org/10.1055/s-0037-1607363>

Yarmand, H., Gharekhani, S., Shirazi, S. F. S., Goodarzi, M., Amiri, A., Sarsam, W. S., ... Kazi, S. N. (2016a). Study of synthesis, stability and thermo-physical properties of graphene nanoplatelet/platinum hybrid nanofluid. *International Communications in Heat and Mass Transfer*, 77, 15–21.  
<https://doi.org/10.1016/j.icheatmasstransfer.2016.07.010>

Yarmand, H., Gharekhani, S., Shirazi, S. F. S., Goodarzi, M., Amiri, A., Sarsam, W. S., ... Kazi, S. N. (2016b). Study of synthesis, stability and thermo-physical properties of graphene nanoplatelet/platinum hybrid nanofluid. *International Communications in Heat and Mass Transfer*, 77, 15–21.  
<https://doi.org/10.1016/j.icheatmasstransfer.2016.07.010>