

## Carbon Quantum Dots for Sustainable Energy and Optoelectronics

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# Carbon Quantum Dots for Sustainable Energy and Optoelectronics

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### Preface

In congruence with all progress made by human society, the thrust on natural resources has escalated incessantly, which has had a detrimental impact on the health of ecosystems and the well-being of people. Hence, striking a balance between progressive industrialization led economic development and consumption of natural resources is the only way forward for the sustainability of the evolution of society. Sustainable development is defined by the United Nations as the development of present society keeping in view the generations to come. As natural resources are limited, they should be used judiciously and optimally to ensure that there is enough left for future generations as well, without affecting the present quality of life. A sustainable society must thrive to be socially responsible, technologically accessible, and economically feasible keeping in view environmental protection and dynamic equilibrium between human and natural ecosystems. The main pillars of sustainable development are energy, water, and health care. The United Nations has declared them as the goals in the United Nations Sustainable Development Goals SDG7 and SDG6 to ensure access to affordable, clean, reliable, sustainable, and modern energy and to ensure availability and sustainability of clean water and sanitation to all without affecting the environment. Scientific community should focus their research toward attaining these goals. Nanotechnology, a recently developed innovative technology dealing with the science and technology in a nano dimension, is established as a promising tool for achieving these goals. Nanotechnology has the potential to fulfill the overwhelming demand for energy and basic commodities and advancing technology without affecting our environment, climate, and natural resources. The global sustainability challenges our world faces today can be solved by nanotechnology as an environmentally acceptable technique. The main components of nanotechnology in the battle are the nanomaterials and quantum dots. Quantum dots, few nanometers in size, are particles where quantum mechanics are predominant, with the associated quantum mechanical waves confined in nano-dimensions and generating size-dependent discrete energy levels. Generally, in a quantum dot, the energy gap between the conduction band and the valence band or the gap between the HOMO and LUMO is dependent on the particle size. The electronic waves associated with the free electrons on the particles are confined within the boundary of the particle (dimension of the particle) and the energy associated with them is quantized according to the size of the particle. So, the optical and electronic properties of the quantum dots differ largely from their bulk counterparts. The quantum dots have the properties lying somewhere between the bulk and the atom/molecules, and they vary with size and shape. Now the carbon quantum dots (CODs) have emerged as a game changer among different quantum dots and other allotropes of nanocarbon because of simple and sustainable fabrication methods involved. There are different types of allotropes of nanocarbon such as carbon nanotube, buckminsterfullerene, graphene and nanodiamond used in nano-engineering facilitating sustainable development. Carbon, the group 14 member of the periodic table, has a very interesting electronic structure and multiple valance and coordination numbers. Because of different oxidation numbers and catenation properties of carbon, there exist a large variety of allotropes with orbital hybridization along with the structure, governing their properties. Among the nanocarbon allotropes, CQDs are attracting a good deal of research interest because of their ease of synthesis and versatile applications. The CQDs can be synthesized from carbon-containing materials, mainly biomaterials, by a simple chemical reduction process. The simple technique for surface passivation and functionalization adds to the host of characteristics of CQDs for applications in different fields for sustainable development.

This book solely focuses on the different aspects of CQDs facilitating sustainable development of our society. First, this book discusses the structure-property relationship of CQDs in optical domains in detail. As the photophysical properties of CQDs are the most interesting and studied ones, we focused on understanding the photophysical properties and their origin. This book also discusses the theoretical modeling of the CQDs from a basic to an advanced level. The synthesis of CQDs is more beneficial compared to other nanomaterials, especially carbon nanomaterials like CNT and graphene, as it does not require sophisticated instrumentation and technology. A facile and cost-effective synthesis method for CQDs makes them very popular among researchers. The third chapter of the book delivers the details of the synthesis method of CQDs. Following the synthesis, the physical properties and different characterization techniques of CQDs are covered. As the properties of the CQDs are predominantly controlled by the surface states of the CQDs, this book pays special attention to the surface functionalization of CQDs in the next chapter. Most of the fabrication methods of CQDs are sustainable ones, but if we want to highlight the role of CQDs in sustainable development, it is mainly derived from the different application aspects of the CQDs. We focus on the application of CQDs in energy harvesting, energy storage, and wastewater treatment to biosensing in other chapters. Biomedical applications of CQDs ranging from bioimaging to theranostics are covered in subsequent chapters. The magnetic applications of CQDs and composites of CQDs are also discussed. Finally, the CQD-based optical and electronic nanodevices are discussed with a special focus on terahertz applications and single electron transistor applications. Another form of carbon nanoparticle, nano-diamond, is explored for photonic and biomedical applications. The book concludes with a summary of recent advancements and future prospects of CQDs for sustainable applications.

> Sudip Kumar Batabyal Basudev Pradhan Kallol Mohanta Rama Ranjan Bhattacharjee Amit Banerjee

### Contents

List Prei		ontributo	ors	xiii xix
1	Pho	tophysic	al properties of carbon quantum dots	1
	Tan	oy Dutta,	, Oendrila Chatterjee, Barsha Chakraborty and	
	Apu	rba Lal I	Koner	
	1.1	Introdu	ction	1
	1.2	Optical	absorption properties of carbon quantum dots	2
	1.3	Factors	influencing the photoluminescence properties of	
			quantum dots	4
			Quantum confinement effect	4
			Doping nonmetallic heteroatoms	5
		1.3.3	Local heterogeneity originated from heteroatom-mediated	
			surface defects	8
		1.3.4	8	8
			Red edge effect	9
		1.3.6		11
			Aggregation-induced emission in carbon quantum dots	12
			Förster resonance energy transfer	20
			Photoinduced electron transfer	22
			Electroluminescence of carbon dots	23
	1.4		sions and future aspect	25
	Refe	erences		26
2	The	physica	l and chemical properties of carbon dots via	
	com	putation	nal modeling	29
	Arup	o Chakra	lborty	
	2.1	Introdu	ction	29
	2.2	Differe	nt carbon dots	29
	2.3	Compu	tational methods applied to study the properties of	
		carbon	dots	31
	2.4	Theore	tical studies of different properties of carbon quantum dots	32
		2.4.1	Electronic structure	32
			Optical properties	33
		2.4.3	Electrocatalytic properties	35
		2.4.4	Transport properties	36
		2.4.5	Kondo effect in carbon quantum dots	36

	2.5	Summ	ary and outlook	37
		erences	-	38
_	~			• •
3			f carbon quantum dots	39
			n, Lopamudra Bhattacharjee and Rama Ranjan Bhattacharjee	•
	3.1	Introd		39
	2.2		Carbon quantum dots	39
	3.2		techniques for carbon quantum dot preparation	42
		3.2.1	T T T T T T T T T T T T T T T T T T T	42
	2.2	3.2.2		48
	3.3	Conclu	usion	52
		erences		52
	Furt	her read	ling	53
4	Cha	racteri	zation and physical properties of carbon quantum dots	55
	Suja	tha D.,	Pardhasaradhi Nandigana, P. Sriram and Subhendu K. Panda	
	4.1	Introd	uction	55
		4.1.1	Carbon quantum dots	56
		4.1.2	Structure of carbon quantum dots	59
		4.1.3	Types	60
	4.2	Physic	cal properties	62
		4.2.1	Physiochemical properties (catalytic)	64
			Optical properties	66
		4.2.3	Photoinduced electron transfer	72
		4.2.4	Biological properties	72
	4.3	Charae	cterization	73
		4.3.1	Structural characterization	73
		4.3.2	Photophysical analysis	80
		4.3.3	Stability of carbon quantum dots	85
	4.4	Conclu	usions	85
	Refe	erences		86
5	Surf	face eng	gineering of carbon quantum dots	91
	Ank	ita Saha	a, Lopamudra Bhattacharjee and Rama Ranjan Bhattacharjee	
	5.1	Introd	uction	91
		5.1.1	Carbon nanotube versus carbon quantum dots	91
		5.1.2	Fundamentals of surface engineering in carbogenic allotropes	93
	5.2	Metho	odology	93
		5.2.1	Hydrothermal carbonization	93
		5.2.2	Microwave-assisted pyrolysis	96
		5.2.3	Sol-gel reaction	99
		5.2.4	Condensation reaction	100
		5.2.5	Oxidation-polymerization reaction	101
	5.3	Conclu	usion	102
	Refe	erences		102

6	Pho	todetector applications of carbon and graphene quantum dots	105				
	Suvi	Suvra Prakash Mondal and Tanmoy Majumder					
	6.1	Introduction	105				
	6.2	Synthesis of carbon quantum dots and graphene quantum dots	106				
		6.2.1 Top-down synthesis process	106				
		6.2.2 Bottom-up synthesis process	107				
	6.3	Optical absorption, emission, and electrical properties	108				
	6.4	Optoelectronics applications of carbon quantum dots and					
		graphene quantum dots	112				
	6.5	Photodetector applications of carbon quantum dots and					
		graphene quantum dots	113				
		6.5.1 FET-based photodetectors using carbon quantum dots and	1				
		graphene quantum dots	114				
		6.5.2 Carbon quantum dots or graphene quantum dots-sensitize	d				
		nanomaterial-based photodetectors	120				
		6.5.3 Polymer nanocomposite-based photodetectors	125				
	6.6	Conclusions	128				
	Refe	erences	128				
7	Pho	tovoltaic application of carbon quantum dots	135				
	Pras	shant Kumar, Arup Mahapatra, Sandeep Kumar and					
	Basi	udev Pradhan					
	7.1	Introduction	135				
	7.2	Carbon quantum dots in dye-sensitized solar cells	136				
		7.2.1 Carbon quantum dots as sensitizer	137				
		7.2.2 Carbon quantum dots as counter electrode	142				
	7.3	1 0	144				
	7.4	1	148				
	7.5	1 1	150				
	7.6	1	154				
	7.7		154				
	Ack	nowledgments	155				
	Refe	erences	155				
8	-	ht-emitting diode application of carbon quantum dots	159				
		teza Sasani Ghamsari and Ashkan Momeni Bidzard					
	8.1	Introduction	159				
	8.2	Synthesis methods of functionalized carbon quantum dots	159				
		8.2.1 Electrochemical synthesis	160				
		8.2.2 Arc discharge	160				
		8.2.3 Pulsed laser ablation/passivation technique	161				
		8.2.4 Microwave-assisted synthesis	161				
		8.2.5 Hydrothermal and solvothermal synthesis	163				
	8.3	Optical properties of carbon quantum dots	163				
		8.3.1 Optical absorption	164				

		8.3.2 I	Photoluminescence emissions from ultraviolet to	
		1	near-infrared regions	164
		8.3.3 I	Electroluminescence	168
	8.4	Carbon	quantum dots device applications	169
		8.4.1 I	Light-emitting diodes	169
		8.4.2 0	Optical gain and lasing	174
	8.5	Summar	ry	175
	Refe	rences		176
9	Nan	oelectror	nic applications of carbon quantum dots	183
			ramani, Soumyo Chatterjee and Kallol Mohanta	
	9.1	General	introduction	183
	9.2	Memory	/ devices	185
		9.2.1	Classifications of memory devices	185
			Random access memory	187
	9.3	Transist	ors	193
		9.3.1 1	Basics of transistor	193
		9.3.2	Carbon quantum dots used in transistor applications	194
	9.4			197
	9.5	Carbon	quantum dot laser	200
	Refe	rence		200
10	Carl	oon quar	ntum dot-based nanosensors	205
	Anki	ta Saha,	Lopamudra Bhattacharjee and Rama Ranjan Bhattacharje	e
	10.1	Introdu	action to nanosensors	205
	10.2	Chemi	cal sensing	206
		10.2.1	Fluorescence-based chemical sensing	206
		10.2.2	Chemical sensors: nanoparticles as	
			superior components	208
		10.2.3	CQDs: fluorescent sensor material	208
		10.2.4	pH sensor	212
		10.2.5		
			surrounding medium	216
		10.2.6	Doped CQDs in sensors: metal ion detection	216
		10.2.7	Gas sensing with conducting carbon dots	218
		10.2.8	A VOC sensor based on CQDs	221
	10.3	Conclu	ision	223
	Refe	rences		223
11			: biomedical applications	225
			Madhavan, Ranjita Ghosh Moulick and Jaydeep Bhattacha	•
	11.1		n dots: structure and functionalization	225
	11.2	•	thesis of carbon dots	226
	11.3		aging applications of carbon dots	226
		11.3.1	Carbon dots: optical properties	227

	11.4	Biomedical applications of carbon dots	227
		11.4.1 Drug delivery	227
		11.4.2 Crossing blood-brain barrier	229
		11.4.3 Gene delivery	230
	11.5	Biosensing applications using carbon dots	230
		Future scope and challenges	232
	Refere		233
12		aging applications of carbon quantum dots	239
		sha Kumari, Jaydeep Bhattacharya and Ranjita Ghosh Moulick	
	12.1	Introduction	239
	12.2	Development of various bioimaging modalities	240
	12.3	1 000	241
	12.4	6 6 6	242
	12.5	1	243
	12.6	5	244
		12.6.1 Chemical ablation	244
		12.6.2 Electrochemical method	244
		12.6.3 Laser ablation	244
		12.6.4 Arc Discharge method	245
		12.6.5 Hydrothermal method	245
		12.6.6 Microwave irradiation	245
		12.6.7 Pyrolysis method	245
	12.7	Surface activation	245
		12.7.1 Surface passivation	246
		12.7.2 Surface functionalization	246
		12.7.3 Doping	247
	12.8	1 1	247
		12.8.1 Fluorescence	247
		12.8.2 Quantum yield	248
	12.9	6 6	248
		12.9.1 In vitro imaging	250
		12.9.2 In vivo imaging	251
		12.9.3 Single-molecule imaging	251
	12.10	Conclusion	254
	Refere	nces	254
13		catalytic applications of carbon quantum dots for	262
		water treatment	263
		Pabiatul Ramzilah P. Remli, A <mark>zrina Abd Azi</mark> z, Lan Ching Sim, si Uddin Manin and Kah Han Loong	
		ij Uddin Monir and Kah Hon Leong	262
	13.1	Overview on advanced oxidation process and photocatalysis	263
	13.2	Mechanism of photocatalysis	266
	13.3	Photocatalysts material	268

13.4 Binary metal oxides

268

	13.5	Metal sulfides	270
	13.6	Fundamentals of carbon quantum dots	271
	13.7	-	274
		13.7.1 Broaden the optical absorption range of photocatalyst	274
		13.7.2 Improved charge separation and electron transfer	276
		13.7.3 Allocate additional surface for adsorption and reaction	277
	13.8	Synthesis route of carbon quantum dots	277
		13.8.1 Top-down method	277
		13.8.2 Bottom-up method	278
	13.9	Hydrothermal treatment of carbon quantum dots	278
	13.10	Watermelon rinds potential as carbon precursor	283
	13.11	Application of carbon quantum dots in photocatalysis	283
		13.11.1 Application of carbon quantum dots-based	
		composite in water purification	283
	Refer	ences	285
14		ent prospects of carbon-based nanodots in photocatalytic CO <sub>2</sub>	
		ersion	295
		unt P. Sahu, Christabel Adjah-Tetteh, Nagapradeep Nidamanuri,	
		t K. Sonkar, Erin U. Antia, Tam Tran, Guanguang Xia, Yudong Wang	
		Simon, Manas Ranjan Gartia, Supratik Mukhopadhyay, Yu Wang an	ıd
		Dong Zhou	
		Introduction	295
	14.2	Synthetic approaches and optical properties of carbon quantum dots	
		14.2.1 Carbon dots and graphene quantum dots: an overview	299
	14.3	Carbon-based quantum dots in $CO_2$ photoconversion	310
		14.3.1 Photocatalytic $CO_2$ reduction	310
		14.3.2 Photophysical characteristics and CO <sub>2</sub> photoconversion	
		with carbon-based catalysts	314
	14.4	Concluding remarks	326
		owledgments	334
	Refer	ences	334
15	Carb	on quantum dots and its composites for electrochemical	
		y storage applications	341
		aris Caroline and Sudip K. Batabyal	
		Introduction	341
		Fundamentals of supercapacitors and batteries	342
		15.2.1 Fundamentals of supercapacitors	342
		15.2.2 Fundamentals of batteries	345
	15.3	Desired properties of carbon quantum dots for charge storage	
		applications	349
		15.3.1 Structural properties	349
		15.3.2 Electrical properties	350
		15.3.3 Optical properties	350

	15.4	Carbon quantum dots for supercapacitors	351
		15.4.1 Carbon quantum dots—inorganic hybrid for	
		supercapacitors	353
		15.4.2 Carbon quantum dots—organic hybrid supercapacitors	355
		15.4.3 Graphene quantum dots	360
	15.5	Carbon quantum dots for batteries	361
		15.5.1 Carbon quantum dots in lithium-ion and sodium-ion	
		batteries	362
		15.5.2 Carbon quantum dots in potassium-ion batteries	364
		15.5.3 Carbon quantum dots in lithium-sulfur batteries	364
		15.5.4 Carbon quantum dots in zinc-ion batteries	365
	Refer	ences	366
16	Magi	netic and nanophotonics applications of carbon quantum dots	377
	Ravi .	P.N. Tripathi, Vidyadhar Singh, Bharat Kumar Gupta and	
	Nikhi	l Kumar	
	16.1	Introduction	377
	16.2	Applications	378
		16.2.1 Magnetic applications	378
		16.2.2 Nanophotonic applications and single-photon emission	382
	16.3	Summary and future perspectives	390
	Ackn	owledgments	392
	Refer	ences	392
17	Carb	on quantum dots: An overview and potential applications in	
	terah	ertz domain	397
	Suran	ijana Banerjee	
	17.1	Introduction	397
	17.2	Characteristic lengths	400
	17.3	Quantum dot	402
		17.3.1 Density of states of electrons in quantum dots	403
	17.4	Fabrication techniques of quantum dots	404
		17.4.1 Quantum dots based on II–VI compound semiconductors	405
		17.4.2 Self-assembled quantum dots	406
	17.5	Optical properties of quantum dots	407
		17.5.1 Optical properties of indirect gap nanocrystal	409
	17.6	Applications of carbon quantum dot in the biomedical field	410
		17.6.1 Optical imaging	410
		17.6.2 Photoacoustic imaging	413
		17.6.3 Drug delivery	413
		17.6.4 Crossing blood-brain barrier	413
		17.6.5 Gene delivery	414
	17.7	Carbon nanostructures in terahertz domain	414
		17.7.1 Terahertz time-domain spectroscopy for generation of	
		coherent radiation	414

		17.7.2	Time-resolved spectroscopy and terahertz conductivity in	
			carbon nanostructures	416
	17.8	Conclus	sion and future prospect	416
	Refer	rences		417
18			based single-electron transistors as electrometer	423
	Sourc	ıv Mitra		
	18.1	Theory		423
		18.1.1	Introduction to single-electron transistor	423
		18.1.2	Origin of coulomb blockade oscillation	423
	18.2		ation: single-electron transistor as an electrometer	426
		18.2.1	Measuring inverse compressibility	426
		18.2.2	Experimental realization of a single-electron transistor electrometer: comparing aluminum single-electron	
			transistor to carbon nanotube single-electron transistor	429
	18.3	Davian	ring published work	434
	16.5	18.3.1		434
		18.3.1	Application of Al-based single-electron transistor Application of carbon nanotube-based single-electron	434
		16.3.2	transistor	442
	18.4	Conclu		452
		ences	sion	453
	Kelei	chees		455
19			ds for advanced photonic and biomedical applications	455
			al, Nikhil Dole, Aditya Banerjee and Amit Banerjee	455
	19.1		ction to nanodiamond photonics	455
		19.1.1	Optical emission from diamond	455
	10.2	19.1.2	ND photonic applications	457
	19.2		r biomedical applications	459
		19.2.1	Cancer therapy applications	459
	10.2	19.2.2	Biomedical imaging applications	462
	19.3	Conclus		466
		owledgm	ients	467
	Refer	rences		467
20			ectives of carbon quantum dots	473
		•	e, Sudip K. Batabyal, Basudev Pradhan,	
			ta and Rama Ranjan Bhattacharjee	
	20.1	Introdu		473
	20.2	Future	perspectives of CQDs	474
		20.2.1	Luminescent doped/co-doped CQDs for optical sensing	474
	20.3	Conclus		477
	Ackn	owledgm	nents	477
	Refer	ences		478

#### Index

481

## Photocatalytic applications of carbon quantum dots for wastewater treatment



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# 13.1 Overview on advanced oxidation process and photocatalysis

The excessive discharge of industrial effluent, worldwide production, and utilization of chemical products, as well as expanding world population contributes significantly to the increasing accumulation of bio recalcitrant organic pollutants in the environment [1]. In developing countries, this unpleasant trend is widespread due to the improper enforcement of environmental regulations and monitoring frameworks. A proportion of these organic pollutants remains unregulated and causes serious deterioration of the freshwater ecosystem. Presently, a large amount of various chemical pollutants containing wastewater are produced from domestic and industrial activities, which eventually pollute the environment [2,3]. Fresh, uncontaminated, and enough sanitary measures remain critical tohuman health and socioeconomic sustainability as these two are becoming endangered commodity at present [1,4,5]. In accordance with solving the present water crisis globally and acquiring more economic gain, an alternative for new water treatment technology that can completely remove organic pollutants is henceforth significant and necessary [6]. Owing to the expanding worldwide concern for environmental protection, advanced oxidation technology (AOT) pointed out the prominent role of a special class of oxidation technology defined as advanced oxidation process (AOPs) was developed. To date, studies have shown that AOPs still upheld as one of the favorable and reasonable methods for treating the water and wastewater to remove the contaminants [7-9]. AOPs stand out as one of the most environmentally friendly techniques used to remove bio recalcitrant organic pollutants that are not easily treatable by existing conventional treatment technologies due to their chemical stability [10]. The advantages and disadvantages of the existing water treatment method and photocatalytic system are summarized in Table 13.1.

AOPs allude to a set of chemical treatment procedures for removing organic pollutants in water and wastewater by oxidation. AOPs were first proposed for portable water treatment in the 1980s [9,11,12], which were then defined as oxidation processes involving the generation of hydroxyl radicals (OH-) in adequate amounts for water purification. The main mechanism of the AOPs function is the generation of highly reactive free radicals. AOPs include two phases of oxidation process: formation of strong oxidants (e.g., hydroxyl radicals) and the reaction of these oxidants with organic contaminants in water [13–15]. Table 13.2 depicts the oxidants used in different wastewater treatment techniques with corresponding oxidation potential values [16]. Among o them, OH- has stronger oxidation power than normally used oxidants and decomposes the organic compounds into moderately harmless compounds, such as  $CO_2$ ,  $H_2O$ , or HCI.

To measure the effectiveness of the treatment, it is necessary to understand the selected type of AOPs, physical and chemical properties of pollutants, and operating parameters of the process. A variety of techniques were classified under the broad definition of AOPs. A list of possible techniques offered by AOPs are given in Fig. 13.1

Photocatalysis was included in the family of AOP, which enlisted many advantages and can likely provide solutions for many environmental problems faced by the modern world. This is because photocatalysis allocates a simple way of utilizing

Water treatment technology	Advantages	Disadvantages
Biological	High reliability High load operation can be processed	Difficulty in securing stable process High level of sludge Operating management requires expertise
Coagulation/ precipitation Fenton	High efficiency of processing Low sites Wide coverage Treatment process is simple and easy to manage Effective colored discoloration of wastewater	Excessive sludge produced Difficult to maintain High operating cost over the use of the Fenton's reagent Removal of the equipment needs iron salts
Photocatalytic advanced oxidation	Low operational and installation cost No sludge produces Possible for nonbiodegradable wastewater treatment	Limited lamp life when UV lamp is used Limitation on photocatalyst recovery facility

 Table 13.1 The advantages and disadvantages of existing water treatment technologies.

Source: Adapted from Lim, S.Y., Shen, W., & Gao, Z. (2015). Carbon quantum dots and their applications. Chemical Society Reviews, 44(1), 362–381. https://doi.org/10.1039/c4cs00269e.

summarizes some of the CQDs-TiO<sub>2</sub> composite that has been conducted as photocatalyst for the degradation variety of target pollutants [29].

As reported in Table 13.6, Sun et al. outlined the fabrication of CQDs-TiO<sub>2</sub> nanotubes photocatalyst with improved visible light absorption and photoelectrochemical response [70]. The result shows that the prepared composite of CQDs-TiO<sub>2</sub> nanotubes exhibited higher degradation efficiency than TiO<sub>2</sub> nanotubes arrays by 14% in 100 min under given experimental condition (MB = 15 mL, 5 mg/L). From this study, it was clear that CQDs-TiO<sub>2</sub> nanotubes composite enhanced photocatalytic activity when illuminated with visible light compared to TiO<sub>2</sub> nanotubes. Subsequently, in 2014, Yu and coworkers reported that CQDs-TiO<sub>2</sub> nanosheets (CQDs-TNS) and CQDs-P25 composite exposed an enhancement in photocatalytic activities, especially CQDs-TNS compared to CQDs-P25. Besides, an increasing amount of CQDs solution from 2.5 to 10 mL increases the degradation efficiency gradually from 27.2% to 95.4%, which indicates that utilization of a suitable amount of CQDs can effectively improve the visible light photocatalytic activity of TNS for RhB degradation [38].

Jun et al. attributed the improvement of the degrading efficiency of CQDs-TiO<sub>2</sub> powder, which is notably higher than the controlled pure TiO<sub>2</sub>. At the point when the volume of CQDs utilized was 10 mL, catalytic activity is most noteworthy, up to 90%, which is almost 3.6 times higher than that of pure TiO<sub>2</sub>. It is fascinating that, with the increase in CQDs from 5 to 10 mL, the catalytic activity of the CQDs-TiO<sub>2</sub> increased drastically due to improved absorbance of visible light and increased separation efficiency of photogenerated charge carriers. However, further increment of CQD content to 15 mL leads to an apparent decrease in photocatalytic performance due to the distribution of CQDs on the surface of TiO<sub>2</sub> [40].

Even though the reported studies successfully improved the visible light photocatalytic efficiency, the research gap still involves combining TiO<sub>2</sub> with CQDs derived from biomass as the carbon precursor, and the utilization of harmless sunlight irradiation for photocatalytic activity needs to be explored. This is because, the former studies were done using graphite rod and I-ascorbic acid as the precursor. The precursor is not cost-effective, not easily available, and meanwhile, the usage of chemicals as the precursor is considered not environmentally friendly [29,38,40,70]. The previous studies also involved harsh and multiple steps procedures, which were time-consuming and sometimes produced poor phase structure and bigger size of photocatalyst that effect the photocatalytic activity [29,40,70].

#### References

- J.O. Tijani, O.O. Fatoba, G. Madzivire, L.F. Petrik, A review of combined advanced oxidation technologies for the removal of organic pollutants from water, Water, Air, & Soil Pollution 225 (9) (2014) 2102. Available from: https://doi.org/10.1007/s11270-014-2102-y.
- [2] M. Anjum, R. Miandad, M. Waqas, F. Gehany, M.A. Barakat, Remediation of wastewater using various nano-materials, Arabian Journal of Chemistry 12 (2016) 4897–4919. Available from: https://doi.org/10.1016/j.arabjc.2016.10.004.

- [3] I. Oller, S. Malato, J.A. Sánchez-Pérez, Combination of advanced oxidation processes and biological treatments for wastewater decontamination—A review, Science of The Total Environment 409 (20) (2011) 4141–4166. Available from: https://doi.org/ 10.1016/j.scitotenv.2010.08.061.
- [4] N.A. Wardrop, A.G. Hill, M. Dzodzomenyo, G. Aryeetey, J.A. Wright, Livestock ownership and microbial contamination of drinking-water: evidence from nationally representative household surveys in Ghana, Nepal and Bangladesh, International Journal of Hygiene and Environmental Health 221 (1) (2018) 33–40. Available from: https://doi. org/10.1016/j.ijheh.2017.09.014.
- [5] C.J. Houtman, Emerging contaminants in surface waters and their relevance for the production of drinking water in Europe, Journal of Integrative Environmental Sciences 7 (4) (2010) 271–295. Available from: https://doi.org/10.1080/1943815X.2010.511648.
- [6] K.M. Reza, A. Kurny, F. Gulshan, Parameters affecting the photocatalytic degradation of dyes using TiO<sub>2</sub>: a review, Applied Water Science 7 (4) (2017) 1569–1578. Available from: https://doi.org/10.1007/s13201-015-0367-y.
- [7] S. Sharma, V. Dutta, P. Singh, P. Raizada, A. Rahmani-Sani, A. Hosseini-Bandegharaei, et al., Carbon quantum dot supported semiconductor photocatalysts for efficient degradation of organic pollutants in water: a review, Journal of Cleaner Production 228 (2019) 755–769. Available from: https://doi.org/10.1016/j. jclepro.2019.04.292.
- [8] M.N. Chong, B. Jin, C.W.K. Chow, C. Saint, Recent developments in photocatalytic water treatment technology: a review, Water Research 44 (10) (2010) 2997–3027. Available from: https://doi.org/10.1016/j.watres.2010.02.039.
- [9] Y. Deng, R. Zhao, Advanced oxidation processes (AOPs) in wastewater treatment, Current Pollution Reports 1 (3) (2015) 167–176. Available from: https://doi.org/ 10.1007/s40726-015-0015-z.
- [10] R. Lakshmipathy, N.C. Sarada, Methylene blue adsorption onto native watermelon rind: batch and fixed bed column studies, Desalination and Water Treatment 57 (23) (2016) 10632–10645. Available from: https://doi.org/10.1080/19443994.2015.1040462.
- [11] W.H. Glaze, Drinking-water treatment with ozone, Environmental Science & Technology 21 (3) (1987) 224–230. Available from: https://doi.org/10.1021/ es00157a001.
- [12] W.H. Glaze, J.-W. Kang, D.H. Chapin, The chemistry of water treatment processes involving ozone, hydrogen peroxide and ultraviolet radiation, Ozone: Science & Engineering 9 (4) (1987) 335–352. Available from: https://doi.org/10.1080/ 01919518708552148.
- [13] A. Stocking, R. Rodriguez, T. Browne, D. Ph, 3.0 Advanced oxidation processes, Evaluation 32 (9–10) (2011) 1031–1041. Available from: https://doi.org/10.1002/ cite.200750374.
- [14] S.C. Ameta, Chapter 1 Introduction, in: S.C. Ameta, R. Ameta (Eds.), Advanced oxidation processes for waste water treatment, Academic Press, 2018, pp. 1–12. Available from: https://doi.org/10.1016/B978-0-12-810499-6.00001-2.
- [15] M.A. Oturan, J.-J. Aaron, Advanced oxidation processes in water/wastewater treatment: principles and applications. A review, Critical Reviews in Environmental Science and Technology 44 (23) (2014) 2577–2641. Available from: https://doi.org/10.1080/ 10643389.2013.829765.
- [16] S.Y. Lee, S.J. Park, TiO<sub>2</sub> photocatalyst for water treatment applications, Journal of Industrial and Engineering Chemistry 19 (6) (2013) 1761–1769. Available from: https://doi.org/10.1016/j.jiec.2013.07.012.

- [17] R. Saravanan, F. Gracia, A. Stephen, Nanocomposites for visible light-induced photocatalysis, Springer, 2017, pp. 19–41. Available from: https://doi.org/10.1007/978-3-319-62446-4.
- [18] A. Ibhadon, P. Fitzpatrick, Heterogeneous photocatalysis: recent advances and applications, Catalysts 3 (1) (2013) 189–218. Available from: https://doi.org/10.3390/catal3010189.
- [19] A. Kumar, A.R. Chowdhuri, D. Laha, T.K. Mahto, P. Karmakar, S.K. Sahu, Green synthesis of carbon dots from Ocimum sanctum for effective fluorescent sensing of Pb<sup>2+</sup> ions and live cell imaging, Sensors and Actuators, B: Chemical 242 (2017) 679–686. Available from: https://doi.org/10.1016/j.snb.2016.11.109.
- [20] R. Saravanan, M.M. Khan, V.K. Gupta, E. Mosquera, F. Gracia, V. Narayanan, et al., ZnO/Ag/CdO nanocomposite for visible light-induced photocatalytic degradation of industrial textile effluents, Journal of Colloid and Interface Science 452 (2015) 126–133. Available from: https://doi.org/10.1016/j.jcis.2015.04.035.
- [21] S. Rehman, R. Ullah, A.M. Butt, N.D. Gohar, Strategies of making TiO<sub>2</sub> and ZnO visible light active, Journal of Hazardous Materials 170 (2) (2009) 560–569. Available from: https://doi.org/10.1016/j.jhazmat.2009.05.064.
- [22] A. Fujishima, K. Honda, Electrochemical photolysis of water at a semiconductor electrode, Nature 238 (5358) (1972) 37–38. Available from: https://doi.org/10.1038/ 238037a0.
- [23] A. Fujishima, T.N. Rao, D.A. Tryk, Titanium dioxide photocatalysis, Journal of Photochemistry and Photobiology C: Photochemistry Reviews 1 (1) (2000) 1–21. Available from: https://doi.org/10.1016/S1389-5567(00)00002-2.
- [24] M.M. Khan, S.F. Adil, A. Al-Mayouf, Metal oxides as photocatalysts, Journal of Saudi Chemical Society 19 (5) (2015) 462–464. Available from: https://doi.org/10.1016/j. jscs.2015.04.003.
- [25] D. Beydoun, R. Amal, G. Low, S. McEvoy, Role of nanoparticles in photocatalysis, Journal of Nanoparticle Research 1 (4) (1999) 439–458. Available from: https://doi. org/10.1023/A:1010044830871.
- [26] Y. Zhang, J.C. Crittenden, D.W. Hand, D.L. Perram, Fixed-bed photocatalysts for solar decontamination of water, Environmental Science & Technology 28 (3) (1994) 435–442. Available from: https://doi.org/10.1021/es00052a015.
- [27] J.C. Colmenares, R. Luque, J.M. Campelo, F. Colmenares, Z. Karpiński, A.A. Romero, Nanostructured photocatalysts and their applications in the photocatalytic transformation of lignocellulosic biomass: an overview, Materials 2 (4) (2009) 2228–2258. Available from: https://doi.org/10.3390/ma2042228.
- [28] M.R. Hoffmann, S.T. Martin, W. Choi, D.W. Bahnemann, Environmental applications of semiconductor photocatalysis, Chemical Reviews 95 (1) (1995) 69–96. Available from: https://doi.org/10.1021/cr00033a004.
- [29] A. Makama, M. Umar, S.A. Saidu, CQD-based composites as visible-light active photocatalysts for purification of water, in: Y. Yao (Ed.), Visible-light photocatalysis of carbon-based materials, IntechOpen, 2018. Available from: https://doi.org/10.5772/ intechopen.74245.
- [30] I.K. Konstantinou, V.A. Sakkas, T.A. Albanis, Photocatalytic degradation of propachlor in aqueous TiO<sub>2</sub> suspensions. Determination of the reaction pathway and identification of intermediate products by various analytical methods, Water Research 36 (11) (2002) 2733–2742. Available from: https://doi.org/10.1016/s0043-1354(01)00505-x.
- [31] O. Carp, C.L. Huisman, A. Reller, Photoinduced reactivity of titanium dioxide, Progress in Solid State Chemistry 32 (1-2) (2004) 33-177. Available from: https:// doi.org/10.1016/j.progsolidstchem.2004.08.001.

- [32] S.H. Hyun, S.Y. Jo, B.S. Kang, Surface modification of γ-alumina membranes by silane coupling for CO<sub>2</sub> separation, Journal of Membrane Science 120 (2) (1996) 197–206. Available from: https://doi.org/10.1016/0376-7388(96)00160-3.
- [33] S. Leong, A. Razmjou, K. Wang, K. Hapgood, X. Zhang, H. Wang, TiO2 based photocatalytic membranes: a review, Journal of Membrane Science 472 (2014) 167–184. Available from: https://doi.org/10.1016/j.memsci.2014.08.016.
- [34] B. Samuneva, V. Kozhukharov, C. Trapalis, R. Kranold, Sol-gel processing of titaniumcontaining thin coatings - Part I Preparation and structure, Journal of Materials Science 28 (9) (1993) 2353–2360. Available from: https://doi.org/10.1007/BF01151665.
- [35] A. Stoyanova, H. Hitkova, A. Bachvarova-Nedelcheva, R. Iordanova, N. Ivanova, M. Sredkova, Synthesis and antibacterial activity of TiO<sub>2</sub>/ZnO nanocomposites prepared via nonhydrolytic route, Journal of Chemical Technology and Metallurgy 48 (2) (2013) 154–161.
- [36] S.Y. Lim, W. Shen, Z. Gao, Carbon quantum dots and their applications, Chemical Society Reviews 44 (1) (2015) 362–381. Available from: https://doi.org/10.1039/C4CS00269E.
- [37] T. Cetinkaya, L. Neuwirthová, K.M. Kutláková, V. Tomášek, H. Akbulut, Synthesis of nanostructured TiO<sub>2</sub>/SiO<sub>2</sub> as an effective photocatalyst for degradation of acid orange, Applied Surface Science 279 (2013) 384–390. Available from: https://doi.org/10.1016/ j.apsusc.2013.04.121.
- [38] X. Yu, J. Liu, Y. Yu, S. Zuo, B. Li, Preparation and visible light photocatalytic activity of carbon quantum dots/TiO<sub>2</sub> nanosheet composites, Carbon 68 (2014) 718–724. Available from: https://doi.org/10.1016/j.carbon.2013.11.053.
- [39] A. Prasannan, T. Imae, One-pot synthesis of fluorescent carbon dots from orange waste peels, Industrial and Engineering Chemistry Research 52 (44) (2013) 15673–15678. Available from: https://doi.org/10.1021/ie402421s.
- [40] J. Ke, X. Li, Q. Zhao, B. Liu, S. Liu, S. Wang, Upconversion carbon quantum dots as visible light responsive component for efficient enhancement of photocatalytic performance, Journal of Colloid and Interface Science 496 (2017) 425–433. Available from: https://doi.org/10.1016/j.jcis.2017.01.121.
- [41] Y. Wang, A. Hu, Carbon quantum dots: synthesis, properties and applications, Journal of Materials Chemistry C 2 (34) (2014) 6921–6939. Available from: https://doi.org/ 10.1039/c4tc00988f.
- [42] V. Sharma, P. Tiwari, S.M. Mobin, Sustainable carbon-dots: recent advances in green carbon dots for sensing and bioimaging, Journal of Materials Chemistry B 5 (45) (2017) 8904–8924. Available from: https://doi.org/10.1039/c7tb02484c.
- [43] K. Dimos, Carbon quantum dots: surface passivation and functionalization, Current Organic Chemistry 20 (2016) 682–695. Available from: https://doi.org/10.2174/ 1385272819666150730220948.
- [44] H. Zhang, H. Huang, H. Ming, H. Li, L. Zhang, Y. Liu, et al., Carbon quantum dots/ Ag<sub>3</sub>PO<sub>4</sub> complex photocatalysts with enhanced photocatalytic activity and stability under visible light, Journal of Materials Chemistry 22 (21) (2012) 10501–10506. Available from: https://doi.org/10.1039/C2JM30703K.
- [45] L. Zhang, J. Lian, L. Wu, Z. Duan, J. Jiang, L. Zhao, Synthesis of a thin-layer MnO<sub>2</sub> nanosheet-coated Fe<sub>3</sub>O<sub>4</sub> nanocomposite as a magnetically separable photocatalyst, Langmuir 30 (23) (2014) 7006–7013. Available from: https://doi.org/10.1021/la500726v.
- [46] R. Miao, Z. Luo, W. Zhong, S.Y. Chen, T. Jiang, B. Dutta, et al., Mesoporous TiO<sub>2</sub> modified with carbon quantum dots as a high-performance visible light photocatalyst, Applied Catalysis B: Environmental 189 (2016) 26–38. Available from: https://doi.org/ 10.1016/j.apcatb.2016.01.070.

- [47] B. Thangaraj, P.R. Solomon, S. Ranganathan, Synthesis of carbon quantum dots with special reference to biomass as a source – a review, Current Pharmaceutical Design 25 (13) (2019) 1455–1476. Available from: https://doi.org/10.2174/1381612825666190618154518.
- [48] T. Zhang, X. Wang, X. Zhang, Recent progress in TiO<sub>2</sub>-mediated solar photocatalysis for industrial wastewater treatment, International Journal of Photoenergy 2014 (March) (2014) 607954. Available from: https://doi.org/10.1155/2014/607954.
- [49] G. Rothenberger, P. Comte, M. Gra, A contribution to the optical design of dyesensitized nanocrystalline solar cells, Fuel and Energy Abstracts 40 (6) (1999) 398. Available from: https://doi.org/10.1016/s0140-6701(99)98992-0.
- [50] T. Sakai, D. Mersch, E. Reisner, Photocatalytic hydrogen evolution with a hydrogenase in a mediator-free system under high levels of oxygen nss, Angewandte Chemie International Edition 52 (2013) 12313–12316. Available from: https://doi.org/10.1002/ anie.201306214.
- [51] X. Zhang, Y. Lin, J. Wu, J. Jing, B. Fang, Improved performance of CdSe/CdS/PbS cosensitized solar cell with double-layered TiO<sub>2</sub> films as photoanode, Optics Communications 395 (C) (2017) 117–121. Available from: https://doi.org/10.1016/j. optcom.2016.05.026.
- [52] K. Maeda, G. Sahara, M. Eguchi, O. Ishitani, Hybrids of a ruthenium(II) polypyridyl complex and a metal oxide nanosheet for dye-sensitized hydrogen evolution with visible light: effects of the energy structure on photocatalytic activity, ACS Catalysis 5 (3) (2015) 1700–1707. Available from: https://doi.org/10.1021/acscatal.5b00040.
- [53] S. Ardo, G.J. Meyer, Photodriven heterogeneous charge transfer with transition-metal compounds anchored to TiO<sub>2</sub> semiconductor surfaces, Chemical Society Reviews 38 (1) (2009) 115–164. Available from: https://doi.org/10.1039/b804321n.
- [54] H. Ozawa, S. Honda, D. Katano, T. Sugiura, H. Arakawa, Novel ruthenium sensitizers with a dianionic tridentate ligand for dye-sensitized solar cells: the relationship between the solar cell performances and the electron-withdrawing ability of substituents on the ligand, Dalton Transactions 43 (2014) 8026–8036. Available from: https://doi.org/ 10.1039/c3dt52873a.
- [55] S.N. Baker, G.A. Baker, Luminescent carbon nanodots: emergent nanolights, Angewandte Chemie International Edition 49 (38) (2009) 6726–6744. Available from: https://doi.org/10.1002/anie.200906623.
- [56] R. Wang, K.-Q. Lu, Z.-R. Tang, Y.-J. Xu, Recent progress in carbon quantum dots: synthesis{,} properties and applications in photocatalysis, J. Mater. Chem. A 5 (8) (2017) 3717–3734. Available from: https://doi.org/10.1039/C6TA08660H.
- [57] X. Wang, L. Cao, F. Lu, M.J. Meziani, H. Li, G. Qi, et al., Photoinduced electron transfers with carbon dots, Chemical Communications 25 (2009) 3774–3776. Available from: https://doi.org/10.1039/b906252a.
- [58] X. Zhang, H. Huang, J. Liu, Y. Liu, Z. Kang, Carbon quantum dots serving as spectral converters through broadband upconversion of near-infrared photons for photoelectrochemical hydrogen generation, Journal of Materials Chemistry A 1 (38) (2013) 11529–11533. Available from: https://doi.org/10.1039/C3TA12568H.
- [59] Q. Jin, A. Gubu, X. Chen, X. Tang, A photochemical avenue to photoluminescent Ndots and their upconversion cell imaging, Scientific Reports 7 (1) (2017) 1–7. Available from: https://doi.org/10.1038/s41598-017-01663-x.
- [60] H. Yu, H. Zhang, H. Huang, Y. Liu, H. Li, H. Ming, et al., ZnO/carbon quantum dots nanocomposites: one-step fabrication and superior photocatalytic ability for toxic gas degradation under visible light at room temperature, New Journal of Chemistry 36 (4) (2012) 1031–1035. Available from: https://doi.org/10.1039/C2NJ20959D.

- [61] P. Chen, F. Wang, Z.-F. Chen, Q. Zhang, Y. Su, L. Shen, et al., Study on the photocatalytic mechanism and detoxicity of gemfibrozil by a sunlight-driven TiO<sub>2</sub>/carbon dots photocatalyst: the significant roles of reactive oxygen species, Applied Catalysis B: Environmental 204 (2017) 250–259. Available from: https://doi.org/10.1016/j. apcatb.2016.11.040.
- [62] C.N.V. McCormick, H.M. Winter, A.M. Goforth, R.M. Mackiewicz, T.M. McCormick, Photocatalytic water reduction using a polymer coated carbon quantum dot sensitizer and a nickel nanoparticle catalyst, Nanotechnology 28 (19) (2017) 195402. Available from: http://stacks.iop.org/0957-4484/28/i = 19/a = 195402.
- [63] K.-H. Ye, Z. Wang, J. Gu, S. Xiao, Y. Yuan, Y. Zhu, et al., Correction: carbon quantum dots as a visible light sensitizer to significantly increase the solar water splitting performance of bismuth vanadate photoanodes, Energy & Environmental Science 10 (2) (2017) 642. Available from: https://doi.org/10.1039/C7EE90006F.
- [64] Z. Ren, X. Liu, H. Chu, H. Yu, Y. Xu, W. Zheng, et al., Journal of colloid and interface science carbon quantum dots decorated MoSe 2 photocatalyst for Cr (VI) reduction in the UV – vis-NIR photon energy range, Journal of Colloid And Interface Science 488 (2017) 190–195. Available from: https://doi.org/10.1016/j.jcis.2016.10.077.
- [65] Y. Wang, F. Wang, Y. Feng, Z. Xie, Q. Zhang, X. Jin, et al., Facile synthesis of carbon quantum dots loaded with mesoporous g-C3N4 for synergistic absorption and visible light photodegradation of fluoroquinolone antibiotics, Dalton Transactions 47 (4) (2018) 1284–1293. Available from: https://doi.org/10.1039/C7DT04360K.
- [66] S.S. Boxi, S. Paria, Visible light induced enhanced photocatalytic degradation of organic pollutants in aqueous media using Ag doped hollow TiO2 nanospheres, RSC Advances 5 (47) (2015) 37657–37668. Available from: https://doi.org/10.1039/ C5RA03421C.
- [67] X. Xiong, H. Chen, Y. Xu, Improved photocatalytic activity of TiO<sub>2</sub> on the addition of CuWO4, The Journal of Physical Chemistry C 119 (11) (2015) 5946–5953. Available from: https://doi.org/10.1021/jp510974f.
- [68] M. Wang, S. Shen, L. Li, Z. Tang, J. Yang, Effects of sacrificial reagents on photocatalytic hydrogen evolution over different photocatalysts, Journal of Materials Science 52 (9) (2017) 5155–5164. Available from: https://doi.org/10.1007/s10853-017-0752-z.
- [69] S. Wonyong, As featured in: energy & environmental science photocatalysis based on modified TiO<sub>2</sub>, Energy & Environmental Science 9 (2016) 411–433. Available from: https://doi.org/10.1039/c5ee02575c.
- [70] M. Sun, X. Ma, X. Chen, Y. Sun, X. Cui, Y. Lin, A nanocomposite of carbon quantum dots and TiO<sub>2</sub> nanotube arrays: enhancing photoelectrochemical and photocatalytic properties, RSC Advances 4 (3) (2014) 1120–1127. Available from: https://doi.org/ 10.1039/c3ra45474f.
- [71] C. Han, M.Q. Yang, B. Weng, Y.J. Xu, Improving the photocatalytic activity and antiphotocorrosion of semiconductor ZnO by coupling with versatile carbon, Physical Chemistry Chemical Physics 16 (32) (2014) 16891–16903. Available from: https://doi. org/10.1039/c4cp02189d.
- [72] A. Makama, A. Salmiaton, E.B. Saion, T.S.Y. Choong, N. Abdullah, Microwaveassisted synthesis of porous ZnO/SnS2 heterojunction and its enhanced photoactivity for water purification, Journal of Nanomaterials 2015 (2015). Available from: https:// doi.org/10.1155/2015/108297.
- [73] H. Li, Z. Kang, Y. Liu, S.-T. Lee, Carbon nanodots: synthesis{,} properties and applications, Journal of Materials Chemistry A 22 (46) (2012) 24230–24253. Available from: https://doi.org/10.1039/C2JM34690G.

- [74] Y. Dong, R. Wang, H. Li, J. Shao, Y. Chi, X. Lin, et al., Polyamine-functionalized carbon quantum dots for chemical sensing, Carbon 50 (8) (2012) 2810–2815. Available from: https://doi.org/10.1016/j.carbon.2012.02.046.
- [75] A.B. Bourlinos, A. Stassinopoulos, D. Anglos, R. Zboril, M. Karakassides, E.P. Giannelis, Surface functionalized carbogenic quantum dots, Small 4 (4) (2007) 455–458. Available from: https://doi.org/10.1002/smll.200700578.
- [76] A.M. Alam, B.Y. Park, Z.K. Ghouri, M. Park, H.Y. Kim, Synthesis of carbon quantum dots from cabbage with down- and up-conversion photoluminescence properties: excellent imaging agent for biomedical applications, Green Chemistry 17 (7) (2015) 3791–3797. Available from: https://doi.org/10.1039/c5gc00686d.
- [77] V.N. Mehta, S. Jha, H. Basu, R.K. Singhal, S.K. Kailasa, One-step hydrothermal approach to fabricate carbon dots from apple juice for imaging of mycobacterium and fungal cells, Sensors and Actuators, B: Chemical 213 (July 2015) (2015) 434–443. Available from: https://doi.org/10.1016/j.snb.2015.02.104.
- [78] Y. Dong, N. Zhou, X. Lin, J. Lin, Y. Chi, G. Chen, Extraction of electrochemiluminescent oxidized carbon quantum dots from activated carbon, Chemistry of Materials 22 (21) (2010) 5895–5899. Available from: https://doi.org/10.1021/cm1018844.
- [79] H. Peng, J. Travas-Sejdic, Simple aqueous solution route to luminescent carbogenic dots from carbohydrates, Chemistry of Materials 21 (23) (2009) 5563–5565. Available from: https://doi.org/10.1021/cm901593y.
- [80] J. Tian, R. Liu, Y. Zhao, Q. Xu, S. Zhao, Controllable synthesis and cell-imaging studies on CdTe quantum dots together capped by glutathione and thioglycolic acid, Journal of Colloid and Interface Science 336 (2) (2009) 504–509. Available from: https://doi.org/10.1016/j.jcis.2009.04.064.
- [81] M. Ming, Z. Ma, Y. Liu, K. Pan, H. Yu, F. Wang, et al., Large scale electrochemical synthesis of high quality carbon nanodots and their photocatalytic property, Dalton Transactions 41 (31) (2012) 9526–9531. Available from: https://doi.org/10.1039/c2dt30985h.
- [82] W. Hu, Y.C. Shin, G. King, Modeling of multi-burst mode pico-second laser ablation for improved material removal rate, Applied Physics A: Materials Science and Processing 98 (2) (2010) 407–415. Available from: https://doi.org/10.1007/s00339-009-5405-x.
- [83] N. Gong, H. Wang, S. Li, Y. Deng, X. Chen, L. Ye, et al., Microwave-assisted polyol synthesis of gadolinium-doped green luminescent carbon dots as a bimodal nanoprobe, Langmuir 30 (36) (2014) 10933–10939. Available from: https://doi.org/10.1021/la502705g.
- [84] A. Jaiswal, S. Sankar Ghosh, A. Chattopadhyay, One step synthesis of C-dots by microwave mediated caramelization of poly(ethylene glycol, Chemical Communications 48 (3) (2012) 407–409. Available from: https://doi.org/10.1039/c1cc15988g.
- [85] C. Zhai, N. Teng, B. Pan, J. Chen, F. Liu, J. Zhu, et al., Revealing the importance of non-thermal effect to strengthen hydrolysis of cellulose by synchronous cooling assisted microwave driving, Carbohydrate Polymers 197 (March) (2018) 414–421. Available from: https://doi.org/10.1016/j.carbpol.2018.06.031.
- [86] C. Hu, M. He, B. Chen, B. Hu, A sol-gel polydimethylsiloxane/polythiophene coated stir bar sorptive extraction combined with gas chromatography-flame photometric detection for the determination of organophosphorus pesticides in environmental water samples, Journal of Chromatography A 1275 (2013) 25–31. Available from: https:// doi.org/10.1016/j.chroma.2012.12.036.
- [87] Z.A. Qiao, Y. Wang, Y. Gao, H. Li, T. Dai, Y. Liu, et al., Commercially activated carbon as the source for producing multicolor photoluminescent carbon dots by chemical oxidation, Chemical Communications 46 (46) (2010) 8812–8814. Available from: https://doi.org/10.1039/c0cc02724c.

- [88] S. Wang, L. Zhao, J. Ran, Z. Shu, G. Dai, P. Zhai, Effects of calcination temperatures on photocatalytic activity of ordered titanate nanoribbon/SnO<sub>2</sub> films fabricated during an EPD process, International Journal of Photoenergy 2012 (2012) 472958. Available from: https://doi.org/10.1155/2012/472958.
- [89] H. Sun, Q.F. Zhang, J.L. Wu, Electroluminescence from ZnO nanorods with an n-ZnO/p-Si heterojunction structure, Nanotechnology 17 (9) (2006) 2271–2274. Available from: https://doi.org/10.1088/0957-4484/17/9/033.
- [90] S. Yang, W. Cai, H. Zhang, X. Xu, H. Zeng, Size and structure control of Si nanoparticles by laser ablation in different liquid media and further centrifugation classification, Journal of Physical Chemistry C 113 (44) (2009) 19091–19095. Available from: https://doi.org/10.1021/jp907285f.
- [91] Y. Li, Q. Du, T. Liu, J. Sun, Y. Wang, S. Wu, et al., Methylene blue adsorption on graphene oxide/calcium alginate composites, Carbohydrate Polymers 95 (1) (2013) 501–507. Available from: https://doi.org/10.1016/j.carbpol.2013.01.094.
- [92] S. Zhuo, M. Shao, S. Lee, upconversion GQD\_tio2\_ACSnano 2012.pdf 2 (2012) 1059-1064.
- [93] D. Ma, X. Chen, G. Huang, J. Chen, H. Zhou, L. Fang, Temperature stability, structural evolution and dielectric properties of BaTiO3-Bi(Mg2/3Ta1/3)O3 perovskite ceramics, Ceramics International 41 (5) (2015) 7157–7161. Available from: https:// doi.org/10.1016/j.ceramint.2015.02.036.
- [94] S. Sahu, B. Behera, T.K. Maiti, S. Mohapatra, Simple one-step synthesis of highly luminescent carbon dots from orange juice: application as excellent bio-imaging agents, Chemical Communications 48 (70) (2012) 8835–8837. Available from: https://doi.org/10.1039/c2cc33796g.
- [95] A. Yuliansyah, T. Hirajima, S. Kumagai, K. Sasaki, Production of solid biofuel from agricultural wastes of the palm oil industry by hydrothermal treatment, Waste and Biomass Valorization 1 (2010) 395–405.
- [96] A.N. Bugrov, O.V. Almjasheva, Effect of hydrothermal synthesis conditions on the morphology Of Zro<sub>2</sub> nanoparticles, Nanosystems: Physics, Chemistry, Mathematics 4 (6) (2013) 810–815.
- [97] Z.A. Zulkifli, K.A. Razak, W.N.W.A. Rahman, The effect of reaction temperature on the particle size of bismuth oxide nanoparticles synthesized via hydrothermal method, AIP Conference Proceedings 1958 (2018) 020007. Available from: https://doi.org/ 10.1063/1.5034538.
- [98] G.C. Collazzo, S.L. Jahn, E.L. Foletto, Temperature and reaction time effects on the structural properties of titanium dioxide nanopowders obtained via the hydrothermal method, Brazilian Journal of Chemical Engineering 28 (02) (2011) 265–272.
- [99] A. Emanuele, S. Cailotto, C. Campalani, L. Branzi, C. Raviola, D. Ravelli, et al., Biomass-derived carbon dots and their applications, Energy & Environmental Materials 2 (3) (2019) 172–192. Available from: https://doi.org/10.3390/ nano9030387.
- [100] J. Shen, S. Shang, X. Chen, D. Wang, Y. Cai, Facile synthesis of fluorescence carbon dots from sweet potato for Fe<sup>3+</sup> sensing and cell imaging, Materials Science and Engineering C 76 (2017) 856–864. Available from: https://doi.org/10.1016/j. msec.2017.03.178.
- [101] C. Zhu, J. Zhai, S. Dong, Bifunctional fluorescent carbon nanodots: green synthesis via soy milk and application as metal-free electrocatalysts for oxygen reduction, Chemical Communications 48 (75) (2012) 9367–9369. Available from: https://doi. org/10.1039/c2cc33844k.

- [102] C. Yu, T. Xuan, Y. Chen, Z. Zhao, Z. Sun, H. Li, A facile, green synthesis of highly fluorescent carbon nanoparticles from oatmeal for cell imaging, Journal of Materials Chemistry C 3 (37) (2015) 9514–9518. Available from: https://doi.org/10.1039/ c5tc02057c.
- [103] B.S.B. Kasibabu, S.L. D'souza, S. Jha, R.K. Singhal, H. Basu, S.K. Kailasa, One-step synthesis of fluorescent carbon dots for imaging bacterial and fungal cells, Analytical Methods 7 (6) (2015) 2373–2378. Available from: https://doi.org/10.1039/ c4ay02737j.
- [104] A. Sachdev, P. Gopinath, Green synthesis of multifunctional carbon dots from coriander leaves and their potential application as antioxidants, sensors and bioimaging agents, Analyst 140 (12) (2015) 4260–4269. Available from: https://doi.org/10.1039/ c5an00454c.
- [105] W. Lu, X. Qin, S. Liu, G. Chang, Y. Zhang, Y. Luo, et al., Sun, Economical, green synthesis of fluorescent carbon nanoparticles and their use as probes for sensitive and selective detection of mercury(II) ions, Analytical Chemistry 84 (12) (2012) 5351–5357. Available from: https://doi.org/10.1021/ac3007939.
- [106] U.R.R.P. Remli, A.A. Aziz, Photocatalytic degradation of methyl orange using Carbon Quantum Dots (CQDs) derived from watermelon rinds, IOP Conference Series: Materials Science and Engineering 736 (4) (2020) 91–100. Available from: https://doi.org/10.1088/1757-899X/736/4/042038.
- [107] F. Ajayi, E. Ndor, N. State, Growth and yield of water melon (Citrullus lanatus) as affected by poultry manure application, Journal of Agriculture and Social Sciences 4 (2008) 121–124.
- [108] G. Gladvin, G. Sudhaakr, V. Swathi, K.V. Santhisri, Mineral and vitamin compositions contents in watermelon peel (Rind), International Journal of Current Microbiology and Applied Sciences 5 (5) (2017) 129–133.
- [109] M. Mushtaq, B. Sultana, RSM based optimized enzyme-assisted extraction of antioxidant phenolics from underutilized watermelon (*Citrullus lanatus* Thunb.) rind, J Food Sci Technol. 52 (August) (2015) 5048–5056. Available from: https://doi.org/10.1007/ s13197-014-1562-9.
- [110] U.K. Ibrahim, N. Kamarrudin, M.U.H. Suzihaque, S. Abd Hashib, Local fruit wastes as a potential source of natural antioxidant: an overview, IOP Conference Series: Materials Science and Engineering 206 (1) (2017) 012040. Available from: https://doi. org/10.1088/1757-899X/206/1/012040.
- [111] A.H. Jawad, Y.S. Ngoh, K.A. Radzun, Utilization of watermelon (*Citrullus lanatus*) rinds as a natural low-cost biosorbent for adsorption of methylene blue: kinetic, equilibrium and thermodynamic studies, Journal of Taibah University for Science 12 (4) (2018) 371–381. Available from: https://doi.org/10.1080/16583655.2018.1476206.
- [112] A. Ibrahim, L. Yusof, N.S. Beddu, N. Galasin, P.Y. Lee, R.N.S. Lee, et al., Adsorption study of ammonia nitrogen by watermelon rind, IOP Conference Series: Earth and Environmental Science 36 (1) (2016). Available from: https://doi.org/ 10.1088/1755-1315/36/1/012020.
- [113] C.S.C. Kumar, R. Mythily, S. Chandraju, B. Nagar, Studies on sugars extracted from water melon (*Citrullus lanatus*) rind, A remedy for related waste and its management, International Journal of Chemical and Analytical Science 3 (8) (2012) 1527–1529.
- [114] P.L. Kiew, J.F. Toong, Akademia Baru Progress in Energy and Environment Screening of Significant Parameters Affecting Zn (II) Adsorption by Chemically Treated Watermelon Rind, 6, Akademia Baru, 2018, pp. 19–32.

- [115] O.A. Oseni, V.I. Okoye, Ournal of pharmaceutical and biomedical sciences, Journal of Pharmaceutical and Biomedical Sciences 27 (14) (2013) 508–514.
- [116] J. Treml, K. Šmejkal, Flavonoids as potent scavengers of hydroxyl radicals, Comprehensive Reviews in Food Science and Food Safety 15 (4) (2016) 720–738. Available from: https://doi.org/10.1111/1541-4337.12204.
- [117] L.Y. Jun, R.R. Karri, L.S. Yon, N.M. Mubarak, C.H. Bing, K. Mohammad, et al., Modeling and optimization by particle swarm embedded neural network for adsorption of methylene blue by jicama peroxidase immobilized on buckypaper/polyvinyl alcohol membrane, Environmental Research 183 (January) (2020) 109158. Available from: https://doi.org/10.1016/j.envres.2020.109158.

### Index

Note: Page numbers followed by "f" and "t" refer to figures and tables, respectively.

#### A

Advanced oxidation process and photocatalysis AOP classifications for wastewater treatment, 265f existing water treatment technology, 264t for water and wastewater treatment, 265t Application of carbon quantum dots in photocatalysis for wastewater purification, 284t in water purification, 283-285 Applications of magnetic and nanophotonics magnetic, 378-382 carbon quantum dots decorated, 379 - 382Curie like paramagnetism at 2K, 380f drug delivery and synergistic chemophotothermal therapy, 383f encapsulated 1D magnetic nanostructures. 382 HA-HMCN(DOX) @GQDs nanoplatform targeting drug delivery, 381f nanophotonic applications and singlephoton emission, 382-389 chemical synthesis, optical responses, and circular dichroism characterization, 389f chiral photonics and twistronics, 386 - 388emission states variation and excitedstate lifetime distributions, 390f and GQDs in the context of lightemitting diodes, 385f hotspot and plexcitons generation, 391f light-emitting diodes, 382-384 memory devices, 386 photovoltaic solar cell, 384-385

single/few photons source and cavity-assisted photonics, 388–389 for solar cells and efficiency characterization, 387*f* 

#### B

Bioimaging applications of carbon dots optical property, 227
Bioimaging modalities, development of, 240–241
Biomedical applications of carbon dots of C-dots, 228*f* crossing blood-brain barrier, 229–230 drug delivery, 227–229 gene delivery, 230
Biosensing applications using carbon dots therapeutics, diagnostics, and theranostics of C, 231*f*

#### С

Carbon-based nanodots in photocatalytic CO<sub>2</sub> conversion nanoparticle size and percentage of atoms, 298f photo- and/or electrochemical, 296f photoconversion, 310-326 photoinduced charge transfer process, 298f synthetic approaches and optical properties of, 299-310 Carbon-based quantum dots in CO<sub>2</sub> photoconversion with carbon-based catalysts, 314-326 CO yields over various photocatalysts, 324f electron transfer reactions for the photocatalytic conversion, 325f

Carbon-based quantum dots in CO2 photoconversion (Continued) nanoparticle surface functionalized with PEG diamine molecules, 315f photocatalytic CO<sub>2</sub> conversion employing carbon-based catalysts, 327t photocatalytic CO<sub>2</sub> conversion into methanol, 320f photoinduced electron transfer on, 317f photoluminescence emission curves, 318f standard redox potentials, 319f photocatalytic reduction, 310-314 graphene and graphene oxide-based materials, 313f on semiconductor photocatalyst, 311f solubility characteristics of, 312f photophysical characteristics, 314-326 Carbon dots:biomedical applications bioimaging applications, 226-227 biomedical applications, 227-230 biosensing applications, 230-231 biosynthesis of C-dots, 226 scope and challenges, 232-233 structure and functionalization, 225 Carbon dots types, 30f Carbon nanolights, 243 Carbon nanostructures in terahertz domain spectroscopy for, 414-416, 415f time-resolved spectroscopy and conductivity, 416 Carbon quantum dots applications of, 410-414 carbon nanostructures in, 414-416 characteristic lengths, 400-402 distinctive properties of, 398 fabrication techniques of quantum dots, 404 - 406nano carbon dots in light-emitting diode, phototransistor, 399f optical properties of quantum dots, 407-410 quantum dot, 402-404 structure of 0D fullerene, 398f Carbon quantum dot-based nanosensors chemical sensing, 206-223 introduction to. 205-206 Carbon quantum dot in the biomedical field

applications of, 410-414 crossing blood-brain barrier, 413-414 drug delivery, 413 gene delivery, 414 optical imaging, 410-412 photoacoustic imaging, 413 Carbon quantum dots, 243 in all-weather solar cells, 154 approaches for synthesis of, 244f Arc Discharge method, 245 based bioimaging carbon quantum dots, 243 cDot in bioimaging, 248-253 development of, 240-241 nanomaterials as imaging agents, 242 - 243properties of cDots, 247-248 requirement of imaging agents, 241 surface activation, 245-247 synthesis and modifications in, 244 - 245timeline of important discovery, 240f for battery and charge storage capacity, 365t in lithium-ion and sodium-ion, 362 - 363for charge storage applications electrical, 350 optical, 350-351 property of, 349-351 structural property, 349-350 chemical ablation, 244 as counter electrode bifacial DSSCs, 144f charge carrier transfer processes, 144f device applications EL spectra and true color photographs of multicolor, 171f doping, 247 in dye-sensitized solar cells as counter electrode, 142-144 for electrochemical energy storage applications for battery, 361-366 David V. Ragone plot, 342f properties of, 349-351 for supercapacitors, 351-361 supercapacitors and batteries, 342-349 electrochemical method, 244

EL spectra and true color photographs of multicolor CDs with tethered imidazolidinones (IS-CDs), 173f high-color-purity deep-blue (HCP-DB), 173f quantum dot based LEDs(QD-LEDs), 174fhydrothermal method, 245 laser, 200 laser ablation, 244-245 light-emitting diodes, 169-174 in lithium-ion and sodium-ion with bicontinuous electron and Li/Na ion transfer channels, 362f and charge storage capacity, 363t sodium-ion batteries, 364t in lithium-sulfur battery, 364-365 microwave irradiation, 245 normalized green lasing emission, 175f optical gain and lasing, 174-175 in organic solar cells S-doped CQDs (N,S-CQDs), 147-148 solar device with CQDs, 146f structure of device, 146f tandem cell architecture, 147f in perovskite solar cells current density vs voltage, 153f red CQDs (hydroxyl-rich) (RCQs)doped SnO<sub>2</sub>, 152-154 slow photon effect, 152-154 in potassium-ion battery, 364, 364t pyrolysis method, 245 as sensitizer, 137-142 device structure of, 137f N300-CQDs and N719 cosensitized DSSCs device, 141f NCQDs using citric acid and ammonia via direct pyrolysis method, 140f photovoltaic devices with, 142t synthesis scheme of, 139f in solid-state solar cells cost-effective heterojunction photovoltaic device, 148-150 CQDs-coated SiNWs, 149f solid-state solar cells, 148 for supercapacitors Bibekananda De's group, 355 charge storage capacity, 361t

core-shell CuS@CQDs@carbon hollow nanospheres, 355f CQDs-inorganic supercapacitors, 356t CQDs/PPy, 358f of GO/CDs/PPy composite, 357f graphene quantum dots, 360-361 graphene sheet structure, 351f inorganic hybrid for supercapacitors, 353-355 MnO<sub>2</sub>/CQDs nanowires and cycling, 354f organic CQDs and charge storage capacity, 359t organic hybrid supercapacitors, 355-359 oxygen groups, 352 sources and role of, 353f surface functionalization, 246-247 surface passivation, 246, 246f typical device structure of, 171f in zinc-ion battery, 365-366 Outlook, 365-366 CDot in bioimaging cDots in in vivo and in vitro bioimaging, 249f microbiota using C dots, 250f single-molecule imaging, 251-253 in in vitro bioimaging, 252t in vitro imaging, 250-251 in vivo bioimaging, 253t in vivo imaging, 251 Characteristic lengths de Broglie's theory of wave-particle duality, 400 de Broglie wavelength, 400-401 diffusion, 401 localization, 402 mean free path, 401 screening, 401-402 Characterization and physical properties of carbon quantum dots allotropes structures, 57t carbon quantum dots, 56-59 chemical and electronic structures of, 59-60, 60fenergy structure of, 57f generalized structure of, 59f and its lifetime, 58f structure of, 59-60

Characterization and physical properties of carbon quantum dots (Continued) classification as allotropes, 56f types, 60-61 doped, 61, 62f hydrophilic, 61 hydrophobic, 61 undoped, 61, 62f Characterization of carbon quantum dots photophysical analysis, 80-84 fluorescence, 82-83, 82f Forster resonance energy transfer, 83 - 84photoluminescence, 80-82, 81f riboflavin absorption, 85f stability of, 85 structural atomic force microscopy, 79, 79f Fourier-transform Infrared, 78-79, 78f Raman spectroscopy, 75-76, 76f scanning electron microscope, 74 transmission electron microscope, 74 - 75UV-vis spectra, 79-80 X-ray photoelectron spectroscopy, 77-78, 77f X-ray powder diffraction, 73-74 XRD pattern of crystalline, 73f Charging energy, 425 Chemical sensing CQDs, 208-212 fluorescence from, 208-210 fluorescence sensing by, 211-212 free zigzag sites with a carbine-like triplet ground state, 210 quenching and, 211–212, 212f radiative recombination in small nanodomains. 209-210 fluorescence-based, 206-208 band edge and other transitions, 207f strong emission characteristics in, 207 - 208gas sensing with conducting carbon dots, 218 - 221CQDs on the electrical properties of conducting polymers, 221 designing of gas sensors using carbonaceous, 220-221

gas sensors using carbonaceous nanomaterials, 220-221 of PSS-CQDs, 220f for Gas/VOC sensing, 222-223 pH sensor, 212-215 doped CQDs in sensors, 216-218 extensive fluorescence quenching of DTT/C-dots, 217f hydrothermal cutting of oxidized GSs, 213f pH-dependent UV-vis absorption and PL spectra, 214f pH sensing with, 214-215 red emitting carbon dots for specific metal ion detection, 218 sensing dielectric of surrounding medium, 216 of solvent sensing, 216 surface groups in pH sensor applications of, 213-214 time-resolved fluorescence decay curves, 217f time-resolved fluorescence decay curves of NRCDs, 219f sensors, 208 VOC sensor based on, 221-223 Coulomb blockade oscillation (CBO), 425 Crossing blood-brain barrier, 229-230

#### D

DeBroglie wavelength, 400–401 DNA intercalator, 21 Drug delivery, 227–229

#### Е

Excitation-dependent fluorescence emission, 80 Exhibiting electroluminescence (EL), 70–71

#### F

Fabrication techniques of quantum dots absorption spectra of CdS nanocrystals, 406*f*based on II-VI compound semiconductors, 405–406
CdSe single nanocrystal absorption spectra, 407*f*growth modes, 405*f*PL and El spectra of InAs/GaAs, 407*f*

self-assembled, 406 Fundamentals of carbon quantum dots MB degradation by, 273f semiconductors, 273-274 simple and facile one-pot synthesis of fluorescent, 274 Fundamentals of supercapacitors of batteries, 345-349 lithium-ion, 346f lithium-ion and sodium-ion, 348f reaction mechanism in LSB. 348-349 and battery, 342-349 charge storage in EDLC, 343f in a hybrid supercapacitor, 344f in pseudocapacitors, 344f of supercapacitors, 342-345 Future perspectives of carbon quantum dots, 474 - 476Colloidal CODs for Green Optoelectronics, 475-476 coupling of Rh nanoparticles (NPs), 476f CQDs as Contrast Agent in Imaging, 475 Graphene QD-based PDs, 477f heavy metal-free QDs, 475-476 luminescent doped/co-doped CQDs for optical sensing, 474-476 multipotential applications of doped, 475f Natural and Biogreen CQDs for Biotechnology and Nanomedicine Applications, 475 new development, 474f potential drug delivery applications, 476f

#### G

Gene delivery, 230 Graphene quantum dots (GQDs), 40

#### H

Hotoluminescence properties of carbon quantum dots aggregation-induced emission in, 12–20, 13*f* added metal ions effect, 18–20 changes in, 15*f* concentration-dependent luminescence mechanism, 19*f* digital photographs of, 17*f* 

emission maxima versus concentration of, 18f energy transfer process, 18f excitation-dependent emission of, 14f GSH-CQDs in varying ethanol-water mixtures, 16f material concentration effect, 15f of N-CODs, 20f NDI-FONPs, 15f of OPD-CQDs, 19f PL decay curves of, 17f PL spectra of GSH-CQDs, 16f solvent polarity effect, 13-16 THF-water solvent percentage, 15f blue-, green-, yellow-, and red-emissive, 5fconfinement effect, 4 CQDs in daylight, 5f doping nonmetallic heteroatoms, 5-8 nitrogen precursor effect, 7f wavelength-dependent emission maxima, 7f electroluminescence of carbon dots, 23 - 25CQD-LED and doped CQD-LEDs, 25fFörster resonance energy transfer, 20-22 coupling chemistry for FRET-CQD-DDS, 19f CQD-EtBr based FRET pair, 22f FRET-COD-DDS, 19f heteroatom-mediated surface defects, 8 influence of edge states, 8-9 nonpolar solvents, fluorescence PL, 10f normalized PL spectra of, 6f photoinduced electron transfer, 22-23 red edge effect, 9-11 surface defect states, 11-12 passivated carbon dots with PEGs, 12f triple carbenes at zigzag sites, 9f Hydrothermal carbonization, 245 Hydrothermal treatment of carbon quantum dots type of green precursor, 281t

#### I

Imaging agents, 241 Inner filter effect, 95 Ionization energy, 402

#### K

Kondo effect, 36-37

#### L

Light-emitting diode application of carbon quantum dots device applications, 169–175 optical properties of, 163–169 synthesis methods of functionalized, 159–163 Localization length, 402, 436–437

#### М

Magnetic and nanophotonics applications of carbon quantum dots applications, 378-389 context of magnetic and nanophotonics, 379f Mechanism of photocatalysis principle of photocatalytic, 267f semiconductor photocatalyst, 266 Memory devices classifications of, 185-186, 187f light excitation dependence of photoinduced current and persistent photoconductivity, 187f random access memory, 187-193 artificial neuromorphic networks (ANNs), 190f cyclic multiple-valued voltage, 192f effects of UV irradiation on RS, 188f energy band diagram of the rectifying memory device, 189f RS behavior and retention characteristics, 191f WRER cycles, 192f Mesoscopic devices, 400 Metal sulfides dye wastewater, 272t Methodology of carbon quantum dots amino-functionalized fluorescent, 94 amino-functionalized, 95 branched polyethylenimine functionalized, 94-95 by hydrothermal carbonization of chitosan, 94f spiropyran-functionalized, 95-96 condensation reaction, 100-101 Europium-adjusted, 101

off-fluorescence probe of, 101*f* microwave-assisted pyrolysis, 96–99 hyperbranched polyethylenimine and isobutyric amide functionalized Cdots, 96–97 organic dye-functionalized, 98–99 organosilane functionalized, 97–98, 98*f* synthesis of organic dyes, 99*f* oxidation-polymerization reaction, 101–102 preparation of CD-PANI, 102*f* sol-gel reaction, 99–100 synthesis of CDs@MIP, 100*f* 

#### Ν

Nanocarbon-based single-electron transistors as electrometer reviewing published work, 434-452 single-electron transistor as an electrometer, 426-434 theory, 423-425 Nanodiamond photonics ND photonic applications, 457-459 xz-profile of individual NV color center, 458f optical emission from diamond, 455-456 SiV defect, 456f spectrum showing SiV, 457f Nanodiamonds for advanced photonic and biomedical applications NDs for biomedical applications, 459 - 466photonics, 455-459 Nanoelectronic applications of carbon quantum dots carbon quantum dot laser, 200 memory devices, 185-193 sensors, 194-197 transistors, 193-197 Nanomaterials as imaging agents, 242-243 NDs for biomedical applications cancer therapy, 459-462 gain attention of researchers, 460f imaging, 462-466 contrast agents for, 464-466 as contrast agents in MRI, 462-463 Overhauser effect, 464f PA signal amplitude, 466f as photostable markers in STED, 464

STED microscopy of green, 465*f* tumor metastasis inhibited by cNDs, 463*f* unique electrostatic properties of octahedral DNDs, 461*f* 

#### 0

Optical absorption and emission and electrical property, 108-111 GQDs with size and morphology, 111f PL spectra of NGQDs, 112f spectrum of SNGQDs, 110f Optical absorption properties of carbon quantum dots Kasha's molecular exciton theory, 3f UV-visible spectrum and, 3f Optical properties of carbon quantum dots electroluminescence, 168-169 based LED device structure, 170f optical absorption, 164 daylight photographs, 165f UV excited fluorescence, 166f UV-visible optical absorption, 165f from ultraviolet to near-infrared regions, 164 - 168due to quantum confinement effect, 164 - 165multicolor fluorescent CQDs, 167f surface passivation and functionalization effect, 166-167 up-conversion photoluminescence, 168 UV excited fluorescence images of, 168f Optical properties of carbon quantum dots and graphene quantum dots, 299-310 multicolor fluorescence, 303f structural depiction of CDs, 302f top-down and bottom-up methods, 301f optical properties of, 304 absorption spectrum of aqueous dispersed, 305f chemical structure on the photoluminescence behavior, 308f fluorescence spectra of water-soluble, 306f synthetic approaches and, 299-310 Optical properties of indirect gap nanocrystal enhanced phototransition in a silicon nanocrystal, 409f

Optical properties of quantum dots oscillator strength for phototransition, 408 phototransition in, 408 widening of bandgap, 408 Optoelectronics applications of carbon quantum dots Si/GQDs heterojunction solar cell, 113*f* silicon-based heterojunction solar cell, 112–113

#### P

Photocatalysts material semiconductors on the potential scale, 269f Photocatalytic applications for wastewater treatment advanced oxidation process, 263-266 application of, 283-285 binary metal oxides, 268-270 fundamentals of, 271-274 hydrothermal treatment of, 278-282 mechanism of photocatalysis, 266-267 metal sulfides, 270-271 in photocatalysis, 274-277 photocatalysts material, 268 synthesis route of, 277-278 watermelon rinds potential as, 283 Photodetector applications of carbon and graphene quantum dots, 113-128 optical absorption, emission, and electrical property, 108-111 optoelectronics applications of, 112 - 113synthesis of, 106-108 Photodetector applications of carbon quantum dots FET-based photodetectors using, 114 - 120device structure of DUV photodetector, 114f graphene/NGQDs FET-based phototransistor, 116f MoS<sub>2</sub>/GQDs phototransistor, 119f photodetector fabrication process, 115f schematic image of monolayer (ML), 118fand graphene quantum dots, 113-128 polymer nanocomposite-based photodetectors, 125-128

Photodetector applications of carbon quantum dots (Continued) based photodetectors and performances, 127t growth process of the hybrid device, 125fI-V characteristics, 126f sensitized nanomaterial-based photodetectors, 120-125 IV characteristics of GQD/ZnO NRs photodetector, 123f sensitized ZnO NR/GaN-NT heterostructure, 124f Si nanowire/CQD device, 123f time-dependent photocurrent, 122f ZnO nanorods, 121f Photophysical properties of carbon quantum dots classes of, 2f optical absorption of, 2-3 photoluminescence of, 4-25 Photovoltaic application of carbon quantum dots in all-weather solar cells, 154 in dye-sensitized solar cells, 136-144 in perovskite solar cells, 150-154 in solid-state solar cells, 148-150 Physical and chemical properties of carbon dots via computational modeling applications in different fields, 30f different carbon dots, 29-30 property of, 31 theoretical study of, 32-37 Physical property of carbon quantum dots biological, 72 bioimaging of HeLa cells, 72f catalytic reactions of, 66f composite and hybrid structures of, 66f optical, 66-71 absorption, 67-68 electroluminescence, 70-71 fluorescence, 69 from glucose/NaOH, 69f of MCBF, 67f phosphorescence, 69-70, 70f photoluminescence, 68-70 up-converted photoluminescence, 71, 71f photoinduced electron transfer, 72

physiochemical property, 64–65, 65*f* Properties of cDots fluorescence, 247–248 quantum yield, 248 Property of carbon dots computational methods applied to study, 31 system size *versus* computational cost, 31*f* Pyrolysis, 96

#### Q

Quantum dot density of states of electrons in, 403–404, 404*f* Schrodinger's equation for a quantum box, 403

#### R

Reviewing published work Al-based single-electron transistor, 434 - 442clean IQHE system, 435f color map of spatial density, 443f earlier discovered integer quantum Hall effect (IQHE), 434 electrical imaging of the quantum hall state, 434-440 inverse compressibility, 438f Landau level contours, 437f monolayer graphene, 441fscanning single-electron transistor, 440 - 442application of carbon nanotube-based single-electron transistor, 442-452 cantilever-mounted SET scanning, 444f CNT-based SET, 449 gate-dependent motion of domains, 446f graphene device with a bent channel, 451f optical micrograph of the device, 450f origin of anomalous piezoelectricity in LAO/STO, 443-447 piezoelectric response, 445f resistivity, alternating current, 448f voltage drop and current density in graphene/, 447-452 Roles of carbon quantum dots in photocatalysis

allocate additional surface for adsorption and reaction, 277 broaden the optical absorption range of, 274–275 charge separation and electron transfer, 276–277 photoexcitation of, 276*f* upconversion photoluminescence, 276*f* 

#### S

Scanning cavity microscopy, 457-458 Single-electron transistor as an electrometer Al/AlOx/Al structure, 431f backgate-2DES-SET structure, 427 2DES sample, 430f experimental realization of, 429-434 measurement using a SET scanning probe, 428f measuring inverse compressibility, 426-429 nanoassembled SET device, 432f SETs using Al QD and CNT QD, 433t Single-electron tunneling, 425 S-K mode, 406 Solar cell, 135–136 Surface engineering of carbon quantum dots in carbogenic allotropes, 93 carbon nanotube versus, 91-92 functionalities in, 92f methodology, 93-102 Synthesis methods of functionalized carbon quantum dots arc discharge, 160 liquids for production of, 161f electrochemical synthesis, 160, 160f hydrothermal and solvothermal synthesis, 163, 163f microwave-assisted synthesis, 161-162, 162f pulsed laser ablation/passivation technique, 161, 162f Synthesis of carbon quantum dots basic techniques for, 42-51 carbon quantum dots, 39-42 graphene oxide, 40f principles of synthesis, 41-42 structure of, 40-41, 41f Synthesis of carbon quantum dots and graphene quantum dots

bottom-up synthesis process, 107–108 AFM micrograph of NGQDs, 109*f* boron doped GQDs(BGQDs), 110*f* top-down synthesis process, 106–107 growth of GQDs from SWCNTs, 107*f* treatment procedures of coal samples, 108*f* Synthesis route of carbon quantum dots bottom-up method, 278 methods to synthesize, 279*t* top-down method, 277–278

#### Т

Techniques for carbon quantum dot preparation bottom-up approach, 48-51 hydrothermal method, 49-51, 50f microwave-assisted method, 49, 49f PEG-functionalized FCNs, 52f ultrasound-assisted method, 51 top-down approach, 42-48 arc discharge method, 42-43 chemical ablation/oxidation, 47-48 controllable synthesis of fluorescent nanomaterials, 45f electrochemical synthesis, 46-47 laser ablation, 43-45, 44f plasma treatment, 45-46 from sugarcane bagasse pulp, 48f Theoretical study of carbon quantum dots electrocatalytic, 35-36 properties of CQDs and GQDs, 35f electronic structure, 32-33 energy gap vs the size of the aromatic ring, 32f Kondo effect in, 36-37 optical property, 33-34 PL and fluorescence mechanisms, 34f transport, 36 long-distance interaction via Klein tunneling, 37f Theory energy for states, 426f origin of coulomb blockade oscillation, 423 - 425potential landscape across, 424f SET showing the equivalent electrical circuit, 424f to single-electron transistor, 423

The red edge effect, 11 Transistors basics of, 193–194 carbon quantum dots used in, 194–197 CQD channel, 195*f* current vs voltage curves, 197*f* fluorescence decay traces of, 198*f* humidity sensor fabrication and data acquisition setup, 199*f* phototransistor fabricated with the ZnS/ carbon QDs, 196*f* 

#### U

Unipolar device, 193

#### W

Watermelon rinds potential as carbon precursor, 283 Write-read-erase-read (WRER), 185