

Modification of Capacitive Charge Storage of TiO₂ with Nickel Doping

Syam G. Krishnan^a, P.S. Archana^b, Baiju Vidyadharan^a, Izan Izwan Misnon^a, Bincy Lathakumary Vijayan^a, V. Manikantan Nair^c, Arunava Gupta^b, Rajan Jose^a

^a Nanostructured Renewable Energy Materials Laboratory, Faculty of Industrial Sciences & Technology, Universiti Malaysia Pahang, 26300, Kuantan, Malaysia

^b Department of Chemistry, University of Alabama, 250 Hackberry Lane, Shelby Hall, Tuscaloosa, 35401, AL, USA

^c Dept. of Physics, H. H. Maharaja's College for Women, University of Kerala, Trivandrum, India

ABSTRACT

For practical deployment of supercapacitors characterized by high energy density, power density and long cycle life, they must be realized using low cost and environmentally benign materials. Titanium dioxide (TiO₂) is largely abundant in the earth's crust; however, they show inferior supercapacitive electrochemical properties in most electrolytes for practical deployment. In this paper, we show that nickel doped TiO₂(Ni:TiO₂) nanowires developed by electrospinning showed five times larger capacitance ($\sim 200 \text{ F g}^{-1}$) than the undoped analogue ($\sim 40 \text{ F g}^{-1}$). Electrochemical measurements show that the Ni:TiO₂ nanowires have 100% coulombic efficiency. The electrodes showed no appreciable capacitance degradation for over 5000 cycles. The superior charge storage capability of the Ni:TiO₂ could be due to its high electrical conductivity that resulted in five orders of magnitude higher ion diffusion as determined by cyclic voltammetry and electrochemical impedance spectroscopy measurements.

KEYWORDS: Ceramic nanostructures; Renewable energy; Batteries; TiO₂ nanostructures; One-dimensional nanostructures

DOI: [10.1016/j.jallcom.2016.05.183](https://doi.org/10.1016/j.jallcom.2016.05.183)