Jurnal Teknologi

EVALUATION OF PIPER BETLE L. EXTRACTS AND ITS ANTIVIRULENCE ACTIVITY TOWARDS P. AERUGINOSA

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Graphical abstract



Abstract

The virulence factor of bacteria such as P. aeruginosa causes severe problems affecting human health and environmental quality. In this study, Piper betle undergoes an extraction process yielding extract to diminish the virulence factor of P. aeruginosa. The efficiency of Piper betle treatment on P. aeruginosa was measured using Pyoverdine assay. The different factors affected the Piper betle extract yield such as leaves to a solvent ratio (1:6 and 1:10), extraction method (maceration and sonication) and different solvents (methanol, ethanol, ethyl acetate and hexane) were tested. Pyoverdine assay illustrates ethyl acetate exhibits the lowest peak (OD₆₃₀ = 0.2320) compared to methanol, ethanol and hexane due to the presence of a bioactive compound reducing the virulence factor. The ratio of 1:10 has a higher yield of 4.53±0.05 g and the ratio of 1:6 yields 2.86±0.05 g of extracts because of a better contact area. Maceration with agitation indicated the highest yield of 0.5210±0.05 g followed by maceration without agitation at 0.2660±0.05 g and 0.2792±0.05 g for sonication. The yield of Piper betle with different solvents showed the lowest yield is hexane 0.4741±0.05 g followed by ethyl acetate 2.4975±0.05 g, ethanol 3.7658±0.05 g and methanol 6.3331±0.05 g due to solvent polarity. This study aims to provide insightful knowledge of applied factor affecting Piper betle extracts and the ability of Piper betle as antivirulence and antibacterial agent against P. aeruginosa.

Keywords: Piper betle, P. aeruginosa, antivirulence, antibacterial, Pyoverdine assay

Abstrak

Faktor virulens untuk bakteria seperti P. aeruginosa menyebabkan masalah teruk yang boleh mempengaruhi kesihatan manusia dan kualiti persekitaran. Dalam kajian ini, Piper betle menjalani proses pengekstrakan bagi menghasilkan ekstrak untuk menurunkan faktor virulens P. aeruginosa. Kecekapan rawatan Piper betle terhadap P. aeruginosa diukur dengan menggunakan Pyoverdine assay. Faktor yang berbeza mempengaruhi jumlah ekstrak Piper betle seperti nisbah daun ke pelarut (1:6 dan 1:10), kaedah pengekstrakan (maserasi dan sonikasi) dan pelarut yang berbeza (metanol, etanol, etil asetat dan heksana) telah diuji. Pyoverdine assay menunjukkan etil asetat menunjukkan puncak terendah berbanding

85:1 (2023) 133–140 | https://journals.utm.my/jurnalteknologi | eISSN 2180–3722 | DOI: https://doi.org/10.11113/jurnalteknologi.v85.18892 |

Full Paper

Article history

Received 19 July 2022 Received in revised form 6 November 2022 Accepted 9 November 2022 Published Online 26 December 2022

*Corresponding author mazmir@ump.edu.my metanol, etanol dan heksana kerana adanya sebatian bioaktif yang mengurangkan faktor virulen. Nisbah 1:10 menghasilkan ekstrak yang lebih tinggi iaitu 4.53±0.05 g dan nisbah 1:6 menghasilkan 2.86±0.05 g ekstrak kerana mempunyai kawasan kontak yang lebih banyak. Macerasi dengan pergolakan menunjukkan hasil tertinggi 0.5210±0.05 g diikuti dengan maserasi tanpa pergolakan pada 0.2660±0.05 g dan 0.2792±0.05 g untuk sonikasi. Hasil Piper betle dengan pelarut yang berbeza menunjukkan hasil terendah ialah heksana 0.4741±0.05 g diikuti oleh etil asetat 2.4975±0.05 g, etanol 3.7658±0.05 g dan metanol 6.3331±0.05 g kerana kekutuban pelarut. Kajian ini bertujuan untuk memberi pengetahuan yang mendalam tentang faktor gunaan yang mempengaruhi ekstrak Piper betle dan keupayaan Piper betle sebagai agen antivirulen dan antibakteria terhadap P. aeruginosa.

Kata kunci: Piper betle, P. aeruginosa, antivirulen, antibacteria, Pyoverdine assay

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1.0 INTRODUCTION

Piper betle taxonomical hierarchy comes from Kingdom: Plantae; Division: Magnoliphyta; Class: Magnolipsida; Order: Piperales; Family: Piperaceae; Genus: Piper and Species: betle [1, 2]. Piper betle common name differ from one to another based on locality such as Sirih in Malaysia and Indonesia, Paan in India and Kun in Myanmar [3, 4]. Piper betle substantially cultivated in East Asian countries such as Malaysia, Indonesia, Philippines, Vietnam, Laos, Thailand, Myanmar, Singapore, Bangladesh, Pakistan, India and Sri Lanka [3, 5]. Piper betle is a medium to large-sized climbers' plants that can grow up to 20 meters if left unattended [6]. Piper betle plants are shrubby herbs with a pile of heart-shaped leaves, thin and flexible stem, flower and seed [2, 7]. Piper betle leaves are smooth on the surface and veiny on the back. The leaves' colour transforms from dark green to lighter green and yellowish-green when it gets older. Piper betle leaves give off a pungent and savoury odour when the leaves break apart or are crushed [2, 6, 8]. The plant basically delicate plant to grow off the tropical region [4]. Piper betle preferred warm and humid condition with neutral soil bed between pH 7 to 7.5 [4, 5]. Piper betle is used in traditional medicine to treat open wounds and bruises [5, 9, 10]. The previous study exhibits Piper betle as a medium in antibacterial, antioxidant, antifungal, antidiabetic, and anticancer agents [5, 11, 12]

Piper betle have been actively researched on capability as antibacterial medium. Previous research of Piper betle against a different types of bacterial strain shown a positive outcome on antimicrobial activity. Nayaka et al. (2021) tested Piper betle ethanol extracts against E. coli and P. aeruginosa by agar well diffusion method. The concentration of Piper betle extracts lays between 50 µg/mL to 100 µg/mL [13]. E. coli have bigger area of inhibition range from 8.9 mm to 11.0 mm compared to P. aeruginosa range from 6.7 mm to 7.2 mm [13]. Widyaningtias et al. (2014) used two (2) types of solvent to extract the Piper betle;

(1) n-hexane and followed by (2) ethanol at four (4) different concentration of 2.5 mg/mL, 5 mg/mL, 10 mg/mL, and 20 mg/mL [14]. The Piper betle extract tested against P. acnes using zone inhibition resulting in 7.01 mm, 8.92 mm, 13.28 mm and 21.08 mm respectively [14]. Hoque et al. (2012) had tested ethanol extracts of Piper betle for area of inhibition against E. coli and S. aureus. E. coli have a larger area of inhibition of (14.67±1.15) mm compared to S. aureus (14.67±0.57) mm [9]. Datta et al. (2011) performed zone of inhibition on antimicrobial activity of Piper betle ethanolic extracts on four (4) different bacterial strain such as Staphylococcus aureus, Pseudomonas aeruginosa, Klebsiella pneumonia and Proteus vulgaris [15]. At 0% of ethanolic dilution, the largest zone of inhibition is Pseudomonas aeruginosa (16±0.24) mm, followed by Klebsiella pneumonia (14±0.15) mm, Staphylococcus aureus (13±0.43) mm and Proteus vulgaris (10±0.5) mm [15]. Datta et al. (2011)also undergo Minimum Inhibitory Concentration (MIC) test on the bacterial strain are Staphylococcus aureus (40 µg), Pseudomonas aeruginosa (35 µg), Klebsiella pneumonia (25 µg), and Proteus vulgaris (25 µg) [15]. Kaveti et al. (2011) uses 50 µL Piper betle ethanol extracts to test four (4) bacterial strain of B. subitilis, S. aureus, E. coli and P. aeruginosa. The largest area of inhibition is *B. subitilis* (13.2+0.22) mm, followed by S.aureus (9.7+0.02) mm, E. coli (8.9+0.21) mm and P. aeruginosa at (7.2+0.42) mm [16].

Multiple extraction methods available to extract any plant components such maceration, soxhlet extraction, sonication, microwave-assisted extraction (MAE), supercritical fluid extraction (SFE) and ultrasound-assisted extraction (UAE) [5, 17, 18, 19]. The typical extraction method used in plant extraction is maceration and sonication due to simple experimental setup, less hassle, easier in material handling, can be operated in room temperature and is capable of extracting heat-sensitive volatile compounds [18, 20, 21]. Maceration technique used soaking method to extract plant components with/without additional support such as agitation or induce temperature at specific periods [7, 18]. In comparison, sonication uses electromagnetic waves to burst the air bubble produce by water ripples and causes the active pharmaceutical ingredients (API) or plant compounds to break apart from the plant and dissolve in the solvent [20, 22, 23, 24].

The solvent extracts the bioactive compounds or API from the plants. The solvent capability to extricate bio-active compounds depend on the polarity of the solvents, either polar or non-polar solvent. The polarity of solvent used in plant extraction affected the extract percentage of yield and the antioxidant activity [25, 26, 27]. The more polar the solvent, the more phenolic compound extracted out of plants [28]. The example of solvent based on higher polarity to lower polarity is methanol, ethanol, water, ethyl acetate and hexane. The number and types of components extracted from plants are undefined and variant depending on the polarity of solvent [29]. The bio-active compounds present in Piper betle are eugenol, flavonoids (quercetin), tannins, chavibetol and hydroxychavicol [5, 10, 30, 31, 32].

Bacterial infection faced by any living organism such as humans, animals and even plants causes difficulty performing the daily routine or can be deadly as well. Bacteria can infect both externally and internally depending on the route of infections. Skin infections causes by bacteria penetrated through open wounds that not treated carefully. The skin infection can be reduced and soothing the inflammation by applying antibacterial topical cream or gel on the infected skin with bacteria [33]. The treatment should relieve bacterial infection after a few applications depending on the level of severity. Pseudomonas aeruginosa (P. aeruginosa) commonly inhabits soil, water and vegetation. The ability of P. aeruginosa to adapt to the harsh condition and metabolic flexibility is proved by their evolution from mere bacteria into virulence bacteria such as PA14 and PA01 [34,35]. The virulence factor in bacteria in an environmental context focuses on P. aeruginosa in decreasing water quality by producing biofilm [36].

formation started with Biofilm bacterial communication. Quorum sensing (QS) is a form of communication between bacteria [37, 38]. The biofilm causes the virulence factor of bacteria to emerge and indirectly causes bacterial infection. The virulence factor of bacteria can be measured quantitatively by Pyoverdine assay (PA). Pyoverdine is a siderophore produced by P. aeruginosa that undergoes a stress environment and surroundings [39, 40, 411. Siderophore is a small molecule responsible for producing the virulence factor of P. aeruginosa scavenging for iron (III) in biofilm formation [39, 42, 43]. Iron (III) is an essential compound required by P. aeruginosa in bacterial growth and virulence factor [44].

Fe Dye^{3- λ} + L^{X-} -> Fe L^{3-X} + Dye λ -

In this study, the effectiveness of Piper betle extracts from different solvents uses a Pyoverdine assay by measuring the differences of the Optical density (OD) between t_0 and t_{10} . The other factor affecting the Piper betle extract yield is further discussed such as leaves to solvent ratio, extraction method and different solvent. The Piper betle extraction method focused on two (2) standard extraction techniques: maceration and sonication. Different ratios of Piper betle leaves to solvent determine by using two (2) different ratios (1:6 and 1:10). The total yield of Piper betle extract determines by using multiple solvents named methanol, ethanol, ethyl acetate and hexane. The results of this study allowed the researcher to gain knowledge on the substitution of raw material in antivirulence and antibacterial agents.

2.0 METHODOLOGY

2.1 Materials

Piper betle mature leaves (consistently purchased from a florist in Sungei Way, Petaling Jaya, Selangor), ethyl acetate, methanol, ethanol, hexane, P. aeruginosa, King's B media, Chrome Azurol S (CAS) reagent, Iron(III) chloride (FeCl₃.6H₂O) hexahydrate, Hydrochloric acid (HCI), hexadecyltrimethylammonium (HDTMA), Dimethyl sulfoxide (DMSO), distilled water.

2.2 Piper betle Extraction

Piper betle leaves was washed under running tap water thoroughly and rinse with distilled water [45]. The leaves dried in oven for 3 days at 50 °C. The leaves are then crushed using dried blender and sieved using mesh. 50 g of Piper betle leaves soaked in 500 ml of ethyl acetate with a ratio of 1:10 for 12 hours [18]. The extraction of Piper betle undergoes maceration with agitation using a magnetic stirrer for 8 hours at a speed of 800 rpm. The extraction was kept inside the fume hood with a cover to avoid the release of solvent fume into the surroundings. The mixture was filtered using a filter paper, and the filtrate is then evaporated using a rotary evaporator. The temperature was set at 40 °C until the evaporation process ended and extracts were collected. The extract scrapes out and was kept in the amber sample bottle. The whole procedure was repeated for all the other remaining solvents such as methanol, ethanol and hexane in order to obtain respective extracts.

2.3 Subculture of Bacteria

The bacteria undergo an agar subculture to recover bacteria [34]. A single colony from agar was subcultured then transferred into King's B media [46]. The bacterial cultured for 16 hours at 37 °C [34]. The bacterial culture was then measured using a spectrophotometer at $OD_{600} = 0.05$.

2.4 Preparation of CAS Solution

The Chrome Azurol S (CAS) solution was prepared [47,48]. 0.06 g of CAS reagent mixed with 50 mL of distilled water (Solution A). 0.0027 g of FeCl₃.6H₂O was dissolved in 10 mL HCl (Solution B). 0.073 g of HDTMA dissolved in 40 mL of distilled water (Solution C). Solution A added with 9 mL of solution B, and the mixture was swirled gently to prevent air bubble formation. After that, solution C was added to the mixture. The mixture was transferred into Scott bottle and autoclave for 20 min. The Scott bottle was wrapped with aluminium foil to minimise the penetration of light into the mixture solution. The solution is solution.

2.5 Culture Treatment

The bacterial subculture undergo treatment with *Piper* betle extracts from different solvents such as methanol, ethanol, ethyl acetate and hexane. The extract was prepared by diluting the solidified extracts with DMSO until the final concentration of extract was 200 mg/mL. A mixture of 20 μ L of bacterial culture and 20 μ L of *Piper betle* extract was added introduced into new 20 mL King's B broth [46]. The cultured treatment was then incubated at 37 °C for 16 hours [34]. The colour change can be monitored at the end of a cycle.

2.6 Pyoverdine Assay

The cultured treatment undergoes separation using refrigerated centrifuges (Eppendorf Centrifuge 5810R) at 4 °C, 3094 g for 30 minutes. Then, the supernatant was collected and filtered with a 0.5 mm syringe filter. The filtrate was analysed via Pyoverdine assay. 0.5 mL of filtrate was placed in 1 mL cuvette, and 0.5 mL CAS reagent was added to the cuvette. The reading was measured using a spectrophotometer at OD₆₃₀ (Optical density used to measure bacterial biofilm production) [49]. The absorbance reading is taken at to and t10. The untreated P. aeruginosa was served as a control to differentiate the quorum sensing activity between P. aeruginosa with other extracts from different solvents such as methanol, ethanol, ethyl acetate and hexane. The experiment was carried in triplicate. The average data was recorded and tabulated.

3.0 RESULTS AND DISCUSSION

3.1 P. aeruginosa Agar Subculture

The agar subculture of virulence *P. aeruginosa* emits bluish-greenish colour (Figure 1) on the agar plate on areas inhabited by bacterial colonies [34,50]. Pyocyanin secreted by *P. aeruginosa* is a pigment responsible for producing bluish-greenish colour to signify the exhibition of a virulence factor in *P*. aeruginosa [34,50,51].



Figure 1 Agar subculture of *P. aeruginosa* in King's B agar shown the blue hues from the pyocyanin compound

3.2 P. aeruginosa Culture Treatment against Piper betle

The bacterial culture undergoes culture treatment with *Piper betle* extracts from multiple solvents (Figure 2). The culture treatment can be monitored qualitatively by observing colour variation between the type of solvent used at the end of treatment. The solvent used in *Piper betle* extraction process is methanol, ethanol, ethyl acetate and hexane. The solvent was selected based on the polarity of the solvent [29].

Figure 2 shows colour differences between the mixtures shifted from green to greenish-yellow and yellowish-chalky. Figure 2(a) contained only P. aeruginosa without treatment. The mixture's emphasis on areen colour signifies pyocyanin's presence that emphasises the presence of P. aeruginosa virulence factor [34,50,51]. P. aeruginosa without treatment also served as a control throughout the experiment to differentiate the reaction between treated P. aeruginosa and non-treatment P. aeruginosa. Figure 2(b) mixtures colour appears greenish-yellow is a treatment between P. aeruginosa with P. betle extracts using hexane as solvent. The colour implies that hexane incapable or has insufficient ability to extract out the required component or bio-active compound responsible for blocking the formation of a virulence factor of P. aeruainosa due to hexane beina a non-polar solvent. Figure 2(c), Figure 2(d) and Figure 2(e) are a mixture of P. aeruginosa treated with Piper betle extracts from methanol, ethanol and ethyl acetate respectively. All three (3) beaker shows a vellowish-chalky mixture indicated that the reaction P. aeruginosa broth subculture grows without the presence of virulence factor represented by a green colour (pyocyanin) [34,50,51].

Due to the lack of colour differences between the three of the mixture, further clarification or bio-assay is required to determine the strength and capabilities of methanol, ethanol and ethyl acetate extract on the virulence factor of *P. aeruginosa*. Therefore, in this study, further experiment was conducted by performing Pyoverdine assay.



Figure 2 P. aeruginosa subculture in P. betle extracts (a) P. aeruginosa without treatment (control), (b) P. aeruginosa in P. betle extracted from hexane, (c) P. aeruginosa in P. betle extracted from ethanol, (d) P. aeruginosa in P. betle extracted from methanol, (e) P. aeruginosa in P. betle extracted from ethyl acetate

3.3 Pyoverdine Assay

Pyoverdine assay is a method to measure the capabilities of the extract to treat *P. aeruginosa* virulence factor [35]. This method is a quantitative method as the mixture's optical density (OD) is measured at a time interval of 0 to 10 minutes using a spectrophotometer [49]. Figure 3 shows the graph of *P. aeruginosa* used to determine the quorum sensing (QS) activity after adding CAS reagent. The treatment culture incorporates CAS reagent measured at t_0 and t_{10} . The differences in absorbance reading between the OD are calculated and normalised (Table 1).

Virulence bacteria produce potent ligands during the formation of virulence factors. The high affinity towards iron III during quorum sensing chemical reaction leads to detection of siderophore (Pyoverdine) using CAS reagent as an indicator [47,48]. A potent ligand such as pyoverdine removed iron III from CAS reagent causes blue colour shifting between t_0 and t_{10} [49]. Thus, the more biofilm present, the more colour changes monitored and vice versa due to strong ligand (siderophore) to grab the iron III compound and remove the blue dyes from the mixtures. The absorbance was measured by differences in Optical density (OD) between t_0 and t_{10} . The larger the differences in OD between to and to measured, the more interaction between the bacteria in the bacterial colonies and the more intense the virulence factor or biofilm production.

Table	1	Pyoverdine	Assay
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Solvent	Average±SD	
PA01	1	
Methanol	0.3863±0.05	
Ethanol	0.4381±0.05	
Ethyl Acetate	0.2320±0.05	
Hexane	0.5567±0.05	



Figure 3 Graph of QS activity of *P. aeruginosa* against *P. betle* extracted from different solvent (a) *P. aeruginosa* without treatment (control), (b) *P. aeruginosa* in *P. betle* extracted from methanol, (c) *P. aeruginosa* in *P. betle* extracted from ethanol, (d) *P. aeruginosa* in *P. betle* extracted from ethyl acetate, (e) *P. aeruginosa* in *P. betle* extracted from hexane

P. aeruginosa has shown the highest peak ($OD_{630} =$ 1) due to throughout QS activity because of nil treatment. Thus, the interaction between the bacteria to produce virulence factor was accomplished at the uttermost level. The second highest peak (OD₆₃₀ = 0.5567) in the graph is hexane extract. Hexane is a nonpolar solvent. Thus, many bioactive compounds do not mix or dissolve in hexane, resulting in lower phenolic bioactive compounds extracted [28]. The third highest peak ($OD_{630} = 0.4381$) is ethanol, whereas the fourth highest peak ($OD_{630} = 0.3863$) is methanol. Methanol is a more polar solvent than ethanol, resulting in a higher bioactive compound in methanol to lower the virulence factor. Finally, ethyl acetate has the lowest peak ($OD_{630} = 0.2320$) among the solvents. Ethyl acetate is a polar solvent. Even though the polarity of ethyl acetate was lower than both methanol and ethanol, ethyl acetate contains the active compound that minimises the virulence factor of P. aeruginosa [18,45]. Thus, the QS activity between bacteria lowered down and proved by the Pyoverdine assay.

3.4 Factor Affecting Yield of Piper betle: Different Ratio of Piper betle Leaves to Ethyl Acetate Solvent

In this experiment, *Piper betle* leaf to solvent was tested with two (2) different ratios of 1:6 and 1:10 [18,45,52]. The 1:6 ratio shows that the extracts' volume yields 2.86 mL while the 1:10 ratio manifests 4.53 mL of *Piper betle* extract (Table 2). Thus, the volume yields of *Piper betle* in ratio 1:10 almost double the yield extracted from 1:6 ratios. The more significant difference in *Piper betle* leaf ratio to solvent gives off a higher yield due to the higher surface volume of solvent contact with *Piper betle* leaves [45]. Thus, it increases the rate of extraction and produces a higher yield.

 Table 2 Extraction of Piper betle using different ratios of leaf to solvent

Ratio of Piper betle leaf to volume of solvent	Piper betle leaf (g)	Volume of Ethyl Acetate (ml)	Average Volume of extract (g)
1:6	50	300	2.86±0.05
1:10	50	500	4.53±0.05

3.5 Factor Affecting Yield of Piper betle: Extraction Method

The experiment was carried out using two methods; sonication and maceration. Aside from stand still maceration techniques, the maceration with agitation is introduced to the system to test whether the agitation will improve the percentage yield of an extract. The experiment was used ethyl acetate as a solvent. Table 3 showed extraction yield using a different type of extraction method. The maceration without agitation yield 0.2792 g of yield, whereas maceration with agitation yield 0.5210 g of yield. Maceration took up to 12 hours for *Piper betle* extraction. Maceration techniques are the most appointed method for extraction due to simple procedure, less hassle, and cheaper [5,18].

Sonication yield 0.2660 g of extract. Sonication involves the electromagnetic waves that produce bubbles, vibrate at incredible speed, and burst the water bubble causes the bioactive compound to release to the surroundings and absorbed by solvent [22]. The sonication technique can preserve volatile bioactive compounds that disintegrate when exposed to heat [20,21]. Thus, this method will allow the volatile bioactive compound to be collected rather than disintegrate or break due to heat. However, due to the low yield of extracts, maceration with agitation is the preferable method to extract Piper betle. As the maceration with agitation without heat induced was applied during this study, the disintegration of volatile bioactive compound will be avoided.

 Table 3 Extraction yield using a different type of extraction method

Extraction Method	Piper betle leaf (g)	Volume of Ethyl Acetate (ml)	Ratio of Piper betle leaf to volume of solvent	Average Yield of extract (g)
Sonication Maceration (Without Agitation)	10 10	100 100	1:10 1:10	0.2660±0.05 0.2792±0.05
Maceration (With Agitation)	10	100	1:10	0.5210±0.05

3.6 Factor Affecting Yield of Piper betle: Types of Solvent

Piper betle leaf extracted with different solvents named methanol, ethanol, ethyl acetate and hexane

based on 1:10 ratio of Piper betle leaves to solvent. The test was conducted to determine the Piper betle weight of yield extracted based on the type of solvent. Table 4 illustrates that methanol has the highest yield of 6.3331 g, followed by ethanol 3.7658 g, ethyl acetate 2.4975 g and hexane 0.4741 g. The methanol extracts exhibit the highest yield of extracts because methanol is the most polar compared to other solvents such as ethanol, ethyl acetate and hexane. The more polar the solvent used in the extraction method using Piper betle leaves, the better the extraction performance and the higher the average weight of yield of Piper betle extracts [25,26,27,28]. The yield does not represent the capabilities of the bioactive compound toward virulence factor but focuses solely on the yield of extracts. Based on the observation, the average weight of yield increased by the polarity of solvent started from methanol, ethanol, ethyl acetate and hexane [28].

Table 4 Total volume of extract using different solvent

Solvent	Piper betle leaf (g)	Volume of Solvent (ml)	Ratio of Piper betle leaf to volume of solvent	Average weight of extract (g)
Ethyl Acetate	50	500	1:10	2.4975±0.05
Ethanol	50	500	1:10	3.7658±0.05
Methanol	50	500	1:10	6.3331±0.05
Hexane	50	500	1:10	0.4741±0.05

4.0 CONCLUSION

The agar subculture of P. aeruginosa shown the blue hues from the pyocyanin compound on the agar plate proved the presence of virulence factor. The culture treatment of P. aeruginosa treated with Piper betle extracts from methanol, ethanol, ethyl acetate, and hexane showed that methanol, ethanol, and ethyl acetate positively react against virulence factors. The graph on Pyoverdine assay tested P. aeruginosa culture with methanol, ethanol, ethyl acetate and hexane Piper betle extracts shown highest peak is P. aeruginosa without the treatment followed by hexane, ethanol, methanol and ethyl acetate. Ethyl acetate has the lowest peak due to its capabilities to extract bioactive compounds to minimise the virulence factor of P. aeruginosa compared to other solvents.

The yield of *Piper betle* extract is affected by many factors such as the ratio of *Piper betle* leaves to solvent, type of extraction method and different solvents. The extraction method with the ratio of 1:10 producing a higher yield compared to 1:6 ratios due to the larger surface area of *Piper betle* leaves contact with the solvent. The extraction techniques of maceration and sonication show that maceration with agitation has a double yield compared to the

other method. Weight yield of *P. betle* extract based on four (4) different solvents; hexane, ethanol, methanol and ethyl acetate exhibits methanol has the highest yield compared to other solvents.

Even though the ability of ethyl acetate Piper betle extract to minimise and decrease the biofilm formation and virulence factor of *P. aeruginosa* proved by Pyoverdine assay, the absolute compound that is still unknown. Thus, further investigation and research regarding the bioactive compound are required. The endless possibility of *Piper betle* can be discovered aside from its antibacterial effect such as antifungal, antioxidant, antidiabetic, and anticancer properties that will be beneficial for human society as *Piper betle* easy to grow, produce lushes leaves and requires minimum care to grow healthily.

Acknowledgement

This work was supported by the Ministry of Education Malaysia (MOE) under the Fundamental Research Grant Scheme (grant no. FRGS/1/2019/STG05/UMP/02/9) and Universiti Malaysia Pahang (UMP) under the UMP Research Grant Scheme (grant no. RDU190338).

References

- Wendy Voon, W. Y., Ghali, N. A., Rukayadi, Y., Meor Hussin, A. S. 2014. Application of Betel Leaves (*Piper betle L.*) Extract for Preservation of Homemade Chili Bo. International Food Research Journal. 21 (6): 2399-2403.
- [2] Ravindran, P. N., Pillai, G. S. and Nirmal Babu, K. 2004. Under-utilized Herbs and Spices. Handbook of Herbs and Spices. 53-103.

Doi: https://doi.org/10.1533/9781855738355.1.53.

- [4] Shah, S. S., Garg, G., Jhade, D. & Patel, N. 2016. Piper betle: Phytochemical, Pharmacological and Nutritional Value in Health Management. Int. J. Pharm. Sci. Rev. Res. 38(2): 181-9.
- [5] Azahar, N. I., Mokhtar, N. M., & Arifin, M. A. 2020. Piper betle: A Review on Its Bioactive Compounds, Pharmacological Properties, and Extraction Process. IOP Conference Series: Materials Science and Engineering. 991: 012044. Doi: https://doi.org/10.1088/1757-899X/991/1/012044.
- [6] Chan, E. and Wong, S. 2014. Phytochemistry and Pharmacology of Three Piper Species: An Update. International Journal of Pharmacognosy. 1: 534-54. Doi: https://doi.org/10.13040/IJP.
- [7] Pin, K. Y., Chuah, T. G., Abdull Rashih, A., Rasadah, M. A., Law, C. L. and Choong, T. S. Y. 2006. Aqueous Extraction of Hydroxychavicol from Piper betle L. Leaves. Proceedings of the 1st International Conference on Natural Resources Engineering & Technology 2006. 146-152.
- [8] Sarma. C., Rasane, P., Kaur, S., Singh, J., Singh, J., Gat, Y., ... Dhawan, K. 2018. Antioxidant and Antimicrobial Potential of Selected Varieties of Piper betle L. (betel leaf). Anais Da Academia Brasileira de Ciências. Doi: https://doi.org/10.1590/0001-3765201820180285.
- [9] Hoque, M. M., Rattila, S., Shishir, M. A., Bari, M. L., Inatsu, Y. and Kawamoto, S. 2012. Antibacterial Activity of Ethanol Extract of Betel Leaf (*Piper betle L.*) against some Food Borne Pathogens. Bangladesh Journal of Microbiology. 28(2). Doi: https://doi.org/10.3329/bjm.v28i2.11817.

- [10] Mohanto, S., Datta, S., and Mandal, S. 2017. Piper betel Linn: A Brief Study. International Journal of Current Medical and Pharmaceutical. 3(2): 1290-96.
- [11] Thamaraikani, I. and Kulandhaivel, M. 2017. Purification of Hydroxychavicol from Piper betle linn and Evaluation of Antimicrobial Activity Against Some Food Poison Causing Bacteria. Journal of Pure and Applied Microbiology. Doi: https://doi.org/10.22207/JPAM.11.4.28.
- [12] M. Guha, P. and Nag, A. 2019. Extraction of Betel Leaves (Piper betle I.) Essential Oil and Its Bio-actives Identification: Process Optimization, GC-MS Analysis and Anti-microbial Activity. Industrial Crops and Products. 138: 111578. Doi: https://doi.org/10.1016/j.indcrop.2019.111578.
- [13] Nayaka, N., Sasadara, M., Sanjaya, D. A., Yuda, P., Dewi, N., Cahyaningsih, E., & Hartati, R. 2021. *Piper betle* (L): Recent Review of Antibacterial and Antifungal Properties, Safety Profiles, and Commercial Applications. *Molecules* (Basel, Switzerland). 26(8): 2321. Design Marguet (Japane 2002).

Doi: https://doi.org/10.3390/molecules26082321.

- [14] Widyaningtias, N. M. S. R., Yustiantara, P. S., Paramita, N. L. P. V. 2014. Uji Aktivitas Antibakteri Ekstrak Terpurifikasi Daun Sirih Hijau (Piper betle L.) Terhadap Bakteri Propionibacterium acnes, J. Farm. 50-53.
- [15] Datta, A., Ghoshdastidar, S., & Singh, M. 2011. Antimicrobial Property of Piper betel Leaf against Clinical Isolates of Bacteria. International Journal of Pharma Sciences and Research. 2(3): 104-109.
- [16] Kaveti, B., Sarnnia, L. T., Tan, S. K., Baig, M. 2011. Antibacterial Activity of Piper Betel Leaves. IJPTP. 2(3):129-132.
- [17] Zhang, Q. W., Lin, L. G. and Ye, W. C. 2018. Techniques for Extraction and Isolation of Natural Products: A Comprehensive Review. *Chinese Medicine*. 13(1). Doi: https://doi.org/10.1186/s13020-018-0177-x.
- [18] Azwanida, N. N. 2015. A Review on the Extraction Methods Use in Medicinal Plants, Principle, Strength and Limitation. Medicinal & Aromatic Plants. 04(03). Doi: https://doi.org/10.4172/2167-0412.1000196.
- [19] Ingle, K. P., Deshmukh, A. G., Padole, D. A., Dudhare, M. S., Moharil, M. P. and Khelurkar, V. C. 2017. Phytochemicals: Extraction Methods, Identification and Detection of Bioactive Compounds from Plant Extracts. Journal of Pharmacognosy and Phytochemistry. 6(1): 32-6.
- [20] Ali, A., Lim, X. Y., Chong, C. H., Mah, S. H. and Chua, B. L. 2018b. Ultrasound-assisted Extraction of Natural Antioxidants from Betel Leaves (*Piper betle*): Extraction Kinetics and Modeling. Separation Science and Technology. 53(14): 2192-2205. Doi: https://doi.org/10.1080/01496395.2018.1443137.
- [21] Das, S., Ray, A., Nasim, N., Nayak, S., and Mohanty, S. 2019. Effect of Different Extraction Techniques on Total Phenolic and Flavonoid Contents, and Antioxidant Activity of Betel Vine and Quantification of its Phenolic Constituents by validated HPTLC Method. *Biotech.* 9(1). Doi: https://doi.org/10.1007/s13205-018-1565-8.
- [22] Foo, L. W., Salleh, E. and Hana, S. N. 2017. Green Extraction of Antimicrobial Bioactive Compound from *Piper betle* Leaves: Probe Type Ultrasound-assisted Extraction vs Supercritical Carbon Dioxide Extraction. *Chemical Engineering Transactions*. 56: 109-14. Doi: https://doi.org/10.3303/CET1756019.
- [23] Ali, A., Lim, X. Y., Chong, C. H., Mah, S. H. and Chua, B. L. 2018a. Optimization of Ultrasound-assisted Extraction of Natural Antioxidants from *Piper betle* using Response Surface Methodology. *LWT*. 89: 681-8. DOI: https://doi.org/10.1016/j.lwt.2017.11.033.
- [24] Dhanani, T., Shah, S., Gajbhiye, N. A., & Kumar, S. 2017. Effect of Extraction Methods on Yield, Phytochemical Constituents and Antioxidant Activity of Withania somnifera. Arabian Journal of Chemistry. 10: S1193-S1199. Doi: https://doi.org/10.1016/j.arabjc.2013.02.015.
- [25] Ghasemzadeh, A., Jaafar, H., Juraimi, A., & Tayebi-Meigooni, A. 2015. Comparative Evaluation of Different Extraction Techniques and Solvents for the Assay of

Phytochemicals and Antioxidant Activity of Hashemi Rice Bran. *Molecules*. 20(6): 10822-10838. Doi: https://doi.org/10.3390/molecules200610822.

[26] Ghasemzadeh, A., Jaafar, H. Z., Rahmat, A. 2011. Effects of Solvent Type on Phenolics and Flavonoids Content and Antioxidant Activities in Two Varieties of Young Ginger (Zingiber officinale Roscoe) Extracts. J Med Plants Res. 5(7): 1147-1154.

Doi: https://doi.org/10.5897/JMPR.9000540.

- [27] Barchan, A., Bakkali, M., Arakrak, A., Pagán, R., Laglaoui, A. 2014. The Effects of Solvents Polaritiy on the Phenolic Contents and Antioxidant Activity of Three Mentha Species Extracts. Int J Curr Microbiol App Sci. 3(11): 399-412.
- [28] Nawaz, H., Shad, M. A., Rehman, N., Andaleeb, H., & Ullah, N. 2020. Effect of Solvent Polarity on Extraction Yield and Antioxidant Properties of Phytochemicals from Bean (Phaseolus vulgaris) Seeds. Brazilian Journal of Pharmaceutical Sciences. 56. Doi: https://doi.org/10.1590/s2175-97902019000417129.
- [29] Taukoorah, U., Lall, N. and Mahomoodally, F. 2016. Piper betle L. (betel quid) Shows Bacteriostatic, Additive, and Synergistic Antimicrobial Action when Combined with Conventional Antibiotics. South African Journal of Botany. 105: 133-140.
- [30] Dwivedi, V. and Tripathi, S. M. 2014. Review Study on Potential Activity of Piper betle. Journal of Pharmacognosy and Phytochemistry. 3(4): 93-8.
- [31] Patil, R. S., Harale, P. M., Shivangekar, K. V., Kumbhar, P. P. and Desai, R. R. 2015. Phytochemical Potential and In Vitro Antimicrobial Activity of Piper betle Linn. Leaf Extracts. Journal of Chemical and Pharmaceutical Research. 7(5): 1095-1101.
- [32] Pradhan, D., Suri, K. A., Pradhan, D. K. and Biswasroy, P. 2013. Golden Heart of the Nature: Piper betle L. J Pharmacog Phytochem. 1(6): 147-67.
- [33] Budiman, A., Rusnawan, D. W., Yuliana, A. 2018. Antibacterial Activity of Piper betle L. Extract in Cream Dosage forms against Staphylococcus aureus and Propionibacterium acne. J. Pharm. Sci. & Res. 10(3): 493-496.
- [34] LaBauve, A. E., and Wargo, M. J. 2012. Growth and Laboratory Maintenance of Pseudomonas aeruginosa. Current Protocols in Microbiology. 25(1): 1-8. Doi: https://doi.org/10.1002/9780471729259.mc06e01s25.
- [35] Sykes, J. E. 2014. Gram-negative Bacterial Infections. Canine and Feline Infectious Diseases. 355-363. Doi: https://doi.org/10.1016/B978-1-4377-0795-3.00036-3.
- [36] Vukić Lušić, D., Maestro, N., Cenov, A., Lušić, D., Smolčić, K., Tolić, S., Maestro, D., Kapetanović, D., Marinac-Pupavac, S., Tomić Linšak, D., Linšak, Ž., & Glad, M. 2021. Occurrence of *P. aeruginosa* in Water Intended for Human Consumption and in Swimming Pool Water. *Environments*. 8(12): 132.

DOI: https://doi.org/10.3390/environments8120132.

[37] Samanta, I. and Bandyopadhyay, S. 2020. Alternative Antiinfective Therapy. Antimicrobial Resistance in Agriculture. 343-355.

Doi: https://doi.org/10.1016/B978-0-12-815770-1.00030-4.

- [38] Frederix, M., & Downie, J. A. 2011. Quorum Sensing. Advances in Microbial Physiology. 23-80. Doi: https://doi.org/10.1016/B978-0-12-381043-4.00002-7.
- [39] Kang, D., Kirienko, D. R., Webster, P., Fisher, A. L., & Kirienko, N. V. 2018. Pyoverdine, a Siderophore from *Pseudomonas* aeruginosa, Translocates into C. elegans, removes iron,

and Activates a Distinct Host Response. Virulence. 9(1): 804-817.

Doi: https://doi.org/10.1080/21505594.2018.1449508.
[40] Lyczak, J. B., Cannon, C. L., & Pier, G. B. 2000. Establishment of *Pseudomonas aeruginosa* Infection: Lessons from a Versatile Opportunist. *Microbes and Infection*. 2(9): 1051-1060.

Doi: https://doi.org/10.1016/\$1286-4579(00)01259-4.

[41] Lamont, I. L., Beare, P. A., Ochsner, U., Vasil, A. I., & Vasil, M. L. 2002. Siderophore-mediated Signaling Regulates Virulence Factor Production in *Pseudomonas aeruginosa*. *Proceedings of the National Academy of Sciences*. 99(10): 7072-7077.

Doi: https://doi.org/10.1073/pnas.092016999.

- [42] Meyer, J. M., Neely, A., Stintzi, A., Georges, C., & Holder, I. A. 1996. Pyoverdin is Essential for Virulence of Pseudomonas aeruginosa. Infection and Immunity. 64(2): 518-523. Doi: https://doi.org/10.1128/iai.64.2.518-523.1996z.
- [43] Hider, R. C. 1984. Siderophore Mediated Absorption of Iron. Siderophores from Microorganisms and Plants. 58: 25-87. Springer Berlin Heidelberg. Doi: https://doi.org/10.1007/BFb0111310.
- [44] Bonneau, A., Roche, B., & Schalk, I. J. 2020. Iron Acquisition in *Pseudomonas aeruginosa* by the Siderophore Pyoverdine: An Intricate Interacting Network Including Periplasmic and Membrane Proteins. *Scientific Reports*. 10(1).

Doi: https://doi.org/10.1038/s41598-019-56913-x.

- [45] Saha, J., and Deka, S. C. 2016. Functional Properties of Sonicated and Non-sonicated Extracted Leaf Protein Concentrate from Diplazium esculentum. International Journal of Food Properties. 20(5): 1051-1061. Doi: https://doi.org/10.1080/10942912.2016.1199034.
- [46] Lamichhane, J. R. and Varvaro, L. 2012. A New Medium for the Detection of Fluorescent Pigment Production by Pseudomonads. *Plant Pathology*. 62(3): 624-632. Doi: https://doi.org/10.1111/j.1365-3059.2012.02670.x.
- [47] Louden, B. C., Lynne, A. M., & Haarmann, D. 2011. Use of Blue Agar CAS Assay for Siderophore Detection. *Journal of Microbiology & Biology Education*. 12(1): 51-53. Doi: https://doi.org/10.1128/jmbe.v12i1.249.
- [48] Srimathi, K. & Ann Suji, H. 2018. Siderophores Detection by using Blue Agar CAS Assay Methods. International Journal of Scientific Research in Biological Sciences. 5(6): 180-5.
- [49] Virpiranta, H., Banasik, M., Taskila, S., Leiviskä, T., Halttu, M., Sotaniemi, V.-H., & Tanskanen, J. 2020. Isolation of Efficient Metal-binding Bacteria from Boreal Peat Soils and Development of Microbial Biosorbents for Improved Nickel Scavenging. Water. 12(7): 2000. Doi: https://doi.org/10.3390/w12072000.
- [50] Nawas, T. 2018. Extraction and Purification of Pyocyanin: A Simpler and More Reliable Method. MOJ Toxicology. 4(6). Doi: https://doi.org/10.15406/mojt.2018.04.00139.
- [51] Krieg, N. R. and Padgett, P. J. 2011. Phenotypic and Physiological Characterization Methods. Taxonomy of Prokaryotes. 15-60. Doi: https://doi.org/10.1016/B978-0-12-387730-7.00003-6.
- [52] Sayyar, S., Abidin, Z. Z., Yunus, R., & Muhammad, A. 2009. Extraction of Oil from Jatropha Seeds-Optimization and Kinetics. American Journal of Applied Sciences. 6(7): 1390-1395.

Doi: https://doi.org/10.3844/ajassp.2009.1390.1395.